Phytochemical Profile and Efficacy of Free Radical Scavenging Activity of *Alangium salvifolium* (l.f.) Wangerin.

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Abstract: Alangium salvifolium (l.f.) wangerin is an important medicinal plant belonging to the family Alangiaceae commonly known as Alangi in tamil was distributed in South India. Alangium species used as natural medicine, prescribed by Ayurveda, Siddha medical practitioners for various diseases due to its wide range of biological profile. The leaves of A. salvifolium are used as astringent, laxative, refrigerant and used to treat rheumatism, leprosy, syphilis and asthma. In the present study qualitative, quantitative analysis and antioxidant studies were undertaken.

Keywords: Alangium salvifolium, Ayurveda, Siddha, astringent, laxative, refrigerant, antioxidant studies, rheumatism, leprosy, syphilis and asthma.

1. Introduction

The plant Alangium salviifolium belongs to the family Alangiaceae is an important medicinal plant used for the treatment of various diseases. It is used in both codified and non codified system of medicines. All parts of the Alangium salviifolium plant are medicinally very important. Fruits are sweet, cooling and purgative and used as poultice for treating burning sensation and hemorrhage. Drug development is a complex process, and only companies with a consequent investment in research and development can afford to bring drugs for conception to market. Today, many new chemotherapeutic agents are obtained synthetically, based on "rational" drug design. The study of natural products has many advantages over synthetic drug design. The former leads to materials having new structural features with novel biological activity. In this context not only plants continue to serve as possible sources for new drugs, but chemicals derived from the various parts of these plants can also extremely used as lead structures for synthetic modification and optimization of bioactivity. The starting materials for about one half of the medicines we use today come from natural sources. Antioxidants are compounds that can delay or inhibit the oxidation of lipids or other molecules by inhibiting the initiation or propagation of oxidative chain reactions. There is no doubt that the future of plants as sources of medicinal agents for use in investigation, prevention, and treatment of diseases is very promising. Hence the present investigation reports phytochemical constituents and antioxidant properties of A.salviifolium.

2. Materials and Methodology

The plant specimens for the proposed study were collected from Bolluvampatty village, of Coimbatore District, Tamilnadu, India. Flowering shoots of the plant were also collected for identification. The collected plant material was identified and their authenticity was confirmed by the voucher specimen at the Botanical Survey of India (BSI), Southern circle, Coimbatore, Tamilnadu. The collected leaves of *Alangium salvifolium* plant were thoroughly washed in tap water and shade dried for about 15-30 days made into coarse powder with mixer grinder separately. The powder obtained was passed through 60 mesh sieve plate and then used for extraction and determination of the quality. Phytochemical analysis of the extracts of plant was carried out and their bioactive compounds ^[1-2] and antioxidant activity^[3-6] were carried with the extracts.

3. Result and Discussion

Alangium salvifolium is a deciduous, shrub or small tree, (Plate-I) branchlets appressed - tomentose, sometimes straggling, sometimes spinous. Bark grey, orange-yellow when young; wood olive-brown, hard and close-grained, scented. Leaves oblong- or elliptic-lanceolate, entire, attenuate or sub acute, base oblique - subacute, more or less 3-5 nerved at the base. White-scented flowers in irregular axillary cymes or clusters, buds about 75 in. long, tawnypubescent; Calyx tube cupular, five to ten lobes. Petals five to ten lobes, linearly oblong. Stamens 10-30; anthers linear. Ovary inferior, unilocular; Ovule one, pendulous; style simple. Berries globose, covered by persistant calyx, stigma capitate. Seeds solitary and ovoid. A. salvifolium is distributed in Circars, Deccan and Carnatic, in dry regions, in the plains and low hills, common on roadsides

The results of qualitative phytochemical analysis of leaves of A. salvifolium are presented in Table- 1. The qualitative presence phytochemical study shows the of carbohydrate, protein, aminoacid, fixed oil and fat, gums and mucilage, alkaloids, saponins, glycosides, flavonoids, terpenoids, phenols, tannins and steroids in leaves. Intensity of reaction is more in methanol when compared to water. The results of quantitative phytochemical constituents of the study materials are recorded (Table-2). The water extracts contains flavonoids, tannins phenol, alkaloids, saponins and glycosides 3.51%, 2.03%, 5.03%, 2.62%, 1.83% and 1.29% respectively. The methanol contains flavonoids, tannins phenol, alkaloids, saponins and glycosides saponins and glycosides 4.87%, 4.60%, 7.9%, 3.64%, 1.07% and 1.15% respectively.

