

Potential Source Rocks of Ngaye Clayed Deposit (Northern Cameroon)

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Abstract: *The aim of this paper is to identify the parent rocks of Ngaye clay deposit. To this end, mineralogical and geochemical analysis has been done, on the lithological formation found around the deposit and on the samples of clay deposit. The mineralogical analysis have been done by X-ray diffraction, specific treatments ethylene glycol, the hydrazine test and heating between 550°C and 600°C have confirmed the mineralogy of clay materials. For the geochemical analysis major, trace and rare earth elements have been analyzed. The rocks consist of granits, gneisses, amphibolites and quartzites. The metamorphic rock is mainly composed of amphibole, plagioclase and quartz. On its part, granits are constituted by plagioclase, feldspar, biotite and garnet. Mineralogical results of clays samples showed that it made by kaolinite, feldspar, quartz and muscovite. Positive correlation between major elements and barium in the clay sample compared to rocks suggested that gneisses could be the source rocks of these clay materials. The high contents of REE in clay compared to rocks contents, shows that there would have been an upgrading of REE during the supergene alteration of parent rocks. CIA values showed that the lithological formation have undergone moderate chemical alteration. This is in accordance with mineralogical procession of clay deposit, and the climate of the region.*

Keywords: clayed deposit, Ngaye, source rocks

1. Introduction

The clay material is a product resulting from the responses of their source materials. The processes of weathering, acting on pre-existing minerals, or material to form clay minerals. Therefore the chemistry of lithology determines the kind of clay minerals in a soil [1] [2] [3]. Highly basic parent material or poor drainage induces montmorillonite formation. Kaolinite-halloysite formation is favoured by a highly based depleting environment such as high rainfall, good drainage and high permeability. Intermediate conditions are conducive to vermiculite formation either by synthesis or mica alteration. Clay minerals in a soil may originate by inheritance from parent material, alteration and degradation of primary minerals, and synthesis [4][5][6] [7]. These mechanisms operate under different environmental conditions together with the process of the translocation of material result in soil clay mineral composition becoming a function of soil depth. In this paper it is desired to determine the potential parent rock of Ngaye clay deposit. Attention, therefore, will be focused on the lower horizon of clay profile sampled and the rock surrounding the clay deposit (Figure 1). In the vast majority of cases this is a C-horizon, the soil parent material. Despite the moderate alteration, It is here that the contribution of the parent rock would be least complicated by weathering processes.

2. Materials and Methods

The investigated Ngaye clayed deposits are located at 7°11'04"N and 14°58'50"E (Figure 2). The climate is semi-arid, with a mean annual rainfall of 970 mm, and a mean annual temperature of 27 °C. The lithologies include granits, gneisses, quartzites and amphibolites, the surface rock ranged from Pre-Cambrian to Tertiary-Pliocene [8]. Gneisses appears like flagstones and have foliated structure. They are gray and composed of alternating white beds rich in feldspars between 1 to 4 mm thick and dark beds rich in amphibole (Figure 3a). Amphibolites appears underground

or sometimes in the middle of gneiss (Figure 3b). Quartzites are less represented in the area, and are lined with phyllitous minerals consisting mainly of quartz (Figure 3c). The granits are light gray and also appear as flagstones in the area (Figure 3d). For mineralogical and geochemical analyses, 20 g of powders of five samples of rock and 4 samples of clay material samples were treated with cold 1 M HCl and H₂O₂ to remove organic matter. Adding sodium hexametaphosphate deflocculated the samples. The mineral compositions were determined by X-ray Diffraction (XRD) analysis. Samples for geochemical analysis were crushed using a jaw crusher with steel plates, and pulverized in a ball mill made of 99.8% Al₂O₃. Powders were first heated at 105 °C in the presence of oxygen to drive off remaining volatile components and oxidized Fe. The loss on ignition (LOI) was determined at 1000 °C on dry (105°C) samples. X-ray Fluorescence (XRF) was used to determine the major element concentrations after LOI measurement. The sample powders were first ignited and melted with a lithium tetraborate flux and then analyzed with a Rigaku RIX-3000 wavelength dispersive X-ray Fluorescence spectrometer. Inductively Coupled Plasma-Mass Spectrometry (ICP-MS) was used to quantify trace and rare-earth element concentrations. Sample powders for ICP-MS analyses were digested with a mixture of HCl and HClO₄ at 120 °C in sealed Teflon containers for one week, and then rinsed out of their containers with dilute HNO₃ and dried. The residue was again dissolved in a mixture of three acids (HNO₃, HCl and HF) at 100°C. Sample solutions were analyzed in a Perkin Elmer Elan 9000 ICP-MS instrument. The geochemical results (major, trace and rare-earth elements) of rocks samples were compared to clay samples.

3. Results and Discussion

Mineralogy

Metamorphic rocks of Ngaye are constituted by amphibole, plagioclase and feldspars. Amphibolite also has sphère and opaque minerals (Figure 4a). Gneiss has orthose in their

mineralogy, and has granoblastic texture showing their magmatic origin. (Figure 4b). Amphibolite are also granoblastic, however granites has porphyritic grained texture (Figure 4c). The absence of kyanite in these metamorphic rocks showed that they can results to the metamorphism of magmatic rocks [9]. The collection of clays samples are made by kaolinite, smectites, feldspars and quartz. They are also few presences of muscovite, goethite and hematite. Specific analysis with glycol, hydrazine and heating at 550°C [10] confirmed the presence of smectites, kaolinite and illite (Figure 5). These clay samples are sandy clays with clay content between 15-43%, and sand between 48 and 58% (Table I). The uniform texture would be related to hydrolytic processes that affect source rocks [11] [12]. The presence of different types of clay minerals could be related to the several types of rocks and hydrolytic process [13].

