

Phytoremediating Potentials of *Sida acuta* and *Duranta erecta* for Lead, Cadmium, Cobalt and Zinc

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Abstract: Recently pollution of the environment has gathered an increased global interest. In this respect, contamination of soils with heavy metals has always been considered a critical challenge in the scientific community. Remediation of soil contaminated by heavy metals is necessary in order to reduce the associated risks, make the land resource available for agricultural production, enhance food security, and scale down land tenure problems. Immobilization, soil washing, and phytoremediation are frequently listed among the best available technologies for cleaning up heavy metal contaminated soils but have been mostly demonstrated in developed countries. Non-edible African plants –*Sida acuta* and *Duranta erecta* were used to study the absorption of Cadmium, Lead Zinc and Cobalt from soils inoculated with the metal ions. 0.1M, 0.5M and 1M solutions of the metal ions were used in the inoculation. The Leaves, stems and roots of *Sida acuta* and the stems and roots of *Duranta erecta* were collected in the first instance at six weeks, and then, at ten weeks of planting. Atomic Absorption Spectrophotometer was used to determine the metal ion concentration in the plants' parts. Lead was more absorbed by *Sida acuta* than did *Duranta erecta*, with the highest absorption of 1.223mg/kg in the former occurring in the roots. Absorption increased as the concentration of the inoculant solution increased, and also on moving from 6 weeks' to 10 weeks' samples for concentrations less than 0.1M. Cadmium was the only absorbed by *Sida acuta*, with a highest value of 6.495mg/kg in the roots., *Duranta erecta* was poisoned by Cd^{2+} in all concentrations. Zinc was more absorbed by *Duranta erecta* than did *Sida acuta*, with the highest absorption of 7.898mg/kg in the former occurring in stems. Absorption increased as the concentration of the inoculants solution increased. Cobalt was most absorbed by *Sida acuta* with the highest value of 9.354mg/kg found in the stem. Generally, Phytotoxicity was shown in the plants at inoculants concentration above 0.5M, after 6 weeks except for *Duranta erecta* inoculated with Zn^{2+} . The tolerance for Lead, Cadmium, and Cobalt by *Sida acuta* show a good promise for its phytoremediation and recovery.

Keywords: Phytoremediation ;*Sida acuta*; *Duranta erecta*; Phytotoxicity; Lead; Cadmium; Cobalt, Zinc

On behalf of all authors, the corresponding author states that there if no conflict of interest

1. Introduction

One of the greatest problems that the world is facing today is that of environmental pollution, increasing with every passing year and causing grave and irreparable damage to the earth(Meagher, 2000).

Industrialization is considered vital to the nation's socio-economic development as well as its standing in the international community. Ideally, the siting of industries should achieve a balance between socio- economic and environment considerations(Mohammedand Folorunsho 2015).

In recent years, heavy metal contamination has become a serious problem all over the world as these metals persist in the soil for longer period due to their non biodegradability(Kavitha et al, 2013). Since the beginning of the industrial revolution, soil pollution by these toxic metals has accelerated dramatically and has contributed to a variety of toxic effects on living organisms. Soils as the major sink have been contaminated by heavy metals and metalloids through emissions from the rapidly expanding industrial areas, sewage sludge, pesticides, wastewater irrigation, coal combustion residues, spillage of petrochemicals, and atmospheric deposition (Raymond et al, 2011, Aremu et al, 2010). Heavy metal pollution of the soil is caused by various metals, especially Cu, Ni, Cd, Zn Cr and Pb(Singh and Kalamdhad, 2011).

Heavy metal contamination of soil may pose risks and hazards to humans and the ecosystem through: direct ingestion or contact with contaminated soil, the food chain (soil-plant-human or soil-plant-animal-human)(McLaughlin et al, 2000). Lead is a highly toxic metal whose widespread use has caused extensive environmental contamination and health problems in many parts of the world(Monisha et al, 2014). The decontamination of soil and wastes polluted with anthropogenic chemical is a global problem that has consumed considerable economic resources(Chambers et al, 1991, Watts, 1997, Baker et al, 1994).