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The antioxidant activity determination in linoleic acid emulsion by ferric thiocyanate method

The total antioxidant activity of methanolic extract of leaves of A. salvifolium and standard (Butylated hydroxy toluene) was determined by ferric thiocyanate method. Antioxidant activity increased steadily with increasing concentration. Methanolic extract exhibited effective antioxidant activity. The effects of extract on various concentrations are shown in Table- 3. The percentage inhibition of the methanol extract in the linoleic acid system was high when compared to the water extract (23.68% at 20µg/ml, 35.48% at 40µg/ml, 49.11% at 100µg/ml, 73.50 at 200µg/ml and 81.93% at 500µg/ml). Whereas BHT showed 32.28%, 43.27%, 65.23%, 82.80%, 97.44% at respective concentrations. The result showed that the activity of standard was more than the extract. The differences between concentrations of extract and control were statistically significant. The IC₅₀ value of water extract and methanol extract of leaf and standard are 93.7µg.ml, 98µg/ml and 56.6µg/ml respectively (Table-3).

Reducing power assay by Fe³⁺- Fe²⁺ transformation

The Table -4 depicts the reducing power of the leaf extract and standard (Butylated hydroxy toluene) using the potassium ferric cyanide reduction method. The reducing power of methanol was higher than water extract which is calculated as 17.3% at 20 μ g/ml, 30.26% at 40 μ g/ml, 40.12% at 100 μ g/ml, 52.37% at 200 μ g/ml and 63.35% at 500 μ g/ml. The standard (BHT) exhibited 17.19%, 28.5%, 40.79%, 44.52% and 54.19% at respective concentrations. The results exhibit increasing activity with increasing concentration. The results showed the activity of leaf extracts is more than that of standard (BHT) at higher concentration. IC₅₀ values of water extract of leaf, methanol and standard are 162 μ g/ml, 176 μ g/ml and 298 μ g/ml respectively.

Hydrogen peroxide scavenging activity

The ability of the methanolic extract of leaf and standard to scavenge H_2O_2 was determined and shown in Table- 5. The

water extract of leaf exhibited higher activity than the methanol extract 20.13%, 37.95%, 52.14%, 63.91% and 72.78% at 20µg/ml, 40µg/ml, 100µg/ml, 200 µg/ml, and 500µg/ml respectively. IC_{50} value of water extract, methanol and standard were estimated as 89µg/ml, 149µg/ml and 30.3µg/ml respectively. The standard Butylated hydroxy anisole (BHA) showed 42.04%, 61.02%, 72.27%, 81.11% and 89.18% at 20, 40, 100, 200 and 500µg/ml respectively. The standard exhibited stronger activity than both the extracts of leaf.

DPPH scavenging activity

The radical scavenging activity of the compounds can be measured as a decolourising effect following trapping of the unpaired electrons of 1,1 diphenyl-2-picryl hydrazyl (DPPH). illustrates the activity of extracts and standard (Butylated hydroxy anisole) against the DPPH radical. The methanolic extract of leaf exhibit higher inhibition than that of water extract 17.90%, 30.32%, 34.75%, 54.31%, 61.26% at 20, 40,100, 200 and 500µg/ml respectively. IC₅₀ values of water extract, methanol extract of leaf and standard were 188µg/ml, 193µg/ml and 60.6µg/ml respectively (Table- 6). The results indicate that the standard exhibited stronger radical scavenging activity against DPPH.

Hydroxyl radical scavenging activity

The hydroxyl radical scavenging activity of methanolic extract of leaf of *A. salvifolium* is presented in table 7. The inhibition of hydroxyl radical increase in a dose dependent manner. The hydroxyl radical scavenging activity of methanol extract showed more activity 16.04%, 21.09%, 38.59% and 48.08% at 20, 40, 100 and 500µg/ml respectively. The IC₅₀ values of water extract, methanol extract of leaf and standard are 196µg/ml, 200µg/ml and 354µg/ml respectively (Table- 7). The results indicate that the methanolic extract of leaf has more activity than the standard Butylated hydroxy anisole (BHA).