Geochemistry

SiO₂ contents of gneisses and amphibolites are moderate, compared to higher contents of this element in quartzites and granites (Table II). However Al₂O₃ and Fe₂O₃ contents evolve in the opposite direction (table). The alkaline earth contents (MgO, CaO) evolve in the same way as Fe-Al pair (table). Therefore, alkalis contents are close to silicium contents, there are slightly up in amphibolite (table). MnO and P₂O₅ has low contents in these rocks (Table II). Positive trend between SiO₂ with some oxides like Al₂O₃, Na₂O and K₂O (Figure 6) confirmed their higher presence in the rocks. However, negative trends with the other oxides suggested that feldspars are little weathered [14] [15]. Trace elements contents of granites and gneisses are similar (Figure 7), suggesting that gneisses of Ngaye are situated in grinding area in contact with granites. Positive trend of some trace elements with silica shows a compatibility with these elements. High contents of SiO₂ in clay samples (Table III) could be correlate to quartz content in potential source rocks [16] [17]. Moderate content of MgO could be related to the presence of smectites from weathering of mafic rocks [11]. Moderate contents of CaO, Na₂O and K₂O (Table 2) are correlated to the presence of illite or feldspars [17] [18] and indicators of immaturity of clay sediments. Variation of Na₂O content in clay samples could be related to the solubility of sodium [19]. High content of barium in this material could be related to its content in parents rocks [8].

Figures

High contents of chromium and vanadium in this clay are probably related to their substitution in clay minerals [19] [20] [21]. The values of Chemical Index of Alteration (CIA) confirmed that the potential source rocks (Table II) have undergone moderate chemical alteration [22] [23]. This assumption is confirmed by rare earth patterns with the spider diagram (Figure 8). There are high contents of Rare Earth Elements (REE) in clays sample than in the potential source rock, this suggested that there would have been an upgrading of REE during the supergene alteration (Table IV). Spider diagram of REE of clay samples normalized with PAAS (Post Archaean Australian Average Shale) of McLennan, 1989 [24] and with the potential source rocks, showed that fractionation of REE, enrichment in LREE compared to HREE. There is also a similarity between spider diagram of clay samples normalized with PAAS and with Ngaye gneisses (Figure 8). These suggested that gneisses could be the parent source rocks of Ngaye clay deposit. The gneisses cover a fairly large area and amphibolites and granites appear as enclaves and veins in gneissic bedrock.

4. Conclusion

The presence of different types of clay minerals in Ngaye clay deposit confirms the presence of several types of rocks as founding in the surrounding area. Major elements and some trace elements contents of clay samples and potential source rocks associated with CIA values confirmed the moderate weathering of source rocks. The normalization of rare earth patterns of clays deposits with the potential source rocks showed that gneisses could be the parent source rocks of Ngaye clay deposit.

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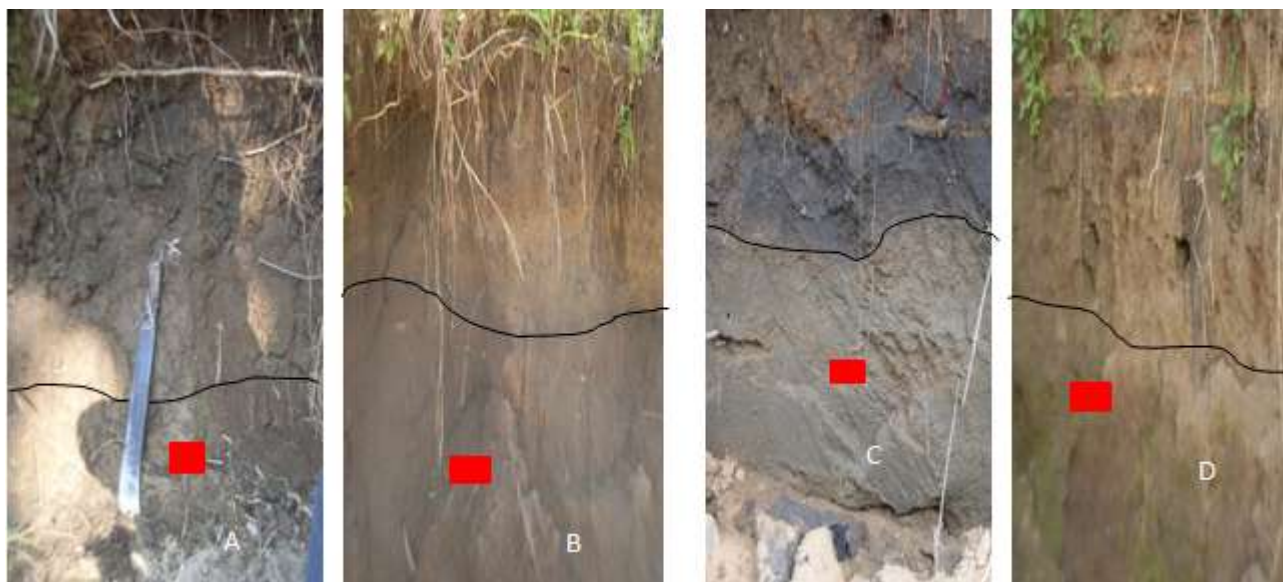


Figure 1: Profiles representing clay samples .A for sample A,B for sampleB, C for sample C and D for sample D

