

Synthesis and *In Vitro* Evaluation of Anticancer activity of Mannich Bases of Benzimidazole Derivatives

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Abstract: Benzimidazole moiety is an important class of heterocycles used for synthesis of medicinal compounds. Mannich bases of Benzimidazole were synthesized. The structures were identified by determining their melting points, TLC, IR spectral analysis and ¹HNMR spectrum. Anticancer activity was performed in vitro for screening of synthesised compounds by Sulfordamine B (SRB) assay. Cytotoxicity of compounds was evaluated for lung, prostate, colon and breast tissue. Compound VIII a and VIII b showed 100% cytotoxicity against colon cancer. The compound VI c showed 89% cytotoxicity against lung cancer. Against prostate cancer compound VIII a resulted in 73% cytotoxicity. The compound VIII a and VI a showed 94%, 93% cytotoxicity respectively against breast cancer.

Keywords: Mannich base, Heterocyclic, Benzimidazole, Anticancer activity, Cytotoxicity, Genotoxicity

1. Introduction

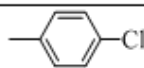
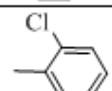
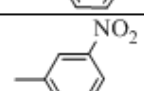
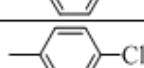
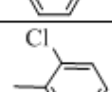
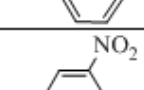
In recent years, cancer has become a major killer and challenge to the world chemists. It is a disease in which cells divide and grow uncontrollably, consuming energy and losing both structure and function due to an inability to adequately differentiate [1]. The most prevalent cancers are lung, breast, prostate, colon cancer. In spite of wide anti-cancer drugs available, it remains as a deadly disease. Anti-cancer drugs act either by killing the cancer cells or modifying their growth. But, such chemotherapy is often associated with cytotoxicity, genotoxicity to normal cells together with development of resistance [2]. Therefore; medicinal chemists are indulged in continuous research for screening of novel and safe anti-cancer agents.

Benzimidazole is a well-established potent anti-cancer molecule. The Imet 3393 is commercially available benzimidazole based anticancer drug [3]. Benzimidazole is a heterocyclic compound and important intermediate in organic reactions. In cancer therapy, resistance to wide range of unrelated drugs may occur after resistance to a single agent has developed. Multiple drug resistance is lack of expected therapeutic response to several disease-specific pharmaceutical agents [4]. Highly drug resistant tumour cells limit the success rate of cancer chemotherapy [5]. To combat this, novel compounds are needed to solve the problem of MDR therefore, compounds are formed by its modification through different chemical reactions to enhance its biological activity. In Pharmaceutical chemistry, Benzimidazole moieties are being developed as DNA minor groove binding agents that have significant anti-tumor activity [6]. Mannich Reaction is very useful for such modification of Benzimidazole molecule and it forms Mannich bases. Mannich reaction is the C-C bond forming condensation reaction of ammonia, primary or secondary amine, formaldehyde and compound containing at least one hydrogen atom of pronounced reactivity [7]. Mannich bases are versatile and reactive intermediates and, therefore, easily converted into other molecules. In addition, Mannich bases of Benzimidazole are known to display anti-cancer activity. Mannich bases, derivatives of the various heterocycles,

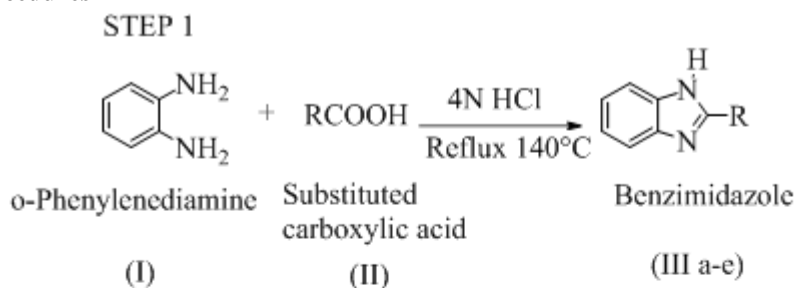
show antiproliferative activity in vitro against human tumor cell lines [8]. Based on these facts, it is worthwhile to synthesize mannich bases with enhanced anti-cancer activity.

