

Synthesis and Biological Implication of Metal complexes of Chalcone-Coumarin Hybrid

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Abstract: *The present study focuses on the synthesis, characterization, and biological evaluation of Ni(II) complexes derived from a chalcone-coumarin hybrid ligand. The hybrid ligand was designed by integrating the pharmacologically significant chalcone and coumarin scaffolds, which are well known for their diverse biological activities [1-2]. In this synthesis Resorcinol and Beta keto ester refluxed overnight in acidic medium (with Conc. Sulphuric acid H₂SO₄) to form 4-methyl-7-hydroxy coumarin. This 4-methyl-7-hydroxy coumarin further reacts with acetyl chloride (CH₃COCl) in the presence of pyridine (a basic medium) to form 4-methyl-7-acetoxy coumarin. This 4-methyl-7-acetoxy coumarin further undergoes re-arrangement in the presence of Anhyd. AlCl₃ (Fries Rearrangement) to form 8-acetyl-7-hydroxy-4-methyl-2H-chromen-2-one. The structures of the newly synthesized compounds were confirmed by IR, ¹H NMR, ¹³C NMR spectral data and elemental analysis. All these compounds were screened for their antibacterial and antifungal activities. The results were compared with standard drugs tested under similar conditions. Some of these compounds showed promising antimicrobial activities.*

Keywords: Chalcone, Coumarin, Fries Rearrangement, IR, ¹H NMR, ¹³C NMR, Antibacterial activity, Antifungal activity

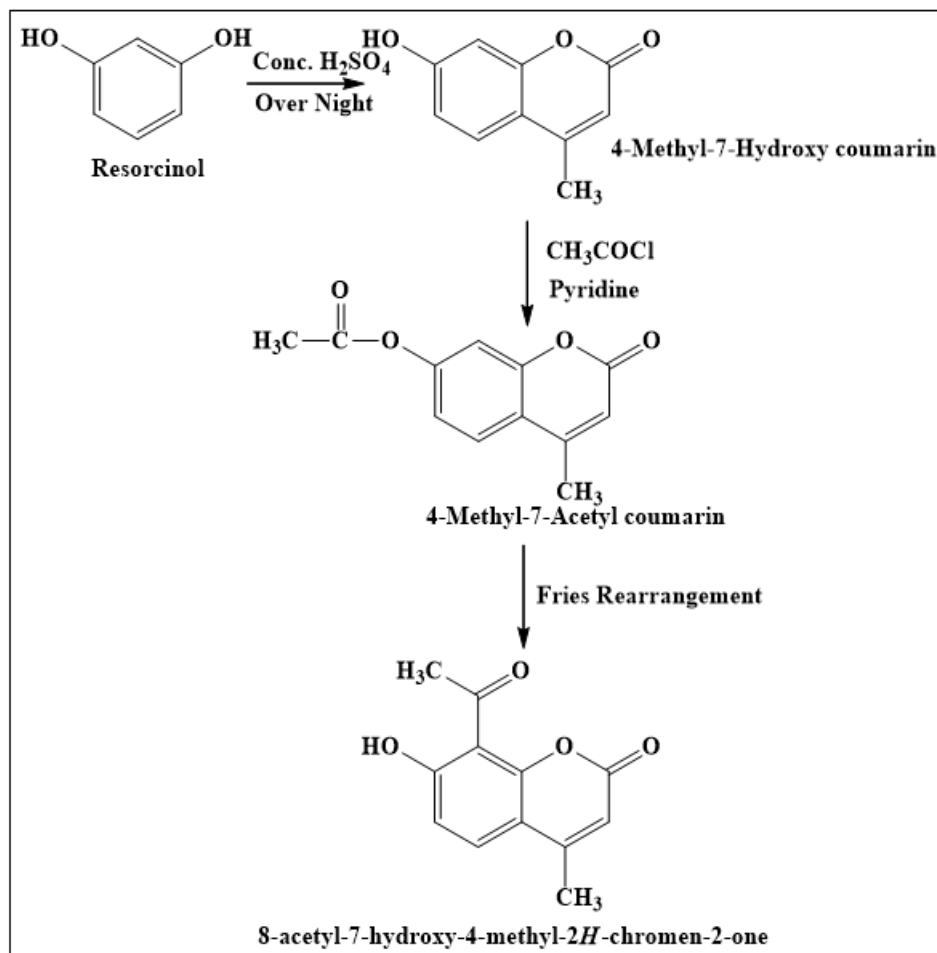
1. Introduction

Chalcone and coumarin are two important classes of oxygen-containing heterocyclic and alpha-beta unsaturated compounds that have attracted considerable attention because of their diverse pharmacological properties [3-6]. Chalcones characterized by the presence of an open-chain flavonoid like enone system, exhibit a broad spectrum of biological activities, including antibacterial [7], antifungal [8], anti-inflammatory [9], antioxidant [10], antitumor [11], and antimalarial [12-13] effects. Similarly, coumarin and its derivatives possess significant biological potential and have been investigated for their antimicrobial [14], antioxidant [15], anticoagulant, anti-inflammatory [16], anticancer, and enzyme-inhibitory properties.