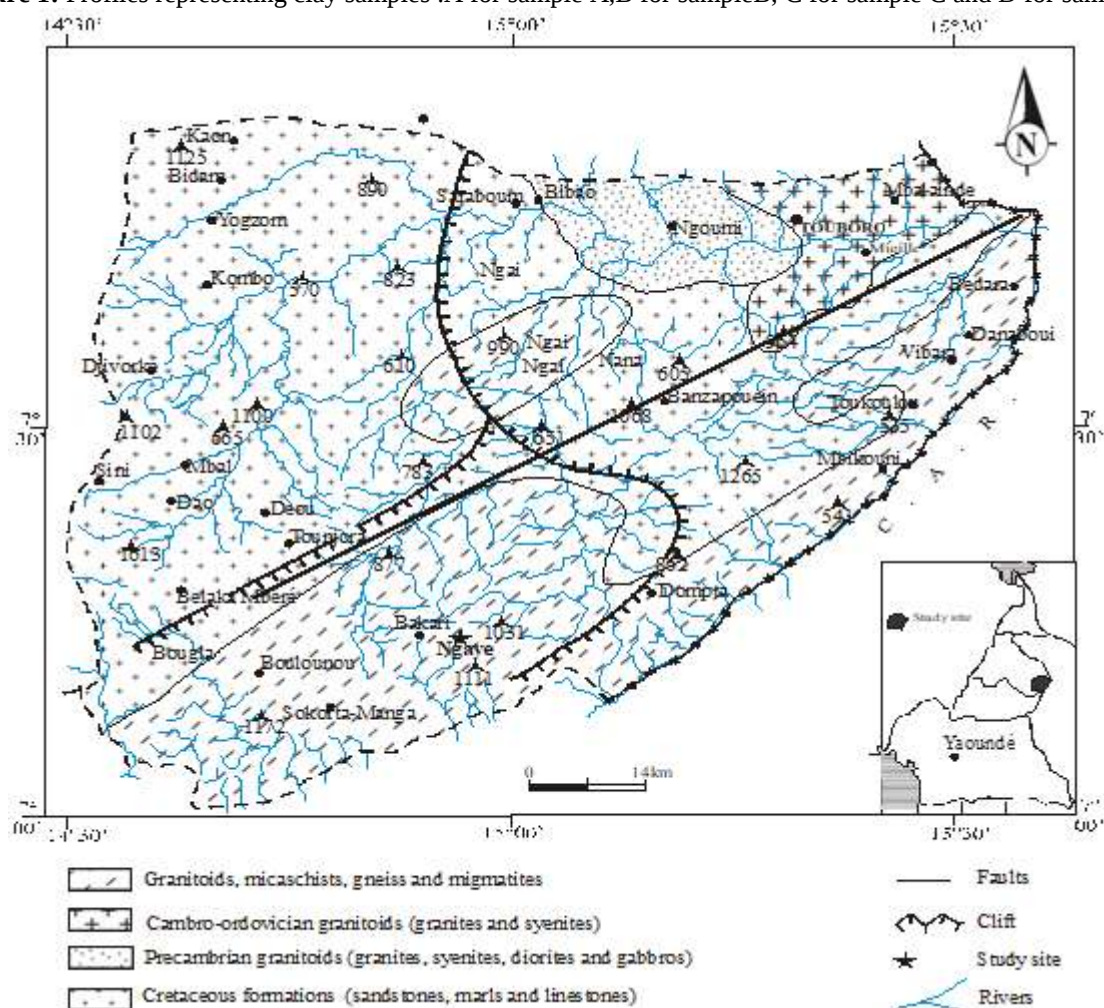


Figure 2: Location map combined with geology of the study site



Figure 3: Pictures of potentials rock source: a) Gneiss ;b) Amphibolite ; c)Quartzite; d) Granite

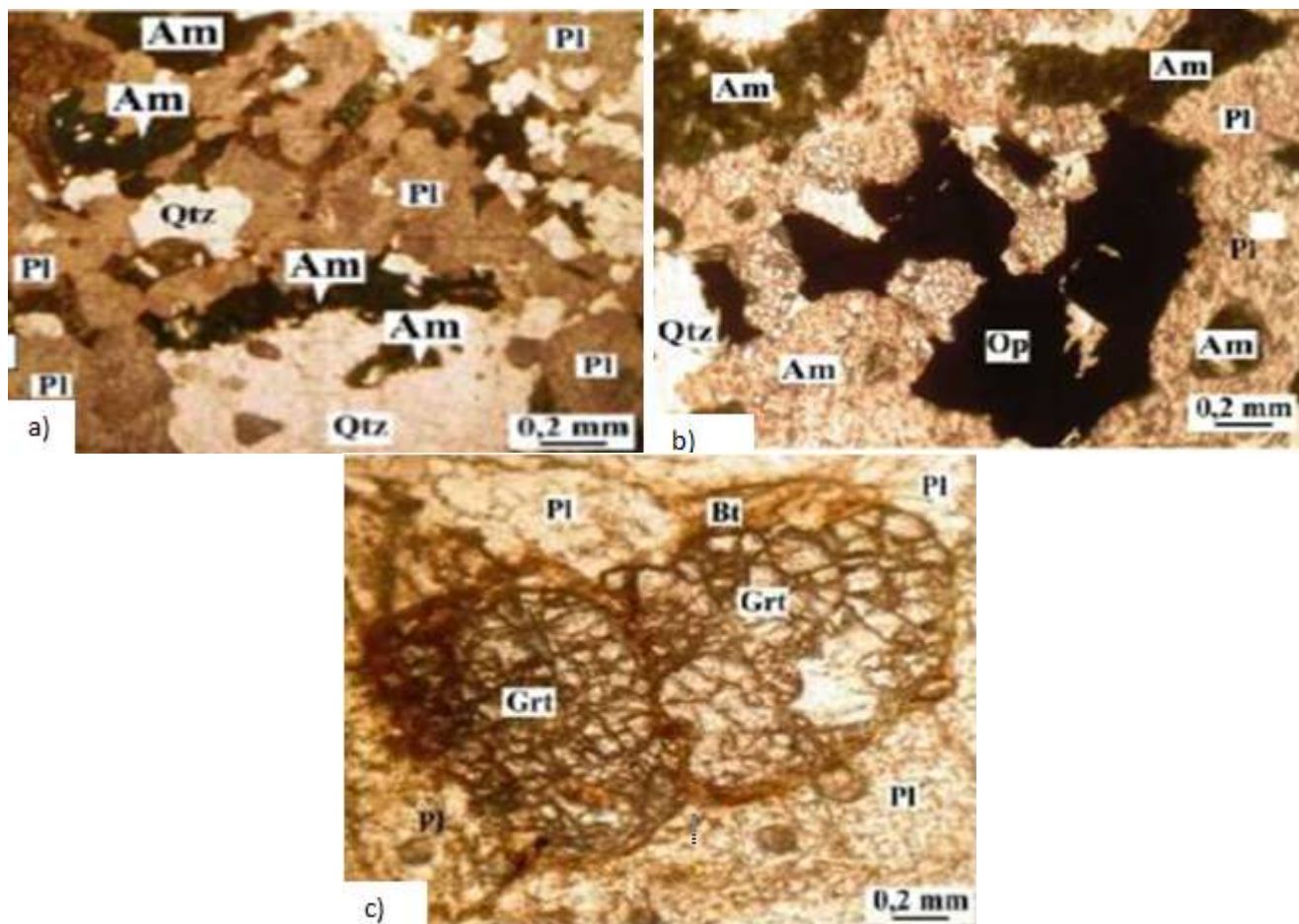


Figure 4: Tin sections of rocks observed with polarized and analyzed light: a) amphibolite; b) gneiss; c) granit; Am=amphibole; Qtz=quartz; Pl=plagioclase; Bt=biotite; Grt=garnet; Op= opac