2. Materials and Method

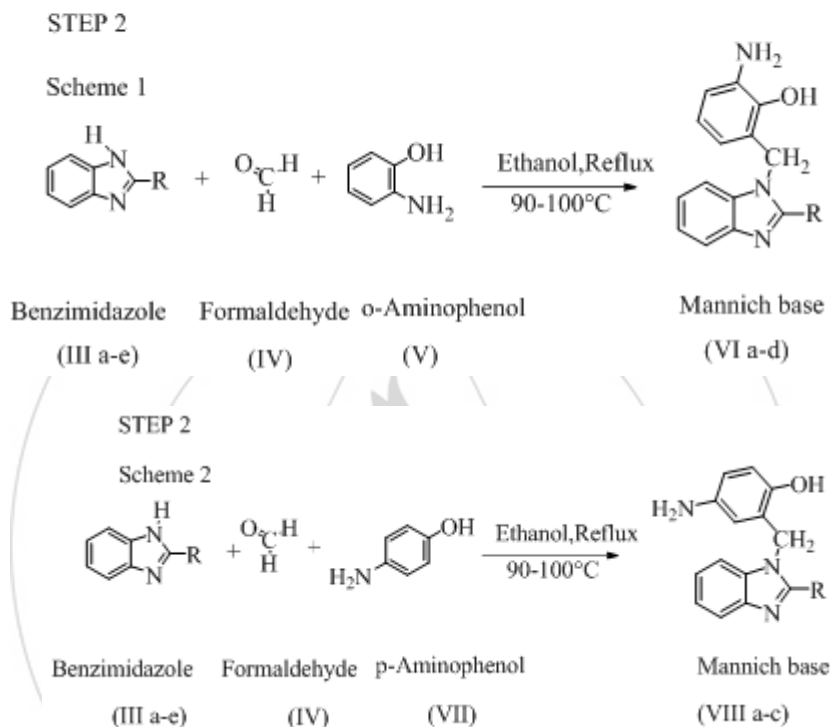
The chemicals to synthesise compound were procured from Merck, Lobachemie and were synthesised by using parallel synthesiser. The melting points of the synthesised compounds were determined by visual melting point apparatus. Reaction and purity of the compounds were analysed by use of thin layer chromatography (Chloroform: Methanol; 9:1). The IR spectra was recorded on ATR spectrophotometer. ¹HNMR spectra was recorded at 300MHz using DMSO as solvent and TMS as internal standard.

Compounds	R
VI a	H
VI b	
VI c	
VI d	
VIII a	
VIII b	
VIII c	

2.1 Synthetic Procedures



Substituted carboxylic acid : formic acid, benzoic acid, 2-chlorobenzoic acid, 4-chlorobenzoic acid, 3-nitrobenzoic acid



1. For synthesis of Benzimidazole[9]

In parallel synthesizer, ortho phenylenediamine (0.02 mole) was refluxed with formic acid, benzoic acid and its derivatives (0.02 mole) in the presence of 4N hydrochloric acid (4N HCl) for 8-9 hrs at temperature 140 °C, rpm 800. The completion of the reaction was checked by TLC (chloroform: methanol 9:1). On completion, 10% NaOH (w/v) was gradually added until the reaction mixture turns alkaline. To obtain precipitate, reaction mixture was cooled in ice bath and allowed to stand for 5 min. The product was filtered, dried and recrystallized from ethanol.

2. For synthesis of Mannich bases

Using ortho and para aminophenol as substrate (Scheme 1 and Scheme 2) 0.005 mole of o-aminophenol or p-aminophenol and 0.005 mole of formaldehyde (formalin) were added to the solution of substituted Benzimidazole (0.005 mole) in 10 ml of ethanol in 250ml round bottom flask. The mixture was then refluxed for 8-10 hour on parallel synthesizer at 90-100 °C. On cooling, reaction mixture was poured on crushed ice. The lump mass of precipitate was obtained which was then washed with ether, filtered and dried. The solid precipitate was isolated and dried in air.