The combination of these two pharmacophores into a single molecular framework can provide a chalcone-coumarin hybrid, potentially producing enhanced or complementary biological effects. The incorporation of metal ions into

biologically active organic ligands is an established strategy in coordination chemistry and medicinal chemistry. Metal complexation can modify the electronic structure, lipophilicity, solubility, stability, and overall biological behavior of a ligand. The synthesis of metal complexes of chalcone-coumarin hybrids generally involves the preparation of the hybrid ligands followed by its reaction with selected metal salts under appropriate conditions. Functional groups such as carbonyl, phenolic hydroxyl, or other donor atoms present in the hybrid ligands can participate in coordination with metal ions, resulting in mononuclear complexes with different geometries and coordination environments. The synthesized complexes can be characterized using spectroscopic and analytical techniques such as FT-IR, UV-Visible spectroscopy, NMR spectroscopy, Mass spectroscopy, elemental analysis and single crystal X-Ray diffraction.

2. Methods

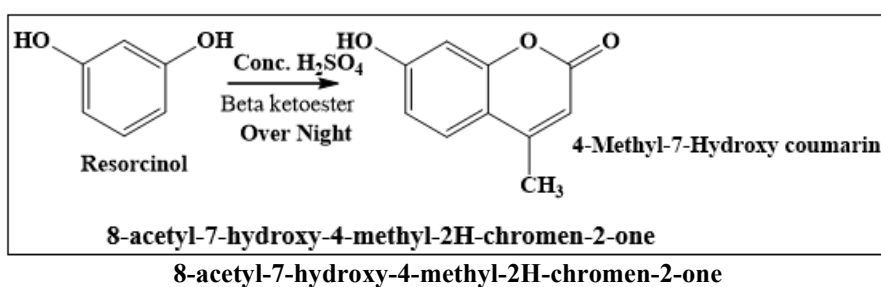


Scheme-1

Step. 1. Synthesis of 2-Acetyl-7-hydroxy-4-methyl coumarin

In this synthesis Resorcinol refluxed with Beta ketoester overnight in acidic medium (with Conc. Sulphuric acid H_2SO_4) to form 4-methyl-7-hydroxy coumarin. This 4-methyl-7-hydroxy coumarin further reacts with acetyl

chloride (CH_3COCl) in the presence of pyridine (a basic medium) 4-methyl-7-acetoxy coumarin. This 4-methyl-7-acetoxy coumarin further undergoes re-arrangement in the presence of Anhyd. AlCl_3 (**Fries Re-arrangement**) to form 8-acetyl-7-hydroxy-4-methyl-2H-chromen-2-one.



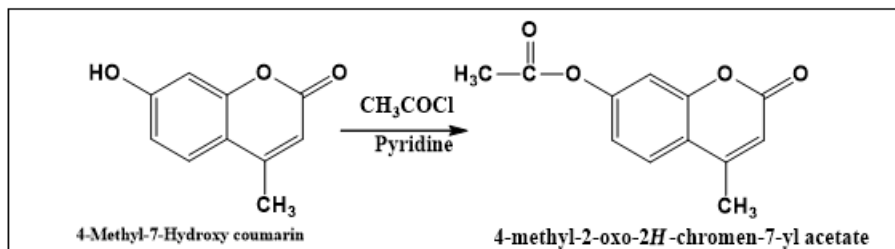
Chemical Formula: $\text{C}_{10}\text{H}_8\text{O}_3$

Exact Mass: 176.05

Molecular Weight: 176.17

m/z: 176.05 (100.0%), 177.05 (11.0%), 178.05 (1.2%)