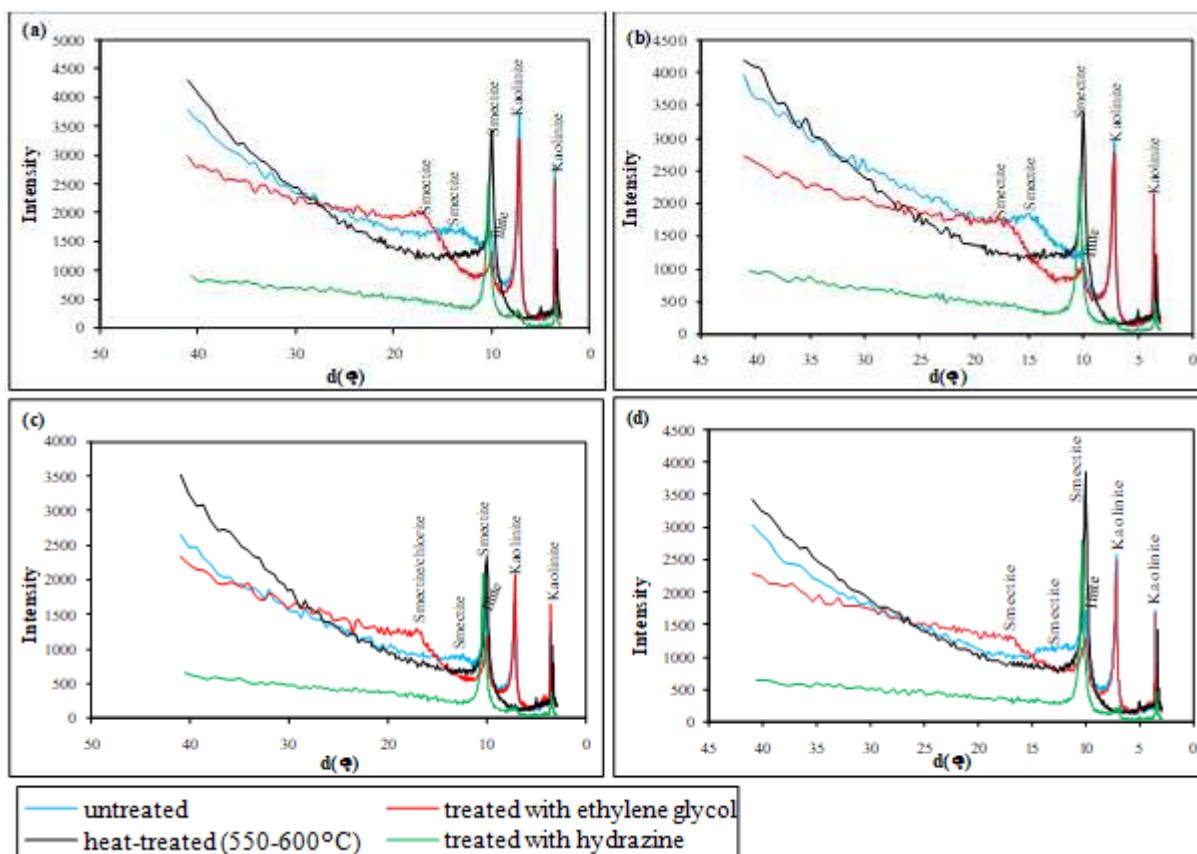


Figure 5: XDR patterns of the samples; (a) for sample A; (b) for sample B; (c) for sample C; (d) for sample D

