

Reduction in Inorganic Loading Rate in Dairy Wastewater through Use of Nanomaterials: Zinc Oxide and Mixture of P₂₅ Degussa & Zinc Oxide

Dushyant Topiwala¹, Basavaraj Balapgol², Pramod Jadhav³

¹Student, D.Y. Patil College of Engineering, Akurdi, Pune (India)

²Principal, D.Y. Patil College of Engineering, Akurdi, Pune (India)

³Professor, KIT College of Engineering, Kolhapur (India)

Abstract: *The aim of this research is to carry out the reduction of Inorganic Loading Rate in Dairy wastewater by the use of nanomaterials: Zinc Oxide and mixture of P₂₅ Degussa & Zinc Oxide. All these reagents belonging to nanoscale gives higher surface area for adsorption mechanism with the photocatalysis process under the Ultraviolet lamp. By this research thesis we can find relative efficiency of treating inorganic waste in the wastewater.*

Keywords: Zinc Oxide, P₂₅ Degussa, Photocatalytic adsorption, Inorganic Loading Rate

1. Introduction

The wastewater growth rate is increasing day by day in the dairy industry with increase in the population demand for dairy products. The upcoming technology is nanotechnology. Making use of nanotechnology in environmental engineering field leads to a new prospect [1]. By using nanoscale powders for treatment of inorganic loading rate in dairy wastewater leads to higher efficiency for removal of inorganic loading waste [2]. Use of P₂₅ Degussa which is the mixture of nanoscale Titanium dioxide minerals Anatase & Rutile i.e. 70 to 80% and 30 to 20% respectively with Zinc Oxide (ZnO) and Zinc Oxide singly in nanoscale for the treatment of dairy wastewater [3]. These powders are insoluble in water solution and hence can be separated easily from the wastewater after the treatment [4]. Taking comparative study for ZnO and mixture of P₂₅ Degussa & ZnO, we can found effectiveness of both for treating inorganic loading rate in dairy wastewater.

2. Apparatus & Materials

- 2.1 Wastewater Sample – Wastewater source from Dairy Industry in Pune
- 2.2 Ultraviolet Light Source – Additional source of UV light for photocatalysis process with closed casing of wooden box so UV rays do not penetrate outside and come in direct contact with skin as UV rays are carcinogenic after extent of exposure
- 2.3 Zinc Oxide – ZnO in nanoscale powdered form
- 2.4 Mixture of ZnO and P₂₅ Degussa – ZnO and P₂₅ Degussa in nanoscale form mixed as 1:1 ratio form
- 2.5 Magnetic stirrer – Magnetic stirrer to provide adequate stirring in the solution so that the insoluble powder uniformly gets distributed throughout the solution

3. Methodology

Taking both the reagents for different concentrations of 0.3gms, 0.5gms, 0.7gms & 0.9gms per 500ml dairy wastewater and making solutions. These solutions are mixed thoroughly by using magnetic stirrer apparatus so the powder gets uniformly distributed in the solution. Taking this uniformly mixed solution and exposing them for 24hours under the UV box for the process to undergo photocatalytic adsorption reaction [5]. After the process is completed take the supernatant treated wastewater for inorganic loading rate test i.e. COD test.

4. Results

ZnO showing maximum effectiveness for 0.9gms/500ml solution i.e. Dairy wastewater with 828 mg/l inorganic loading rate reduced to 165.6mg/l and efficacy of 80% in 24hours of exposure under UV light, while mixture of ZnO and P₂₅ Degussa too showing maximum effectiveness for 0.9gms/500ml dosage which reduced to 184mg/l with efficacy of 77.78% in 24hours of UV light exposure.

5. Conclusion

With increase in concentration of reagent there is decrease in the inorganic loading rate in the dairy wastewater for both reagents. In comparative study of both reagents ZnO works with higher affectivity for reducing inorganic loading rate in dairy wastewater.

6. Acknowledgment

We acknowledge with thanks Prof. Sachin Mane, PG Coordinator, Environmental Engineering, D.Y. Patil College of Engineering, Akurdi, Pune-44, for his valuable guidance continuous encouragement and advice throughout my research work. We are also extremely thankful to Prof S.V. Pataskar, Head of Civil Engineering Department, D.Y. Patil

College of Engineering, Akurdi, Pune-44 for providing valuable suggestion and advice. We would like to thank all the staff members of the Civil Department for their prompt help and encouragement towards the fulfilment of our research. We are also thankful to KIT College of Engineering, Environmental Department for providing these valuable suggestions and advice. We are also thankful to Bubun Ceramics for providing the apparatus reagents. We wish to thank all those who have contributed and provided support either directly or indirectly to our research.