Elemental Analysis: C, 68.18; H, 4.58; O, 27.25



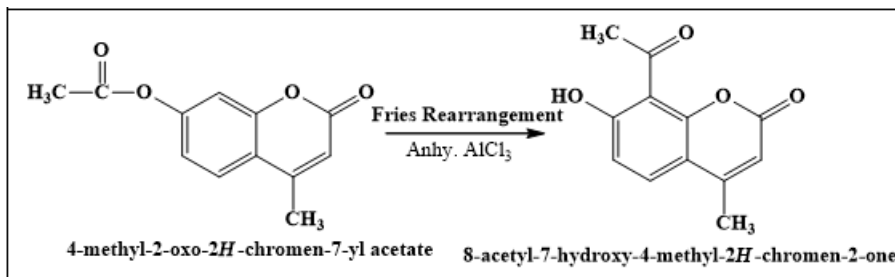
Chemical Formula: $\text{C}_{12}\text{H}_{10}\text{O}_4$

Exact Mass: 218.06

Molecular Weight: 218.21

m/z: 218.06 (100.0%), 219.06 (13.2%), 220.06 (1.6%)

Elemental Analysis: C, 66.05; H, 4.62; O, 29.33



Chemical Formula: $\text{C}_{12}\text{H}_{10}\text{O}_4$

Exact Mass: 218.06

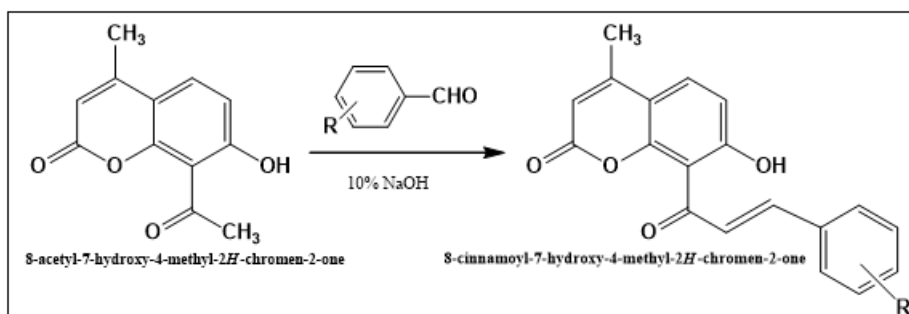
Molecular Weight: 218.21

m/z: 218.06 (100.0%), 219.06 (13.2%), 220.06 (1.6%)

Elemental Analysis: C, 66.05; H, 4.62; O, 29.33

Step 2: Synthesis of Mono chalcones

The mono chalcones were synthesized when 8-acetyl-7-hydroxy-4-methyl-2H-chromen-2-one is subjected to Aldol Condensation with substituted benzaldehyde. This reaction is carried out in the presence of basic medium (10% NaOH solution).



Chemical Formula: $\text{C}_{19}\text{H}_{14}\text{O}_4$

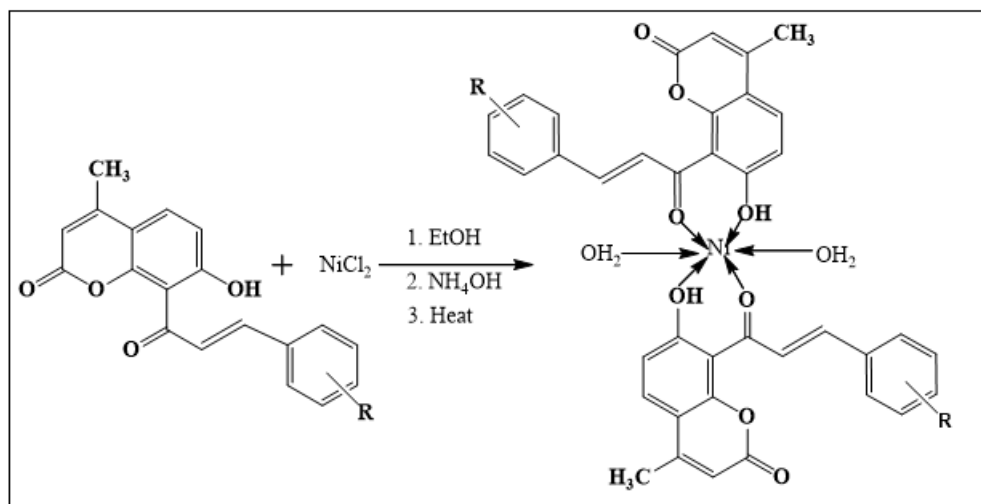
Exact Mass: 306.09

Molecular Weight: 306.31

m/z: 306.09 (100.0%), 307.09 (20.7%), 308.10 (2.1%)

Elemental Analysis: C, 74.50; H, 4.61; O, 20.89

Step 3: Synthesis of metal complexes of Chalcone with metals

**Table 1**

S. No	R	mp(⁰ C)
1	H	87-89
2	2-CH ₃	112-114
3	4-CH ₃	105-106
4	2-OCH ₃	124-125
5	4-OCH ₃	119-120
6	2-Cl	140-141
7	4-Cl	136-137
8	2,4-Cl ₂	148-150
9	3-NO ₂	165-170
10	4-NO ₂	150-152