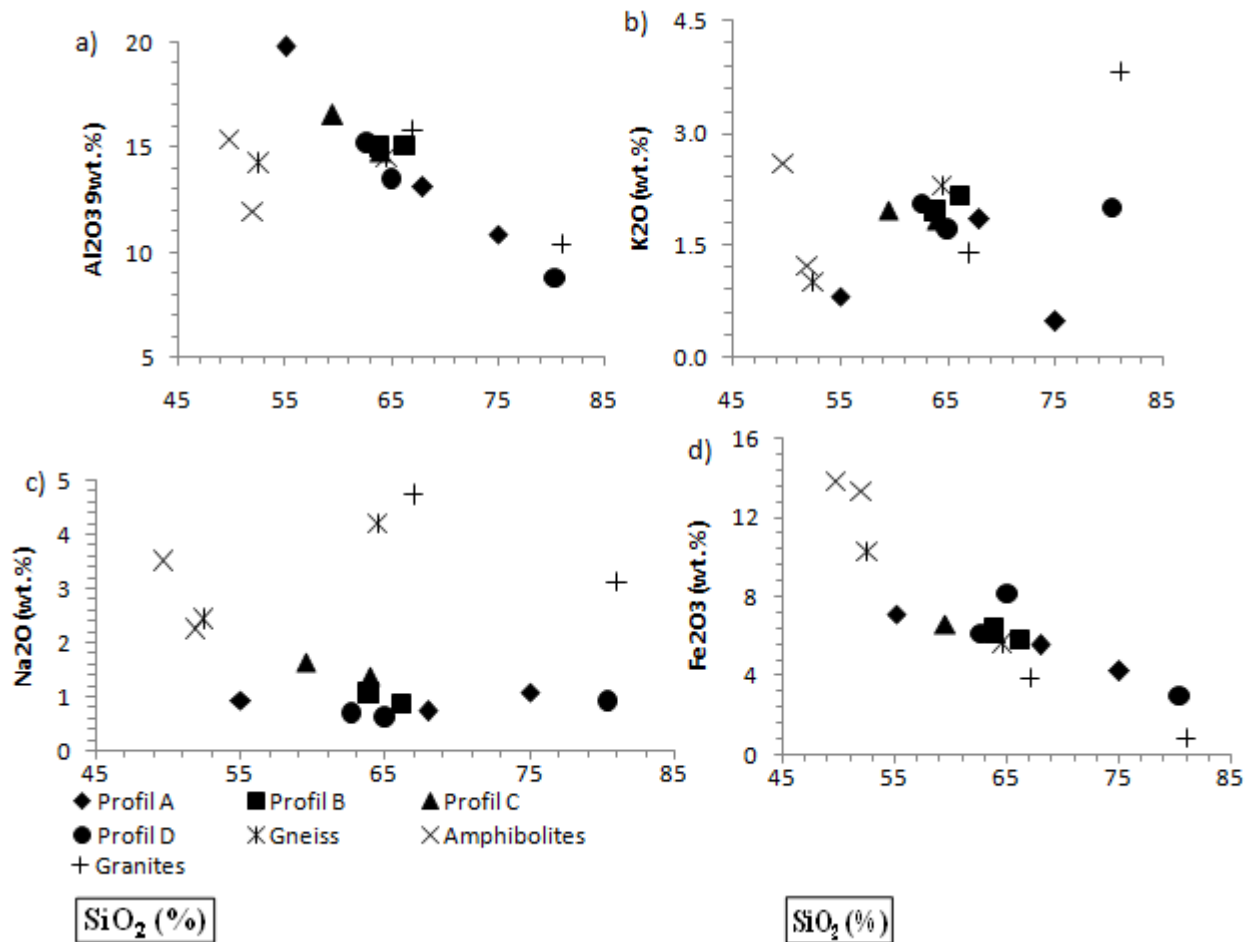


Figure 5: Geochemical Harker diagram of some majors elements of rocks and clays samples a) SiO_2 wt%vs Al_2O_3 wt%; b) SiO_2 wt%vs Na_2O wt%;c) SiO_2 wt%vs K_2O wt%; SiO_2 wt%vs Fe_2O_3 wt%.

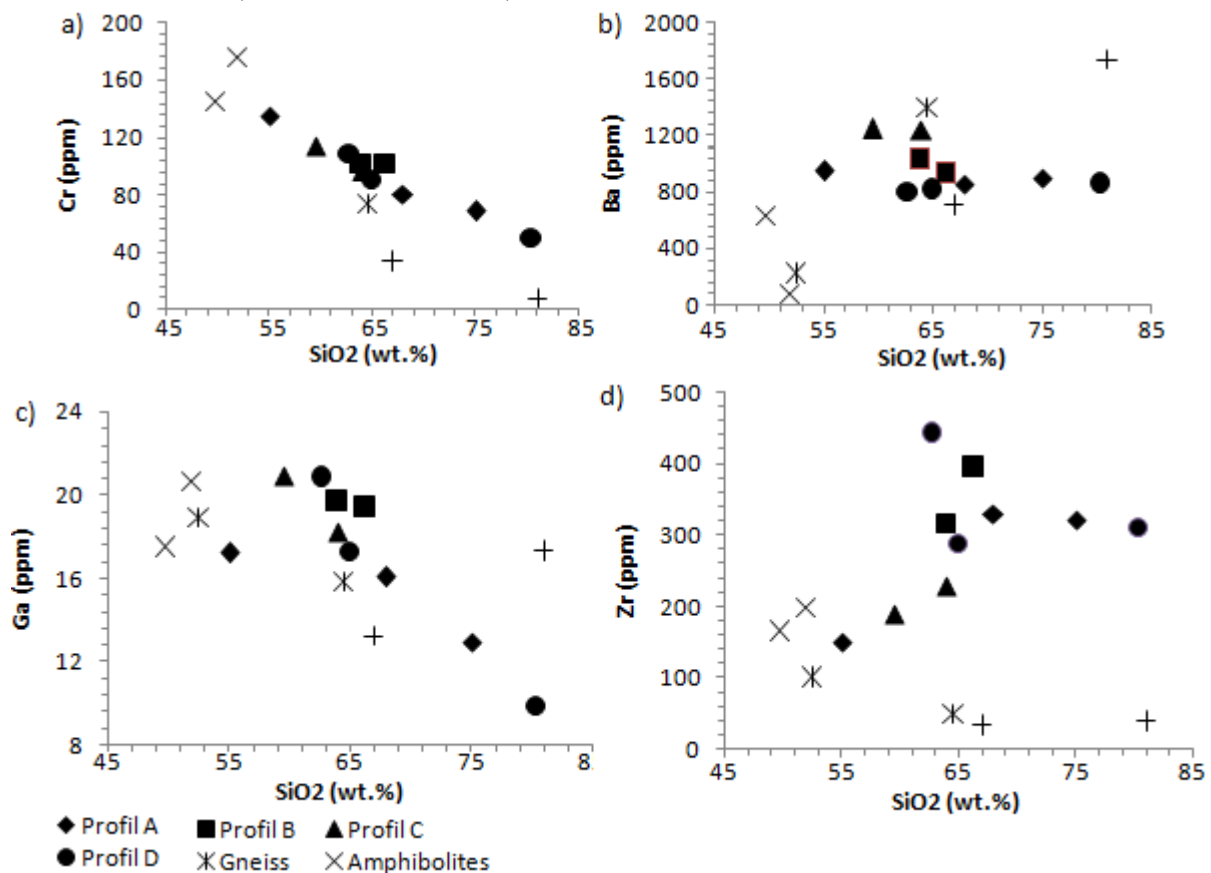


Figure 6: Geochemical Harker diagram of some trace elements of rocks and clays samples a) SiO_2 wt% vs Cr ppm; b) SiO_2 wt%vs Ba ppm; c) SiO_2 wt%vsGa ppm; SiO_2 wt%vs Zr ppm.