2.2 Biological Activity

Anticancer activity[10]

The *in vitro* anticancer activity was carried out at CSIR Jammu. All synthesised compounds were tested for *In vitro* cytotoxicity against four different cancer cell lines including Lung (A549), Prostate (PC-3), Colon (HCT-116) and Breast (MCF-7) cell lines. The compounds were tested at 100µM concentration.

Sulforhodamine B (SRB) assay was performed to determine *In vitro* cytotoxicity. Four human cell lines of various tissue origin were used to evaluate the cytotoxic activity of compounds. In this method, cell suspension was seeded into 96 well flat bottom plates and incubated for 24 h. Test compounds at 100µM were added after 24h incubation. Further, after 48 h incubation, cells were fixed with ice cold TCA for 1 h at 4 °C. After 1h, the plates were washed five times with distilled water and allowed to air dry followed by addition of 100µl of 0.4% SRB dye for 0.5 h at room temperature. Plates were then washed with 1% v/v acetic acid to remove the unbound SRB dye. The bound dye was solubilised by adding 100µl of 10mM Tris buffer to each

well. Then the plates were put on shaker for 5 min to solubilize the dye completely. Finally the reading was taken at 540nm on microplate reader. The cytotoxic screening data of compounds at 100µM concentration is provided in table no. 2

3. Results and Discussion

3.1 Chemistry

The benzimidazole derivatives in above scheme were synthesised by the methods reported in the earlier literature [11]. The physical characterization of Mannich base of Benzimidazole derivatives are presented in table no. 1

Table 1: Physical characterization of Mannich base of Benzimidazole derivatives

S. No.	Compound	Molecular formula	Rf	%yield	Melting Point(°C)
1	VI a	C ₁₄ H ₁₃ N ₃ O	0.774	54%	185-188
2	VI b	C ₂₁ H ₁₇ ClN ₃ O	0.6	68%	235-239
3	VI c	C ₂₀ H ₁₆ ClN ₃ O	0.9	75%	275. 5-278
4	VI d	C ₂₀ H ₁₆ N ₄ O ₃	0.791	55%	213-215
5	VIII a	C ₂₀ H ₁₆ ClN ₃ O	0.75	72%	240-245
6	VIII b	C ₂₀ H ₁₆ ClN ₃ O	0.821	69%	274. 9-277
7	VIII c	C ₂₀ H ₁₆ N ₄ O ₃	0.76	42%	210-211