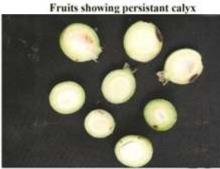


Habit

Twig with fruit

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Cut view of the fruits



Ripe fruits



Dry Fruits

The number of +	indicates	the	intensity	of	reaction	and
compound present.	- indicate	the	absence of	co	mpound.	

salvifolium							
Test Extract							
	water	methanol					
Carbohydrates							
Molisch test	+	++					
Proteins							
Biuret test	+	+					
Amino acids							
Ninhydrin test	+	+					
Fixed oils and fats	+	+					
Spot test							
Gums and Mucilages	+	+					
Alkaloids							
Dragendroff's	+	++					
Wagner's	+	++					
Mayer's	+	++					
Saponins	+						
Common test							
Glycosides	+	++					
Keller-killani test							
Flavonoids	+	++					
Shinoda test							
Terpenoids	++	++					
Salkowski test							
Phenols	++	+++					
Lead acetate							
Tannins	+	++					
Ferric chloride		+					
Lead acetate							
Steroids	++	++					
Libermann-Burchard's test							

Table 1: Phytochemical screening of leaves of A.

Table 2: Quantitative phytochemical estimation of leaves of

 A. salvifolium

A. saivijoitum							
Phytochemical	Water extract (mg/g)	Methanol extract					
Constituents		(mg/g)					
Flavonoids	3.51±0.03	4.87±0.07					
Tannins	2.03±0.05	4.60±0.02					
Total phenols	5.03±0.05	7.9±0.11					
Alkaloids	2.62±0.04	3.64±1.01					
Saponins	1.83±0.03	1.07±0.06					
Glycosides	1.29 ± 0.02	1.15 ± 0.008					

Values are mean of triplicates \pm SD

Table 3: The antioxidant activity of leaf of A. salvifolium in linoleic acid emulsion by ferric thiocyanate method

Concentration		% of inhibition	n		IC_{50}		
(µg/ml)				Standard (BHT)	Water extract	Methanol extract	
	Standard (BHT)	Water extract	Methanol extract				
20	$32.28 \pm 0.33e$	$21.36 \pm 0.06e$	23.68± 0.07e				
40	43.27 ± 0.21 d	$30.41 \pm 0.04d$	$35.48 \pm 0.03 d$				
100	$65.23 \pm 0.21c$	$43.68 \pm 0.02c$	49.11±0.09c	56.6 µg/ml	93.7µg/ml	98µg/ml	
200	$82.80{\pm}0.14b$	$67.90 \pm 0.03b$	$73.50 \pm 0.02b$				
500	97.44±0.31a	$83.33 \pm 0.04a$	81.93±0.06a				

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BHT- Butylated hydroxy toluene; Values are mean of triplicates \pm SD

1	able 4: Total leu	uctive capacity	<i>onum</i> by re - 1	re transforma			
Concentration		% of inhibition			IC_{50}		
(µg/ml)				Standard (BHT)	Water extract	Methanol extract	
	Standard (BHT)	Water extract	Methanol extract				
20	$16.19 \pm 0.14e$	$15.08\pm0.07e$	17.3±0.29e		162µg/ml	176µg/ml	
40	$28.5 \pm 0.09 d$	25.13±0.51d	30.26±0.29d	298µg/ml			
100	$40.79 \pm 0.08c$	$37.78 \pm 0.27c$	40.12±0.15c				
200	$44.52 \pm 0.04 b$	$51.27 \pm 0.44b$	52.37±0.48b				
500	$63.83\pm0.38a$	63.35±0.48a	54.19±0.14a				
	Concentration (µg/ml) 20 40 100 200	$\begin{array}{c} \text{Concentration} \\ (\mu g/ml) \\ \hline \\ & \\ \hline \\ 20 \\ 16.19 \pm 0.14e \\ \hline \\ 40 \\ 28.5 \pm 0.09d \\ \hline \\ 100 \\ 40.79 \pm 0.08c \\ \hline \\ 200 \\ \hline \\ 44.52 \pm 0.04b \\ \hline \end{array}$	$\begin{array}{c} \mbox{Concentration} & & & & & & & & & & & & & & & & & & &$	$\begin{array}{c c c c c c c c c c c c c c c c c c c $	$ \begin{array}{c c c c c c c c c c c c c c c c c c c $	$ \begin{array}{c c c c c c c c c c c c c c c c c c c $	