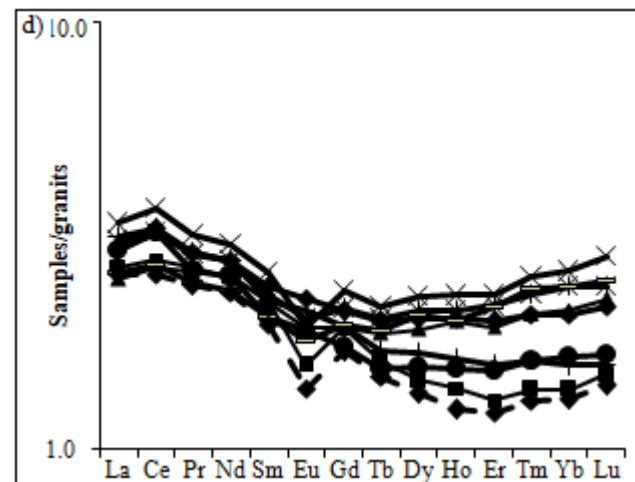
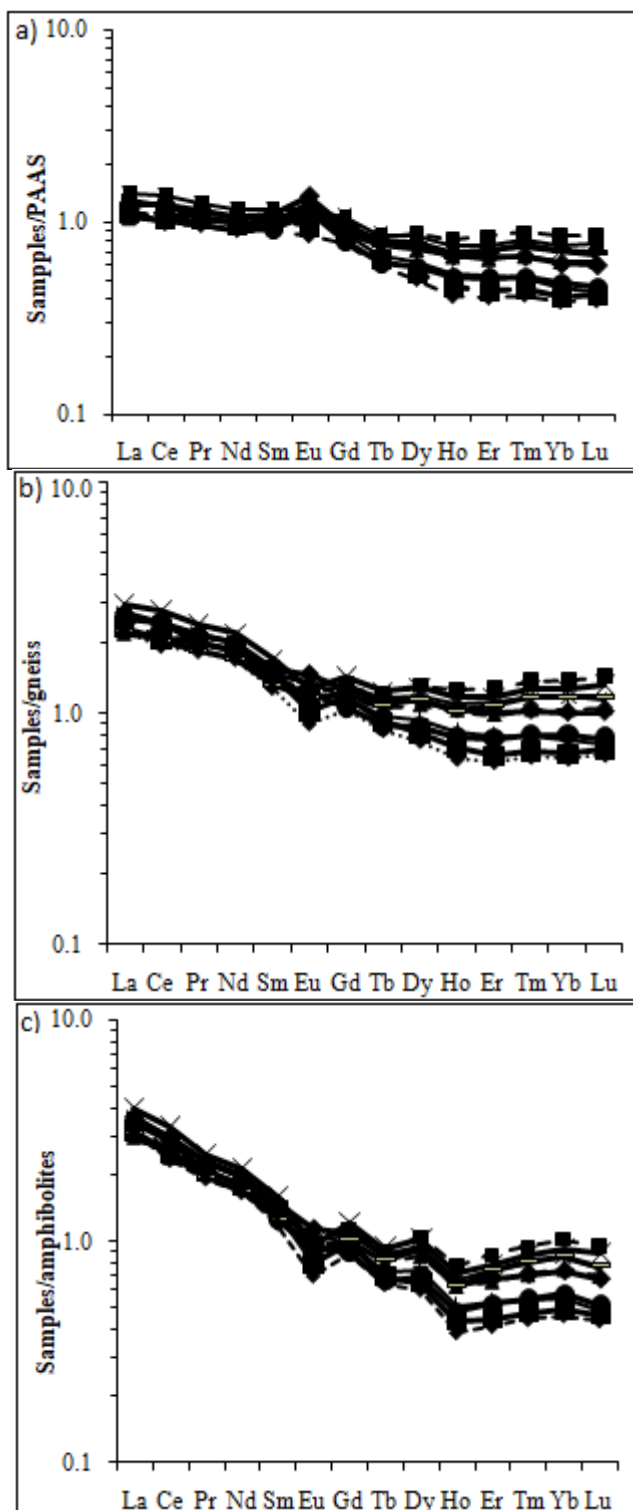


Figure 7: Geochemical spider diagram of clay samples :a)normalized with PAAS; b)normalized with gneiss; c) normalized with amphibolites; d) normalized with granites

Tables

Table I: Particles sizes distribution (%) and texture of clay samples

Samples	Particles sizes		Texture
	Clay	Sand	
A	43.15	55.49	Sandy clay
B	49.93	52.39	Sandy clay
C	44.63	55.03	Sandy clay
D	41.73	57.85	Sandy clay

Table II: Major elements (%) of rocks and clay samples of Ngaye

	Gneisses		Amphibolites		Quartzites	Granites		Clay samples			
	NGY1	NGY2	NGY3	NGY4	NGY5	NGY6	NGY7	A	B	C	D
SiO ₂	52.53	64.54	51.97	49.75	91.67	81.04	67.02	55.05	66.22	63.97	65.01
Al ₂ O ₃	14.27	14.54	11.94	15.37	3.40	10.37	15.86	19.80	15.05	14.77	13.51
Fe ₂ O ₃	10.30	5.67	13.28	13.83	1.36	0.86	3.91	7.16	5.84	6.16	8.11
MnO	0.16	0.10	0.11	0.20	0.01	0.01	0.06	0.07	0.029	0.076	0.067
MgO	5.78	2.90	7.93	3.73	0.10	0.07	1.37	1.21	1.15	1.17	0.93
CaO	10.65	3.95	9.29	7.05	1.32	0.50	3.95	1.04	0.776	1.463	0.636
Na ₂ O	2.44	4.20	2.25	3.51	0.54	3.12	4.74	0.92	0.86	1.37	0.61
K ₂ O	1.00	2.28	1.22	2.59	0.29	3.82	1.39	1.78	0.76	0.67	0.67
TiO ₂	0.41	0.20	1.43	1.61	0.18	0.02	0.25	0.82	2.16	1.83	1.71
P ₂ O ₅	0.07	0.03	0.08	0.19	0.03	0.004	0.06	0.07	0.052	0.053	0.075
LOI	1.87	1.25	1.05	1.93	0.61	0.14	1.39	11.16	6.42	7.14	7.33
Total	99.48	99.64	99.75	99.76	99.51	99.96	99.80	99.08	99.32	98.67	98.66