Spectral Analysis OF Synthesised Compounds

- 2-amino-6-(1H-benzimidazole-1-yl)methylphenol (VI a)
 ATR : OH(3539. 06 cm⁻¹), NH₂(3401. 01 cm⁻¹), CH₂(2920.50, 2853. 04 cm⁻¹), Ar C=C (1600.11 cm⁻¹), CH₂ bend(1475. 24 cm⁻¹), ortho substitution (747. 75 cm⁻¹), ¹HNMR(300MHz, DMSO):δ8. 43(s, 1H, Ar-H), 7. 70(ddd, 2H, Ar-H), 7. 32(td, 1H, Ar-H), 7. 20(td, 1H, Ar-H), 6. 74(ddt, 1H, Ar-H), 6. 68(t, 1H, ArH), 6. 43(dd, 1H), 5. 93(s, 1H, OH), 5. 34(d, 2H, CH₂), 3. 18(s, 2H, NH₂)
- 2-amino-6-([2-(4-chlorophenyl)-1H-benzimidazole-1-yl]methyl) phenol (VI b)
 ATR : OH(3531. 09 cm⁻¹), NH₂(3413. 21 cm⁻¹), CH₂(2889. 38cm⁻¹), Ar C=C (1603. 44, cm⁻¹), CH₂ bend(1444. 36 cm⁻¹), ortho substitution (750.34 cm⁻¹), R-Cl(648. 32 cm⁻¹), ¹H NMR(300 MHz, DMSO) : δ 7. 84-7. 76(m, 1H, Ar-H), 7. 59-7. 53(m, 1H, Ar-H), 7. 52-7. 42(m, 2H, Ar-H), 7. 35 7. 28(m, 1H, ArH), 6. 86(ddt, 1H, ArH), 6. 72(t, 1H, ArH), 6. 51(dd, 1H, ArH), 5. 93(s, 1H, OH), 5. 34(d, 1H, CH₂)
- 2-amino-6-([2-(2-chlorophenyl)1H-benzimidazole-1-yl]methyl) phenol(VI c)
 ATR : OH(3505. 22 cm⁻¹), NH₂(3443. 09 cm⁻¹), CH₂(2978. 21, 2884. 09cm⁻¹), Ar C=C (1610.06, cm⁻¹), CH₂ bend(1441. 28 cm⁻¹), ortho substitution (655. 47 cm⁻¹), ¹HNMR(300MHz, DMSO) : δ 7. 84 – 7. 78 (m, 1H, Ar-H), 7. 58 (td, 2H, Ar-H), 7. 52 – 7. 44 (m, 1H, Ar-H), 7. 41 – 7. 32 (m, 2H, Ar-H), 7. 35 – 7. 27 (m, 2H, Ar-H), 6. 85 (ddt, 1H, Ar-H), 6. 70 (t, 1H, Ar-H), 6. 44 (dd, 1H, Ar-H), 5. 93 (s, 1H, OH), 5. 34 (d, 2H, CH₂), 3. 17 (s, 2H, NH₂).
- 2-amino-6-([2-(3-nitrophenyl)-1H-benzimidazole-1-yl]methyl)phenol (VI d)
 ATR : OH(3544. 74 cm⁻¹), NH₂(3380.33 cm⁻¹), CH₂(2991. 06, 2718. 76cm⁻¹), Ar C=C (1603. 00 cm⁻¹),

CH₂ bend(1475. 96 cm⁻¹), ortho substitution (753. 89 cm⁻¹), R-NO₂(1362. 42 cm⁻¹), ¹HNMR(300MHz, DMSO):δ8. 52(t, 1H, Ar-H), 8. 28(ddt, 2H, Ar-H), 7. 84-7. 76(m, 1H, ArH), 7. 62(t, 1H, Ar-H), 7. 52-7. 44(m, 1H, Ar-H), 7. 36-7. 27(m, 2H, Ar-H), 6. 63-6. 54(m, 2H, Ar-H), 6. 28(dd, 1H, Ar-H), 5. 93(s, 1H, OH), 5. 34(d, 2H, CH₂), 3. 89(s, 2H, NH₂)