3. Results and Discussion

Bis(E) 8-cinnamoyl-7-hydroxy-4-methyl-2H-chromen-2-one Nickel

Yellow solid; IR: 3400 (O-H Stretching), 3010 (C-H arom), 2980 (C-H aliph), 1690 (C=O), 1410 (C=C), 495 (Ni-O stretching) cm⁻¹; ¹H NMR (DMSO-d₆): δ 11.0 (s, 1H –OH proton), 7.99-6.49 (m, 6H, aromatic proton), 7.40 (d, 1 H Ethylene proton), 5.25-5.98 (dd, 2H, methylene proton), 2.53 (s, 3H, methyl proton); ¹³C NMR (δ ppm): δ 19.0, 112.3, 112.9, 115.8, 117.2, 127.5, 127.9, 128.4, 128.8, 132.5, 135.0, 145.9, 150.8, 152.3, 158.5, 160.6, 193.9. Chemical Formula: C₃₈H₃₂NiO₁₀ Exact Mass: 706.13, Molecular Weight: 707.35 m/z: 706.13 (100.0%), 707.14 (41.8%), 708.13 (38.5%), 709.13 (17.7%), 708.14 (10.6%), 710.13 (6.8%), 710.14 (3.3%), 711.13 (2.3%), 709.14 (2.1%), 712.13 (1.9%), Elemental Analysis: C, 64.52; H, 4.56; Ni, 8.30; O, 22.62

Bis(E)-7-hydroxy-4-methyl-8-(3-ortho-tolylacryloyl)-2H-chromen-2-one Nickel

Blue solid; IR: 3410 (O-H Stretching), 3045 (C-H arom), 2985 (C-H aliph), 1675 (C=O), 1450 (C=C), 495 (Ni-O stretching) cm⁻¹; ¹H NMR (DMSO-d₆): δ 10.2 (s, 1H –OH proton), 7.91-6.53 (m, 6H, aromatic proton), 5.23-5.90 (dd, 2H, methylene proton), 2.54 (s, 9H, methyl proton); ¹³C NMR (δ ppm): δ 19.2, 112.0, 112.8, 115.6, 117.5, 127.1, 127.8, 128.2, 128.7, 132.2, 135.5, 145.8, 150.4, 152.6, 158.3, 160.5, 193.6. Chemical Formula: C₄₀H₃₆NiO₁₀, Exact Mass: 734.17, Molecular Weight: 735.40 m/z: 734.17 (100.0%), 735.17 (44.1%), 736.16 (38.5%), 737.16 (18.3%), 736.17 (11.3%), 738.16 (5.3%), 738.17 (5.2%), 739.16 (2.3%), 740.16 (1.5%), 737.18 (1.3%), 737.17 (1.2%),

739.17 (1.1%), Elemental Analysis: C, 65.33; H, 4.93; Ni, 7.98; O, 21.76

Bis(E)-7-hydroxy-4-methyl-8-(3-para-tolylacryloyl)-2H-chromen-2-one Nickel

Dark Blue solid; IR: 3410 (O-H Stretching), 3045 (C-H arom), 2985 (C-H aliph), 1675 (C=O), 1450 (C=C), 495 (Ni-O stretching) cm⁻¹; ¹H NMR (DMSO-d₆): δ 10.2 (s, 1H –OH proton), 7.86-6.40 (m, 6H, aromatic proton), 5.21-5.88 (dd, 2H, methylene proton), 2.54 (s, 9H, methyl proton); ¹³C NMR (δ ppm): δ 19.0, 112.3, 112.9, 115.8, 117.2, 127.5, 127.9, 128.4, 128.8, 132.5, 135.0, 145.9, 150.8, 152.3, 158.5, 160.6, 193.9. Chemical Formula: C₄₀H₃₆NiO₁₀, Exact Mass: 734.17, Molecular Weight: 735.40 m/z: 734.17 (100.0%), 735.17 (44.1%), 736.16 (38.5%), 737.16 (18.3%), 736.17 (11.3%), 738.16 (5.3%), 738.17 (5.2%), 739.16 (2.3%), 740.16 (1.5%), 737.18 (1.3%), 737.17 (1.2%), 739.17 (1.1%), Elemental Analysis: C, 65.33; H, 4.93; Ni, 7.98; O, 21.76

Bis(E)-7-hydroxy-4-methyl-8-(3-ortho-methoxyphenylacryloyl)-2H-chromen-2-one Nickel