CIA(%)	-	-	-	-	-	-	-	87.69	79.86	76.00	85.02
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Table III: Trace elements (ppm) of rocks and clay samples of Ngaye

	Gneisses		Amphibolites		Quartzites	Granites		Clay samples			
	NGY1	NGY2	NGY3	NGY4	NGY5	NGY6	NGY7	A	B	C	D
Cr	411,0	73	34	145,0	30,0	07,0	33,0	135,00	101,00	96,00	90,00
V	146,6	64,6	175,2	277,7	19,2	7,6	56,9	116,00	110,80	93,40	97,90
Ni	112,2	42,7	278,7	45,7	10,1	6,6	28	73,20	42,30	47,00	39,40
Zn	91,0	56,0	256,0	117,0	27,0	<dl	42,0	72,00	56,00	57,00	51,00
Cu	18,4	19,5	414,0	26,2	6,5	7,7	8,5	49,50	36,80	35,52	33,90
Co	35,6	15,2	46,1	27,4	2,0	1,8	9,9	24,85	16,34	20,29	21,94
Sc	26,4	16,1	23,9	37,7	3,1	<ld	11,8	14,50	13,60	11,10	13,30
Ba	222,1	1395,6	360,4	630,7	69,4	>1740	704,2	947,90	933,40	1236,90	817,00
Zr	100,0	49,0	76,0	165,0	104,0	38,0	33,0	149,00	396,00	228,00	287,00
Sr	245,4	422,0	198,0	259,8	73,8	107,3	434,2	147,20	131,30	225,90	102,00
Y	20,86	13,87	11,83	42,7	4,2	0,5	13,5	15,90	21,02	14,64	19,37
Li	4,8	2,4	3,0	3,6	3,9	3,4	1,5	26,01	11,60	10,00	10,20
Ga	18,9	15,8	5,7	17,5	4,6	17,3	13,2	17,20	19,41	18,24	17,27
Pb	5,1	9,5	20,6	8,2	4,8	2,8	6,7	41,73	16,90	15,20	14,00
Rb	40,0	34,3	3,9	59,2	10,5	0,1	3,3	11,71	89,27	56,64	73,30
Nb	5,2	3,2	15,0	13,1	9,2	3,3	2,1	3,84	12,88	8,31	11,98
Hf	2,8	1,4	10,8	4,3	2,5	1,0	6,7	0,78	10,11	5,33	7,30
Mo	0,5	0,3	2,2	2,8	1,0	0,3	0,4	14,51	1,41	0,61	2,14
Th	47,3	0,4	1,0	5,5	3,3	3,4	0,7	1,21	20,27	11,22	15,53
U	1,3	0,08	0,9	1,5	0,5	0,1	0,5	0,61	1,89	0,91	1,39
Ta	0,4	0,2	0,62	0,14	0,3	0,1	0,4	1,72	0,78	0,41	0,76
Be	1,5	1,5	0,5	0,9	0,4	1,3	0,4	0,07	1,07	1,25	0,20
Cd	0,1	0,07	3,8	0,1	0,05	0,06	0,02	0,76	0,05	0,05	0,06
Cs	0,07	0,2	0,1	0,3	0,02	0,1	0,3	0,06	0,66	0,48	0,58
Sn	1,3	1,0	0,05	1,9	0,8	<ld	1,0	0,38	1,23	1,24	0,94
W	0,43	<ld	2,0	0,79	0,30	0,05	0,06	0,33	0,39	0,25	0,37
Tl	0,12	0,13	0,22	0,23	0,04	0,33	0,1	0,05	0,39	0,25	0,34

<dl=inferior to detection limit

Table IV: Rare Earth elements (ppm) of rocks and clay samples of Ngaye

	Gneiss		Amphibolites		Quartzites	Granites		Clay samples			
	NGY1	NGY2	NGY3	NGY4	NGY5	NGY6	NGY7	A	B	C	D
La	22.54	14.05	4.97	22.13	8.91	10.43	8.08	48.35	54.37	47.11	41.8
Ce	50.09	29.77	10.31	57.3	17.95	16.34	20.27	99	111.4	98.07	82.37
Pr	5.83	3.32	1.36	7.52	2.14	1.63	2.95	10.16	11.14	9.26	8.62
Nd	22.12	13.64	5.74	31.31	7.78	4.68	12.93	36.61	39.93	33.52	31.1
Sm	4.54	3.2	1.65	6.54	1.41	0.51	3.15	6.07	6.57	5.19	5.16
Eu	0.94	1.08	0.73	1.89	0.28	0.69	0.95	1.49	1.28	1.25	1.19
Gd	4.04	2.96	2.04	6.23	1.07	0.26	2.99	4.5	5.04	3.72	4.19
Tb	0.62	0.46	0.37	0.85	1.15	0.02	0.45	0.62	0.67	0.48	0.59
Dy	3.37	2.83	0.38	7.43	0.85	0.11	2.59	3.55	3.99	2.73	3.52
Ho	0.73	0.55	0.44	1.66	0.16	0.02	0.45	0.67	0.75	0.51	0.68
Er	2.15	1.59	1.12	4.45	0.46	0.05	1.38	1.9	2.2	1.46	2.05
Tm	0.3	0.22	0.14	0.02	0.07	0.01	0.19	0.27	0.33	0.25	0.31
Yb	1.94	1.46	0.82	3.86	0.45	0.06	1.13	1.72	2.17	1.36	2
Lu	0.29	0.22	0.11	0.66	0.07	0.01	0.16	0.26	0.33	0.2	0.3
REE	119.5	75.27	30.18	15.89	23.76	34.82	57.72	215.87	240.17	205.11	248.57
LREE	106.06	64.98	24.16	126.52	22.65	34.28	48.33	202.58	213.29	174.6	233.92
HREE	13.44	10.29	5.42	25.37	1.11	0.54	9.39	13.29	26.88	30.51	14.65
LREE/HREE	7.89	6.31	4.57	4.99	20.41	64.48	5.15	15.24	7.93	5.72	15.96