Presently, phytoremediation has become an effective and affordable technological solution used to extract and remove heavy metal pollutants from polluted soil. Many species of plants have been successful in absorbing heavy metal pollutants such as lead, cadmium and others from the soil and water(Tangahu at al, 2011, Foluso et al, 2009, Amin et al, 2013, Otaru et al, 2013, Kavitha and Jegadeesan, 2014). The health risks posed by these metals have continued to be of global concern, and have made the European Union to place thirteen metals on the High-Risk-Monitor level. These include Arsenic, Cadmium, Cobalt, Chromium, Copper, Mercury, Manganese, Nickel, Lead, Tin, and Thallium.

Lead, Cadmium, Zinc and Cobalt have been used in this work to study the ability of two nonedible African plants – *Sida acuta* and *Duranta erecta* – to phytoremediate soils polluted with the metal ions in their +2 oxidation states.

2. Methods

Twenty five seedlings, each, of the two plants were grown on soils isolated in polyethylene pots. Forty eight pots were inoculated with 0.1M, 0.5M and 1.0M solutions of Pb^{2+} , Zn^{2+} , Co^{2+} and Cd^{2+} , while controls were left. The plants' parts were harvested after the sixth and eighth week of inoculation. The harvested plants were washed, dried, and ashed at 450°C. After digesting with concentrated HNO_3 , Varian AA240 spectrophotometer was used to determine the metal ions concentrations absorbed in the plants' parts.

3. Results

Table 1: Concentration of Pb^{2+} absorbed by *Sida acuta* in mg/kg

Stem (Ss)	1st Harvest	2 nd Harvest
Ss ₁	0.397	0.496
Ss ₂	0.489	0.634
Ss ₃	1.108	Died

Root(SR)	1st Harvest	2 nd Harvest
SR ₁	0.417	0.523
SR ₂	0.563	0.985
SR ₃	1.223	Died

Leaf(SL)	1st Harvest	2 nd Harvest
SL ₁	0.358	0.405
SL ₂	0.425	0.597
SL ₃	0.713	Died

Table 2: Concentration of Pb^{2+} absorbed by *Duranta erecta* in mg/kg

STEM(Ds)	1st Harvest	2 nd Harvest
Ds ₁	0.251	Died
Ds ₂	2.400	Died
Ds ₃	Died	Died

Root(ER)	1st Harvest	2 nd Harvest
DR ₁	0.197	Died
DR ₂	1.658	Died
DR ₃	Died	Died

Table 3: Concentration of Cd^{2+} absorbed by *Sida acuta* in mg/kg

Stem (DS)	1st Harvest	2 nd Harvest
Ss ₄	1.423	2.004
Ss ₅	4.001	Died
Ss ₆	5.241	Died

Root(DR)	1st Harvest	2 nd Harvest
SR ₄	1.859	2.326
SR ₅	4.510	Died
SR ₆	6.495	Died

Leaf(DL)	1st Harvest	2 nd Harvest
SL ₄	0.921	1.675
SL ₅	3.577	Died
SL ₆	4.932	Died

Table 4: Concentration of Cd^{2+} absorbed by *Duranta erecta* in mg/kg

STEM(Ds)	1st Harvest	2 nd Harvest
Ds ₄	Died	Died
Ds ₅	Died	Died
Ds ₆	Died	Died

Root(DR)	1st Harvest	2 nd Harvest
DR ₄	Died	Died
DR ₅	Died	Died
DR ₆	Died	Died

Table 5: Concentration of Co^{2+} absorbed by *Sida acuta* in mg/kg

Stem (DS)	1st Harvest	2 nd Harvest
Ss ₇	7.421	Died
Ss ₈	8.252	Died
Ss ₉	9.354	Died

Root(DR)	1st Harvest	2 nd Harvest
SR ₇	6.089	Died
SR ₈	6.595	Died
SR ₉	8.074	Died

Leaf(DL)	1st Harvest	2 nd Harvest
SL ₇	6.853	Died
SL ₈	7.532	Died
SL ₉	8.901	Died