Table 4: Total reductive capacity of leaf of *A. salvifolium* by Fe³⁺ - Fe²⁺ transformation

BHT- Butylated hydroxy toluene ;Values are mean of triplicates \pm SD

Table 5: Hydrogen peroxide scavenging activity of leaf of A. salvifolium

Concentration (µg/ml)	% of inhibition				IC ₅₀	
				Standard (BHA)	Water extract	Methanol extract
	Standard (BHA)	Water extract	Methanol extract			
20	$42.04 \pm 0.04e$	$20.13 \pm 0.43e$	$1826 \pm 0.74e$		89µg/ml	149µg/ml
40	$61.02 \pm 0.02 d$	37.95±0.58d	30.03±0.19d			
100	72.27± 0.21c	52.14±0.35c	43.94±0.75c	30.3µg/ml		
200	$81.11 \pm 0.08b$	63.91±0.27b	52.75±0.09b			
500	89.18± 0.13a	72.78±0.41a	74.29±0.58a			

BHA -Butylated hydroxy anisole ;Values are mean of triplicates \pm SD

Table 6: DPPH scavenging activity of leaf of A. salvifolium

Concentration (µg/ml)	% of inhibition				IC_{50}	
				Standard (BHA)	Water extract	Methanol extract
	Standard (BHA)	Water extract	Methanol extract			
20	$21.62 \pm 0.07e$	11.97±0.24e	17.90±0.19e	60.6µg/ml	188µg/ml	193µg/ml
40	42.06 ± 0.04 d	20.36±0.50bd	30.32±0.46d			
100	$66.37 \pm 0.03c$	31.99±0.17c	34.75±0.54c			
200	82.58± 0.03a	51.12±0.16b	54.31±0.09b			
500	89.5± 0.02b	56.55±0.26a	61.26±0.16a			

BHA- Butylated hydroxy anisole; Values are mean of triplicates \pm SD

Concentration		% of inhibition			IC_{50}				
$(\mu g/ml)$				Standard (BHA)	Water extract	Methanol extract			
	Standard (BHA)	Water extract	Methanol extract						
20	11.40±0.01e 17.07±0.16d		16.04±0.21e						
40	$20.14 \pm 0.10d$	20.14±0.10d 18.99±0.31d 21.09±0.15d	21.09±0.15d	354µg/ml	196µg/ml				
100	31.38± 0.02c	35.95±0.18c	38.59±0.47c			200µg/ml			
200	43.6± 0.07 b	52.61±0.07b	48.37±0.29b						
500	64.19±0.14a	61.11±0.23a	63.08±0.26a						

Table 7: Hydroxyl radical scavenging activity of leaf of A. salvifolium

BHA - Butylated hydroxy anisole; Values are mean of triplicates \pm SD

4. Conclusion

Phytochemicals were more in methanol extract. So the methanol extract was well focused for other phytochemical analysis. Quantification studies helped to quantify the polyphenols (phenols, flavonoids, tannins), alkaloids, saponins and glycosides. Based on the traditional use of this plant and the presence of these chemicals evident for the medicinal properties of the leaf. The methanolic extract of leaf analysed for their free radical scavenging activity using various methods which include DPPH scavenging activity, hydroxyl radical scavenging activity, hydrogen peroxide activity, reducing power assay and total antioxidant activity. The extract showed maximum reducing activity and hydroxyl radical scavenging activity. Antioxidants are compounds that inhibit or delay the oxidation of other molecules by inhibiting the initiation or propagation of oxidizing chain reactions. Synthetic antioxidants are suspected of being responsible for liver damage and carcinogenesis. From this study, it can be concluded that the species is effective in scavenging free radicals and has the potential to be a powerful antioxidant.

5. Acknowledgement

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