References

- [1] Keller, W. D. (1956) Clay minerals as influenced by environments of their formation: Bull. Am. Petr. Geol. 40, 2689-2710.
- [2] Barshad.(1966) The effect of a variation in a precipitation on the nature of clay mineral formation in soil from acid and basic igneous rocks. Proc.Int.Clay,167-173.
- [3] Fadil-Djenabou,.Ndjigui, P-D., Mbey, J.A.(2014) Mineralogical and physicochemical characterization of Ngaye alluvial clays (Northern Cameroon) and assessment of its suitability in ceramic production. Journal of Asian Ceramic Society 3, 50-58 .
- [4] Mitchell, B. D. and Mitchell, W. A. (1956). The clay mineralogy of Ayrshire soils and their parent rocks: Clay Minerals Bull. 3, 91-97.

- [5] McKenzie, L. J. & Erickson, A.E.(1954).The use of redox potentials in the studies of soil genesis.*SoilSci. Soc. Amer. Proc.* 18, 481-485.
- [6] Wouatong, G.A.(1997). Mineralogical study of weathering on the Bana complex, West part of Cameroon.*Journal of Sciences*, Hirishima University, Japan, Serie c, vol, II, n°1.
- [7] NgonNgon, G.F., Etame, J., Ntamak-Nida, M.J., Mbog, M.B., Maliengoue, M.M.A., Gerard, M., Yongue-Fouateu, R., Bilong, P. (2012).Geological study of sedimentary clayey materials of the Bomkoul area in the Douala region (Douala sub-basin, Cameroon) for the ceramic industry.*C.R. Geoscience* 344, 366-376.
- [8] Tchameni, R., Pouclet, A., Penaye, J., Ganwa, A.A., Toteu, S.F., (2006). Petrography and geochemistry of the Ngaoundéré Pan-African granitoids in Central North Cameroon: implications for their sources and geological setting. *Journal of African Earth Sciences* 44, 511-529.
- [9] Gardien,V.,Thomson.A.B.,Ulmer.P.(2000)Melting of biotite+ plagioclase+ quartz gneisses: The role of H₂O in the stability of amphibole.*Journal of Petrology*.41,651-666.
- [10] Kunze, G. W. (1955) Anomalies in the ethylene glycol solvation technique: Clays and Clay Minerals 3,88-93
- [11] Ndjigui, P.-D., Beauvais, A., Fadil-Djenabou, S., Ambrosi, J.-P., (2014). Origin and evolution of Ngaye River alluvial sediments, Northern Cameroon: Geochemical constraints. *Journal of African Earth Sciences* 100, 164-178.
- [12] Ndjigui, P.-D., Ebah Abeng, A.S., Ekomane, E., Nzeugang Nzeukou, A., Ngo Mandeng, F.S., Lindjeck, M.M., (2015b). Mineralogy and geochemistry of pseudogley soils and recent alluvial clastic sediments in the Ngog-Lituba region Southern Cameroon: An implication to their genesis. *Journal of African Earth Sciences* 108, 1-14.
- [13] Schroeder, J.E., (1978). Inexpensive high-strength electrical porcelain. *An. Céram. Soc. Bull.* 57, 526.
- [14] Panahi, A., Grant, M., Young, Robert, H., Rainbird, (2000). Behavior of major and trace elements (including REE) during Paleoproterozoic pedogenesis and diagenetic alteration of an Archean granite near Ville Marie Quebec, Canada. *Geochimica et Cosmochimica Acta* 64, 2199-2220.
- [15] Shimizu, H., Masuda, A., (1977) Cerium in chert as an indication of marine environment of its formation. *Nature* 266, 346-348.
- [16] Bhuiyan, M.A., Rahman, J.J., Dampare, S.B., Suzuki, S., (2011). Provenance, tectonics and source weathering of modern fluvial sediments of the Baahmaputra-jamuna River, Bangladesh: inference from geochemistry. *Journal of Geochemical Exploration* 111, 113-137.
- [17] Das, B.K., Al-Mikhlafi, A.S., Kaur, P., 2006). Geochemistry of Mansar lake sediments Jammu, India: implication for source-area weathering, provenance and tectonic setting. *Journal Asian Earth Sciences* 26, 640-668.
- [18] Marques, R., Prudencio, L.M., Rocha, F., (2011). Patterns of rare earth and other trace elements in the different size fractions of clays of Campanian-Maastrichtian deposits from the Portuguese western margin (Aveiro and Taveiro formations). *Chemie Erde-Geochemistry* 71, 337-347.
- [19] Eslinger, E., Peaver. (1988). Clay minerals for petroleum geologists and engineers. Society of Economic Paleontologists and Mineralogists, 22, 428p.
- [20] Guggenheim S., Martin R.T., (1995). Definition of clay and clay mineral: joint report of the AIP and (MS) nomenclature committees. *Clay Minerals* 30, 257-259.
- [21] Velde, B., (1995). Origin and mineralogy of clays. *Clays and the Environment. Geological Magazine*, 334 p.
- [22] Nesbitt, H.W., Young, G.M., (1982). Early Proterozoic climates and plate motions inferred from major element chemistry of lutites. *Nature* 299, 715-717.
- [23] Nesbitt, H.W., Fedo, C.M., Young, G.M., (1997). Quartz and feldspar stability, steady and nonsteady-state weathering, and petrogenesis of siliciclastic sands and muds. *J. Geol.* 105, 173-191.]
- [24] McLennan, S.M., 1989. Rare earth elements in sedimentary rocks: influence of provenance and sedimentary processes. In *Geochemistry and mineral of rare earth elements* (ed. B. R. Pin and G. A. McKay). *Mineral Society of America* 21, 169-200.