Table 6: Concentration of Co^{2+} absorbed by *Duranta erecta* in mg/kg

STEM(Ds)	1st Harvest	2 nd Harvest
Ds ₇	3.689	Died
Ds ₈	4.725	Died
Ds ₉	Died	Died

Root(DR)	1st Harvest	2 nd Harvest
DR ₇	2.142	Died
DR ₈	3.986	Died
DR ₉	Died	Died

Table 7: Concentration of Zn^{2+} absorbed by *Sida acuta* in mg/kg

Stem (Ss)	1st Harvest	2 nd Harvest
Ss ₁₀	2.347	2.418
Ss ₁₁	2.515	2.663
Ss ₁₂	4.441	Died

Root(SR)	1st Harvest	2 nd Harvest
SR ₁₀	2.409	2.434
SR ₁₁	2.557	2.731
SR ₁₂	4.520	Died

Leaf(DL)	1st Harvest	2 nd Harvest
SL ₁₀	1.857	2.001
SL ₁₁	2.409	2.605
SL ₁₂	3.713	Died

Table 8: Concentration of Zn^{2+} absorbed by *Duranta erecta* in mg/kg

STEM(Ds)	1st Harvest	2 nd Harvest
Ds ₀	3.252	3.970
Ds ₁₁	3.754	5.078
Ds ₁₂	7.571	7.898

Root(DR)	1st Harvest	2 nd Harvest
DR ₁₀	2.573	2.834
DR ₁₁	2.975	3.422
DR ₁₂	3.721	3.115

CODES: S- *Sida acuta*, D-*Duranta erecta*, s- stem, R- root, L- leaf, 1,4,7,10= 0.1M, 2,5,8,11= 0.5M, 3,6,9,12= 1.0M

4. Discussions

The result of the analysis generally showed that some African plants could be used to phytoremediate soil polluted with Pb²⁺, Cd²⁺, Co²⁺ and Zn²⁺. Zinc was most absorbed by both plants. Lead was more absorbed by *Sida acuta* than did *Duranta erecta*, with the highest absorption of 1.223mg/kg in the former occurring in the roots. Absorption increased as the concentration of the inoculant solution increased, and also on moving from 6 weeks' to 10 weeks' samples for concentrations less than 0.1M. Phytotoxicity was shown in the plants at inoculants concentration above 0.5M, after 6 weeks for *Sida acuta*, and from time of planting to 10 weeks for *Duranta erecta*. Cadmium was the only absorbed by *Sida acuta*, with a highest value of 6.495mg/kg in the roots, *Sida acuta* could not tolerate Cd²⁺ as concentration of the inoculant solution increased, and also on moving from 6 weeks' to 10 weeks'. *Duranta erecta* was poisoned by Cd²⁺ in all concentrations. Zinc was more absorbed by *Duranta erecta* than did *Sida acuta*, with the highest absorption of 7.898mg/kg in the former occurring in stems. Absorption increased as the concentration of the inoculants solution increased. Cobalt was most absorbed by *Sida acuta* with the highest value of 9.354mg/kg found in the stem. Generally, Phytotoxicity was shown in both plants at inoculants concentration above 0.5M, after 6 weeks except for *Duranta erecta* inoculated with Zn²⁺. The tolerance for Lead, Cadmium, and Cobalt by *Sida acuta* shows a good promise for its phytoremediation and recovery.

5. Conclusion

Soil is the fundamental foundation of our agricultural resources, food security, global economy and environmental Quality. The need to decontaminate the soils of heavy metals cannot be overemphasized. *Sida acuta* and *Duranta erecta* were used. It can be safely concluded that *sida acuta* has better phytoremediating potential for Pb²⁺, Cd²⁺ and Co²⁺ than did *Duranta erecta*. *Duranta erecta* which cannot be used to phytoremediate Cd²⁺ had better phytoremediating potential for Zn²⁺ than did *Sida acuta*.