Blue solid; IR: 3442 (O-H Stretching), 3018 (C-H arom), 2952 (C-H aliph), 1689 (C=O), 1420 (C=C), 490 (Ni-O stretching) cm⁻¹; ¹H NMR (DMSO-d₆): δ 10.2 (s, 1H –OH proton), 7.90-6.52 (m, 6H, aromatic proton), 5.23-5.90 (dd, 2H, methylene proton), 3.45 (s, 6H, methoxy proton) 2.54 (s, 3H, methyl proton); ¹³C NMR (δ ppm): δ 19.2, 112.0, 112.8, 115.6, 117.5, 127.1, 127.8, 128.2, 128.7, 132.2, 135.5, 145.8, 150.4, 152.6, 158.3, 160.5, 193.6. Chemical Formula: C₄₀H₃₆NiO₁₂, Exact Mass: 766.16, Molecular Weight: 767.40, m/z: 766.16 (100.0%), 767.16 (44.1%), 768.15 (38.5%), 769.15 (18.3%), 768.16 (11.8%), 770.16 (5.4%), 770.15 (5.3%), 771.15 (2.3%), 772.15 (1.5%), 769.16 (1.4%), 769.17 (1.3%), 771.16 (1.2%), Elemental Analysis: C, 62.60; H, 4.73; Ni, 7.65; O, 25.02

Bis(E)-7-hydroxy-4-methyl-8-(3-para-methoxyphenylacryloyl)-2H-chromen-2-one Nickel

Blue solid; IR: 3440 (O-H Stretching), 3020 (C-H arom), 2951 (C-H aliph), 1697 (C=O), 1421 (C=C), 491 (Ni-O stretching) cm⁻¹; ¹H NMR (DMSO-d₆): δ 10.1 (s, 1H –OH proton), 7.93-6.52 (m, 6H, aromatic proton), 5.23-5.90 (dd, 2H, methylene proton), 3.45 (s, 6H, methoxy proton) 2.54 (s, 3H, methyl proton); ¹³C NMR (δ ppm): δ 19.2, 54.3, 112.1, 112.9, 115.3, 117.1, 127.0, 127.6, 128.5, 128.8, 132.2,

135.5, 145.8, 150.4, 152.6, 158.3, 160.5, 193.6. Chemical Formula: $C_{40}H_{36}NiO_{12}$, Exact Mass: 766.16, Molecular Weight: 767.40, m/z: 766.16 (100.0%), 767.16 (44.1%), 768.15 (38.5%), 769.15 (18.3%), 768.16 (11.8%), 770.16 (5.4%), 770.15 (5.3%), 771.15 (2.3%), 772.15 (1.5%), 769.16 (1.4%), 769.17 (1.3%), 771.16 (1.2%), Elemental Analysis: C, 62.60; H, 4.73; Ni, 7.65; O, 25.02

Bis(E)-7-hydroxy-4-methyl-8-(3-ortho-chlorophenylacryloyl)-2H-chromen-2-one Nickel

Light green solid; IR: 3400 (O-H Stretching), 3025(C-H arom), 2963 (C-H aliph), 1700 (C=O), 1421 (C=C), 495 (Ni-O stretching) cm^{-1} ; 1H NMR (DMSO- d_6): δ 10.5 (s, 1H -OH proton), 8.03-6.75 (m, 6H, aromatic proton), 5.21-5.92 (dd, 2H, methylene proton), 2.53 (s, 3H, methyl proton); ^{13}C NMR (δ ppm): δ 19.2, 113.1, 114.9, 115.8, 117.9, 127.1, 127.6, 128.5, 128.8, 133.2, 135.8, 145.3, 150.9, 152.8, 158.8, 161.5, 193.0. Chemical Formula: $C_{38}H_{30}Cl_2NiO_{10}$, Exact Mass: 774.06, Molecular Weight: 776.24, m/z: 776.05 (100.0%), 774.06 (97.6%), 777.06 (42.7%), 775.06 (40.8%), 778.05 (39.2%), 779.05 (17.3%), 778.06 (11.3%), 776.06 (10.2%), 780.05 (9.5%), 780.06 (3.8%), 781.05 (3.7%), 779.06 (2.3%), 782.05 (1.7%), 777.05 (1.6%), 777.07 (1.1%), Elemental Analysis: C, 58.80; H, 3.90; Cl, 9.13; Ni, 7.56; O, 20.61

Bis(E)-7-hydroxy-4-methyl-8-(3-para-chlorophenylacryloyl)-2H-chromen-2-one Nickel