References

- [1] Xiaolei Qu, Pedro J.J. Alvarez, Qilin Li, 2012 *Applications of Nanotechnology in Water and Wastewater Treatment*, *Int. J. Science Direct.* 47 3931 – 3946
- [2] Jatin G. Bhadiyadra1, Minakshi V. Vaghani, 2015 *A Review on Applicability of Photo-catalyst Titanium dioxide for Treatment of Grey water*, *Int. J. Engineering Research and Applications* 5 102 – 105
- [3] B. Ohtani, O. O. Prieto-Mahaney, D. Li and R. Abe, 2010 *What is Degussa (Evonik) P25? Crystalline Composition Analysis, Reconstruction From Isolate Pure Particles and Photocatalytic Activity Test*, *J. of Photochemistry and Photobiology A Chemistry* Volume 216, pp 179-182
- [4] Teruhisa Ohno, Koji Sarukawa, Kojiro Tokieda and Michio Matsumura, 2001 *Morphology of a TiO₂ Photocatalyst (Degussa, P-25) Consisting of Anatase and Rutile Crystalline Phases*, *J. of Catalysis* Volume 201 - 203, pp 82-86
- [5] Ruiqun Chen, Jie Han, Xiaodong Yan, Chongwen Zou, Jiming Bian, Ahmed Alyamani, Wei Gao, 2011 *Photocatalytic Activities of Wet Oxidation Synthesized ZnO and ZnO –TiO₂ Thick Porous Films*, *Int. J. Applied Nanosci* 1 pp 37 – 44
- [6] Ming Hua, Shujuan Zhang, Bingcai Pan*, Weiming Zhang, Lu Lv and Quanxing Zhang, 2011 *Heavy Metal Removal From Water/Wastewater by Nanosized Metal Oxides: A Review*, *Int. J. of Hazardous Materials* Volume 2012 pp 317-331
- [7] Xiaojia Wan, Ting Wang, Yamei Dong and Dannong He, 2004 *Development and Application of TiO₂ Nanoparticles Coupled with Silver Halide*, *Int. J. of Nanomaterials* Vol. 2014 pp 5
- [8] Deanna C. Hurum, Alexander G. Agrios and Kimberly A. Gray, 2003 *Explaining the Enhanced Photocatalytic Activity of Degussa P25 Mixed-Phase TiO₂ Using EPR*, *J. Phys. Chem. B* 2003 Volume 107, pp 4545-4549
- [9] Song'ul KARAASLAN AKSUI, Seref G'UCER, 2010 *Investigations on Solar Degradation of Acid Orange 7 (C.I. 15510) in Textile Wastewater with Micro and Nano Sized Titanium Dioxide*, *Turkish J. Eng. Env. Sci.* 34 275 – 279
- [10] Chitpisud Supha, Yuphada Boonto, Manee Jindakaraked, JirapatAnanpattarachai & Puangrat Kajitvichyanukul, 2015 *Long-term Exposure of Bacterial and Protozoan Communities to TiO₂ Nanoparticles in an Aerobic Sequencing Batch Reactor*, *Int. J. Science and Technology of Advanced Materials* 16 pp 12
- [11] Haithem Bel Hadjltaief, Abdesslem Omri, Mourad Ben Zina, Patrick Da Costa and Maria Elena Galvez, 2015 *Titanium Dioxide Supported on Different Porous Materials as Photo-catalyst for the Degradation of Methyl Green in Wastewaters*, *Int. J. Advances in Material Sci. and Eng. Volume* 2015 pp 10
- [12] Harikumar PS*, Litty Joseph and Dhanya A, 2013 *Photo-catalytic Degradation of Textile Dyes by Hydrogel Supported Titanium Dioxide Nano-particles*, *J. of Environmental Engg. And Ecological Sci. Article*
- [13] Dushyant Topiwala, Basavaraj Balapgol, 2017 *A Review On: Reduction of Inorganic Matter in Wastewater Through Use of Titanium Dioxide*, *International Conference on Academic Research in Engg. & Management. Volume* 2017 pp 151-153