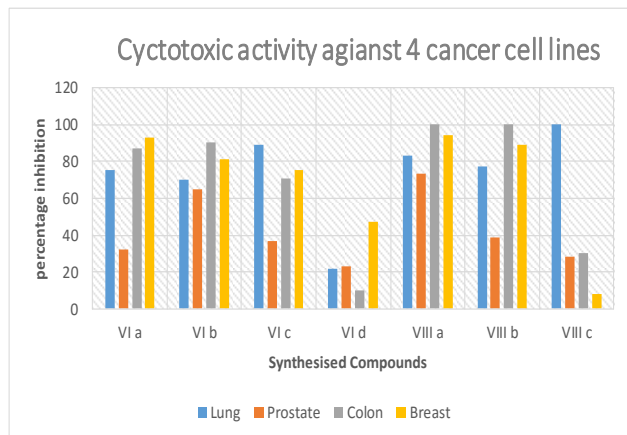
- 4-amino-2-([2-(4-chlorophenyl)-1H-benzimidazole-1-yl]methyl) phenol(VIII a)
 ATR: OH(3534. 13 cm⁻¹), NH₂(3415. 29 cm⁻¹), CH₂(2978. 04, 2887. 34cm⁻¹), Ar C=C (1610.16 cm⁻¹), CH₂ bend(1452. 80 cm⁻¹), para substitution (831. 36cm⁻¹), ¹HNMR (300MHz, DMSO) : δ 7. 84 – 7. 76 (m, 1H, Ar-H), 7. 58 – 7. 53 (m, 1H, Ar-H), 7. 52 – 7. 42 (m, 2H, Ar-H), 7. 35 – 7. 28 (m, 1H, Ar-H), 6. 68 (dt, 1H, Ar-H), 6. 47 – 6. 36 (m, 1H, Ar-H), 6. 10 (s, 1H, OH), 5. 34 (d, 2H, CH₂), 3. 94 (s, 2H, NH₂).
- 4-amino-2-([2-(2-chlorophenyl)-1H-benzimidazole-1-yl]methyl) phenol(VIII b)
 ATR : OH(3555. 01 cm⁻¹), NH₂(3441. 05 cm⁻¹), CH₂(2977. 81, 2886. 41cm⁻¹), Ar C=C (1681. 41 cm⁻¹), CH₂ bend(1460.91 cm⁻¹), para substitution (833. 37cm⁻¹), ¹HNMR (300 MHz, DMSO) :δ 7. 84 – 7. 76 (m, 1H, Ar-H), 7. 58 (ddd, 2H, Ar-H), 7. 52 – 7. 44 (m, 1H, Ar-H), 7. 41 – 7. 32 (m, 2H, Ar-H), 7. 35 – 7. 27 (m, 2H, Ar-H), 6. 72 (dt, 1H, Ar-H), 6. 46 – 6. 35 (m, 2H, Ar-H), 6. 08 (s, 1H, OH), 5. 34 (d, 2H, CH₂), 3. 94 (s, 2H, NH₂)
- 4-amino-2-([2-(3-nitrophenyl)-1H-benzimidazole-1-yl]methyl)phenol (VIII c)
 ATR : OH(3537. 64 cm⁻¹), NH₂(3413. 39 cm⁻¹), CH₂(2919. 97, 2855. 19cm⁻¹), Ar C=C (1605. 05 cm⁻¹), CH₂ bend(1478. 35 cm⁻¹), R-NO₂ (1377. 56 cm⁻¹) para substitution (828. 75cm⁻¹), ¹HNMR (300MHz, DMSO) : δ 8. 47 (t, 1H, Ar-H), 8. 29 (ddt, 2H, Ar-H), 7. 84 – 7. 76 (m, 1H, Ar-H), 7. 63 (t, 1H, Ar-H), 7. 52 – 7. 44 (m, 1H, Ar-H), 7. 36 – 7. 27 (m, 2H, Ar-H), 6. 54 – 6. 48 (m, 2H, Ar-H), 6. 36 (s, 1H, Ar-H), 6. 30 (dd, 1H, OH), 5. 34 (d, 2H, CH₂), 3. 93 (s, 2H, NH₂)

3.2 Anticancer Activity

Compound VIII a and VIII b showed 100% cytotoxicity against colon cancer. The compound VI c showed 89% cytotoxicity against lung cancer. Against prostate cancer compound VIII a resulted in 73% cytotoxicity. The compound VIII a and VI a showed 94%, 93% cytotoxicity respectively against breast cancer.

Table 2: *In vitro* Cytotoxicity against human cancer cell lines

Tissue		Lung	Prostate	Colon	Breast
CELL lines		A549	PC-3	HCT-116	MCF-7
CODE	CONC(100μM)				
VI a	100	75	32	87	93
VI b	100	70	65	90	81
VI c	100	89	37	71	75
VI d	100	22	23	10	47
VIII a	100	83	73	100	94
VIII b	100	77	39	100	89
VIII c	100	28	30	8	50



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4. Conclusion

A series of Mannich base derivatives were synthesised and screened for biological activity. So, it can be concluded that the designed compounds are potent anticancer agent.

The diverse aspects clearly show the high potential of Mannich bases and some compounds showed significant anticancer (100%) activity. In summary, all the Mannich bases of Benzimidazole derivatives showed promising anticancer activity. This work will hopefully be used for further development of potential drugs.

5. Acknowledgement

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