Light green solid; IR: 3405 (O-H Stretching), 3029(C-H arom), 2966(C-H aliph), 1705 (C=O), 1430 (C=C), 495 (Ni-O stretching) cm^{-1} ; 1H NMR (DMSO- d_6): δ 10.2 (s, 1H -OH proton), 8.00-6.73 (m, 6H, aromatic proton), 5.15-5.72 (dd, 2H, methylene proton), 2.50 (s, 3H, methyl proton); ^{13}C NMR (δ ppm): δ 19.4, 115.1, 115.9, 116.8, 117.9, 128.1, 128.6, 129.5, 130.8, 133.2, 135.5, 145.1, 150.5, 152.0, 158.6, 161.3, 193.8. Chemical Formula: $C_{38}H_{30}Cl_2NiO_{10}$, Exact Mass: 774.06, Molecular Weight: 776.24, m/z: 776.05 (100.0%), 774.06 (97.6%), 777.06 (42.7%), 775.06 (40.8%), 778.05 (39.2%), 779.0 (17.3%), 778.06 (11.3%), 776.06 (10.2%), 780.05 (9.5%), 780.06 (3.8%), 781.05 (3.7%), 779.06 (2.3%), 782.05 (1.7%), 777.05 (1.6%), 777.07 (1.1%), Elemental Analysis: C, 58.80; H, 3.90; Cl, 9.13; Ni, 7.56; O, 20.61

Bis(E)-7-hydroxy-4-methyl-8-(3-(2,4-dichloro)-phenylacryloyl)-2H-chromen-2-one Nickel

Green solid; IR: 3460 (O-H Stretching), 3015(C-H arom), 2955(C-H aliph), 1730 (C=O), 1425 (C=C), 490 (Ni-O stretching) cm^{-1} ; 1H NMR (DMSO- d_6): δ 10.9 (s, 1H -OH

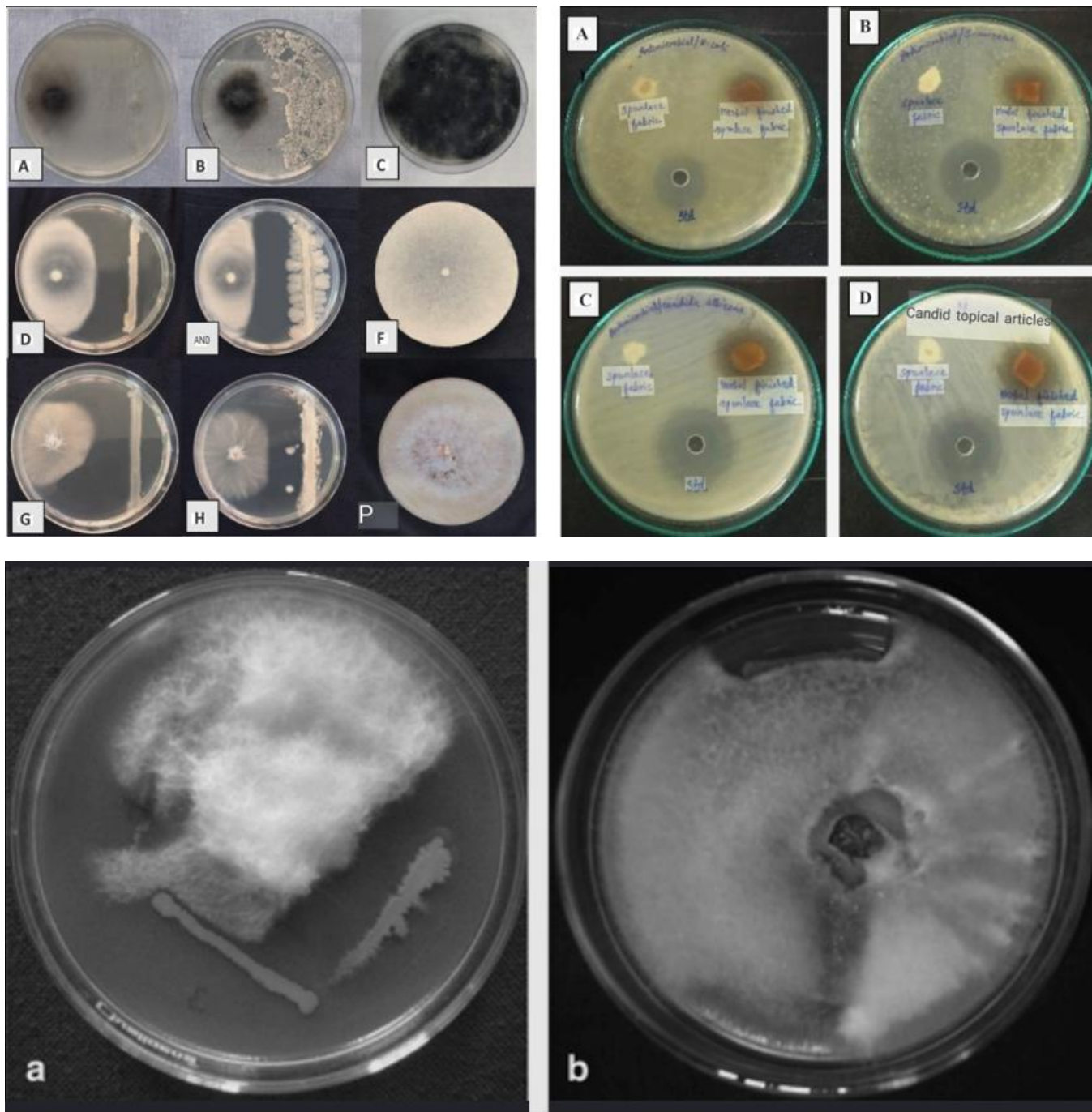
proton), 7.85-6.46 (m, 5H, aromatic proton), 5.15-5.72 (dd, 2H, methylene proton), 2.50 (s, 3H, methyl proton); ^{13}C NMR (δ ppm): δ 19.4, 114.5, 115.6, 116.5, 117.3, 128.0, 128.8, 129.9, 131.2, 133.7, 135.6, 145.6, 150.8, 152.3, 159.9, 161.2, 193.8. Chemical Formula: $C_{38}H_{28}Cl_4NiO_{10}$, Exact Mass: 841.98, Molecular Weight: 845.13, m/z: 843.98 (100.0%), 845.97 (89.2%), 841.98 (77.0%), 844.98 (53.5%), 847.97 (34.5%), 842.98 (31.9%), 843.97 (29.7%), 846.98 (23.3%), 846.97 (17.2%), 848.97 (15.2%), 845.98 (13.9%), 847.98 (10.2%), 849.97 (9.5%), 843.99 (6.6%), 850.97 (3.8%), 849.98 (3.2%), 848.98 (2.0%), 844.99 (1.5%), 851.97 (1.5%), 844.97 (1.3%), 846.99 (1.2%), Elemental Analysis: C, 54.00; H, 3.34; Cl, 16.78; Ni, 6.94; O, 18.93