References

- [1] Amin M., Hamidi A. A., Mohammad A. Z., Shuokr Q. A., M. Razip B. S.(2013),
- [2] Phytoremediation of Heavy Metals from Urban Waste Leachate by Southern Cattail (*Typha domingensis*). *International Journal of Scientific Research in Environmental Sciences (IJSRES)*, 1(4):63-70,
- [3] Aremu, M.O., Atolaiye B.O. and Labaran, L.(2010), Environmental Implication of metal concentrations in soil, plant foods and pond in area around the Derelict Udege Mines of Nasarawa State, Nigeria. *Bull. Chem. Soc. Ethiop.* 24(3): 351-360

- [4] Baker, A.J.M. McGrath, S.P. Sidoli, C.M.D. Reeves, R.D.(1994), The possibility of in situ heavy metal decontamination of polluted soils using crops of metal-accumulating plants. *Elsevier*, 11(1-4):41-49.
- [5] Chambers, C.D., Willis, J., Giti-Pour, S., Zieleniewski, J.L., Rickabough, J.F., Mecca, M.I., Pasin, B., Sims, R.C., Sorensen, D.L., Sims, J.L., McLeam, J.E., Mahmood, R.R., Wagner, K.,(1991), In situ treatment of Hazardous Waste- contaminated Soils, 2nd edn, Noyes Data Corporation, Park Ridge, New York.
- [6] Foluso O.A., Bamidele I.O., Kayode O.A.,(2009), Phytoremediation potential of *Eichornia crassipes* in metal-contaminated coastal water. *Elsevier bioresource technology*, 100(19):4521-4526
- [7] Kavitha, B., Jothimani, P., Ponmani, S., Sangeetha, R., (2013). Phytoremediation of Heavy metals- A review, *International Journal of Research studies in Bioscience*, Coimbatore, 1:17-23
- [8] Kavitha, K. K. and Jegadeesan, M.(2014), Mercury and cadmium accumulation in selected weed plants: Implications for phytoremediation. *Asian Journal of Plant Science and Research*, 4(5):1-4
- [9] Meagher, R.B.,(2000), Phytoremediation of toxic elemental and organic pollutants, current opinion in plan Biology, 3(2): 153-162
- [10] McLaughlin, M. J. Zarcinas, B. A. Stevens, D. P. and Cook, N.(2000) "Soil testing for heavy metals," *Communications in Soil Science and Plant Analysis*, 31(11): 1661-1700,
- [11] Mohammed, S. A.1 and Folorunsho, J.O.(2015), Heavy metals concentration in soil and *Amaranthus retroflexus* grown on irrigated farmlands in the Makera Area, Kaduna, Nigeria. *Journal of Geography and Regional Planning*, 8(8): 210-217
- [12] Monisha J., Tenzin T., Naresh A., Blessy B. M., and Krishnamurthy N. B.(2014), Toxicity, mechanism and health effects of some heavy metals. *Interdisciplinary Toxicology*. 7(2): 60-72.
- [13] Otaru, A.J., Ameh, C.U., Okafor, J.O., Odigure, J.O., Abdulkareem, A.S and Ibrahim, S.(2013), Study on the Effectiveness of Phytoremediation in the Removal of Heavy Metals from Soil Using Corn, *International Journal of Computational Engineering Research*03(4): 87-93.
- [14] Raymond A. Wuana and Felix E. Okieimen(2011), "Heavy Metals in Contaminated Soils: A Review of Sources, Chemistry, Risks and Best Available Strategies for Remediation," *ISRN Ecology*, Article ID 402647, p 20. <https://doi.org/10.5402/2011/402647>.
- [15] Singh Jiwan and Kalamdhad Ajay S.(2011), Effects of Heavy Metals on Soil, Plants, Human Health and Aquatic Life. *International Journal of Research in Chemistry and Environment*, 1(2): 15-21
- [16] Tangahu, B.V., Abudulla, S.T.S., Basri, H., Anuar, N., and Mukhlisin, M.,(2011), A Review on heavy metal (As, Pb and Hg) uptake by plants through phytoremediation, *International Journal of Chemical Engineering, Malaysia*, ID 939161: 1-31
- [17] Watts R.J. (1997), Hazardous Wastes: Sources, Pathways and Receptors. John Wiley, New York.