Bis(E)-7-hydroxy-4-methyl-8-(3-meta-nitrophenylacryloyl)-2H-chromen-2-one Nickel

Yellowish Green solid; IR: 3445 (O-H Stretching), 3005(C-H arom), 2950(C-H aliph), 1725 (C=O), 1420 (C=C), 492 (Ni-O stretching) cm^{-1} ; 1H NMR (DMSO- d_6): δ 10.5 (s, 1H -OH proton), 7.80-6.45 (m, 5H, aromatic proton), 5.10-5.70 (dd, 2H, methylene proton), 2.53 (s, 3H, methyl proton); ^{13}C NMR (δ ppm): δ 19.2, 114.2, 115.8, 116.3, 117.5, 128.5, 128.9, 130.9, 131.8, 133.3, 135.7, 146.6, 151.8, 152.9, 160.9, 161.2, 192.2. Chemical Formula: $C_{38}H_{30}N_2NiO_{14}$, Exact Mass: 796.11, Molecular Weight: 797.34, m/z: 796.11 (100.0%), 797.11 (42.0%), 798.10 (38.5%), 799.10 (18.0%), 798.11 (11.8%), 800.10 (7.3%), 800.11 (3.4%), 801.10 (2.3%), 802.10 (2.0%), 799.11 (1.4%), 799.12 (1.2%), 801.11 (1.1%), Elemental Analysis: C, 57.24; H, 3.79; N, 3.51; Ni, 7.36; O, 28.09

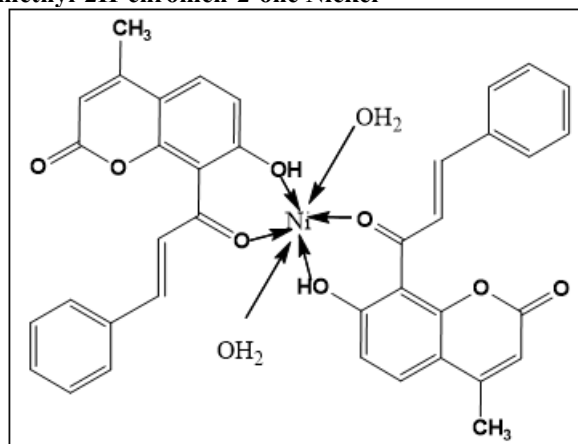
Bis(E)-7-hydroxy-4-methyl-8-(3-para-nitrophenylacryloyl)-2H-chromen-2-one Nickel

Yellowish Green solid; IR: 3448 (O-H Stretching), 3005(C-H arom), 2925(C-H aliph), 1710 (C=O), 1415 (C=C), 490 (Ni-O stretching) cm^{-1} ; 1H NMR (DMSO- d_6): δ 10.5 (s, 1H -OH proton), 7.80-6.42 (m, 5H, aromatic proton), 5.13-5.75 (dd, 2H, methylene proton), 2.55 (s, 3H, methyl proton); ^{13}C NMR (δ ppm): δ 19.4, 114.5, 115.6, 116.5, 117.3, 128.0, 128.8, 129.9, 131.2, 133.7, 135.6, 145.6, 150.8, 152.3, 159.9, 161.2, 193.8. Chemical Formula: $C_{38}H_{30}N_2NiO_{14}$, Exact Mass: 796.11, Molecular Weight: 797.34, m/z: 796.11 (100.0%), 797.11 (42.0%), 798.10 (38.5%), 799.10 (18.0%), 798.11 (11.8%), 800.10 (7.3%), 800.11 (3.4%), 801.10 (2.3%), 802.10 (2.0%), 799.11 (1.4%), 799.12 (1.2%), 801.11 (1.1%), Elemental Analysis: C, 57.24; H, 3.79; N, 3.51; Ni, 7.36; O, 28.0



Images of Antifungal activities of this series of compounds

Bis(E) 8-cinnamoyl-7-hydroxy-4-methyl-2H-chromen-2-one Nickel



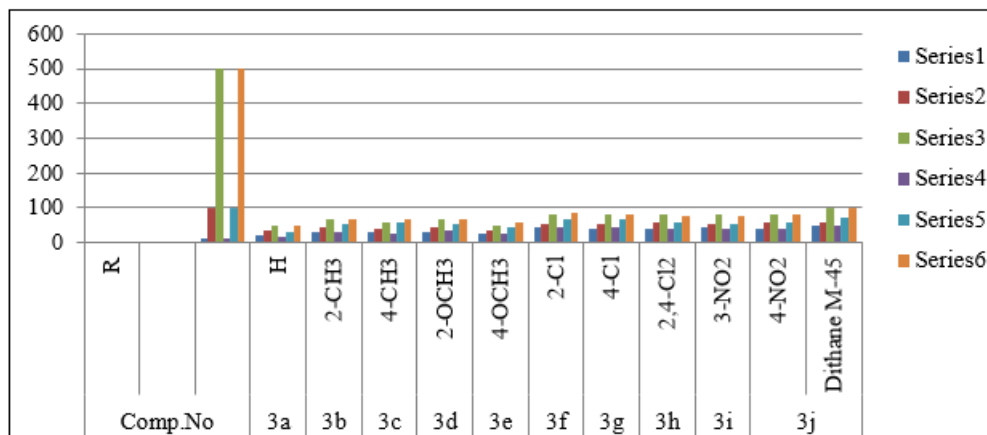
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Table 2

Comp. No	R	Percent Mycelial Inhibition			Compound dose (ppm)		
		Aspergillus Flavus			Candida Albicans		
		10	100	500	10	100	500
3a	H	20	35	51	18	31	48
3b	2-CH ₃	30	45	67	31	54	69
3c	4-CH ₃	29	42	60	27	56	66
3d	2-OCH ₃	32	46	67	34	55	68
3e	4-OCH ₃	24	37	50	25	43	58
3f	2-Cl	46	55	83	45	69	84
3g	4-Cl	41	53	79	44	66	80
3h	2,4-Cl ₂	40	56	80	41	58	78
3i	3-NO ₂	43	55	79	40	55	75
3j	4-NO ₂	41	59	81	42	58	79
	Dithane M-45	50	60	100	50	72	100



4. Conclusion

The synthesis of metal complexes based on chalcone-coumarin hybrid ligands represents a promising approach for the development of biologically active coordination compounds. The molecular hybridization of chalcone and coumarin combines two pharmacologically important scaffolds and may provide enhanced biological properties through synergistic effects. Both chalcone coumarin hybrids and their metal complexes have been associated with important activities such as antimicrobial, antioxidant, anticancer, antimalarial and enzyme inhibitory effects. The 2-Cl, 2,4-Cl₂ and methyl, methoxy substituted ligands shows promising antibacterial and antifungal activities against different bacteria and fungi at different concentrations.

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