

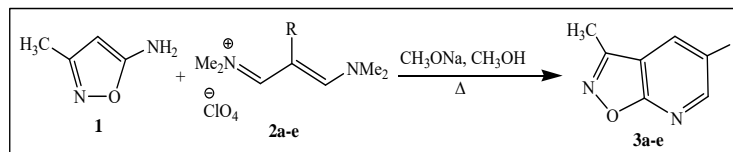
Efficient One-Pot Synthesis of Fused Isoxazolo [5,4-b] Pyridines

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Abstract: A series of substituted isoxazolo[5,4-b]pyridines derivatives has been prepared in one step from condensation of symmetrical vinamidinium salts with 5-amino-3-methylisoxazole. This efficient synthesis has the advantages of high yields and easy operation.

Keywords: isoxazolo[5,4-b]pyridines, vinamidinium salts, 5-amino-3-methylisoxazole, cyclocondensation.



Scheme 1: Synthesis of substituted isoxazolo[5,4-b]pyridines derivatives 3a-e

1. Introduction

The derivatives of isoxazopyridines were already shown to have a very broad spectrum of pharmacological activity, including bacteriostatic, analgesic, anti-inflammatory, antiasthmatic, cardiotoxic, hypotensive, myolytic, antilipidemic and anxiolytic [1]-[10].

Two techniques of Friedländer condensation have been reported for the synthesis of isoxazopyridines. The first method is a Friedländer annulation in the presence of pyridine as the solvent and base catalyst [11]. The second is the condensation of substituted isoxazoles with β -dicarbonyl compounds in a one-step reaction [12]- [14].

Hence, the development of flexible, efficient and simple methodology is still needed for the synthesis of 5-substituted isoxazolo[5,4-b]pyridines.

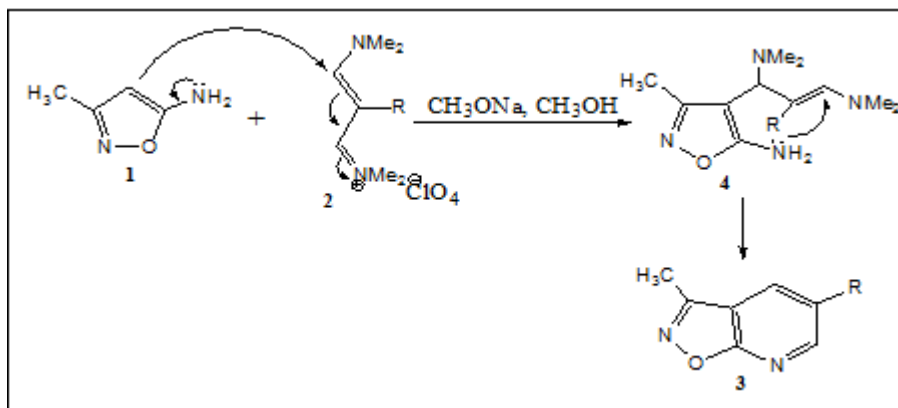
On the other hand, it is well known that vinamidinium salts have been used to prepare many different heterocyclic compounds including isoxazoles [15], pyrazoles [15], pyrimidines [15], [16], pyrroles [17] and thiophenes [18]. Moreover, We have reported the preparation of pyridin-2-ones [19], 3-acyl-6-aryl-2-pyridones [20], pyrido[2,3-d]pyrimidines [21], 2-amino-5-arylpyridine-3-carbonitrile [22] and pyrazolo[3,4-b]pyridin-3-ones [23] using vinamidinium salts. In continuation of our efforts towards

the synthesis of functionalized aromatic and heterocyclic compounds, we present in this work a new and efficient synthetic approach to 5-substituted isoxazolo[5,4-b]pyridines based on vinamidinium salts chemistry.

2. Results and Discussion

The present work is devoted to a one-pot synthesis of derivatives of 3-methylisoxazolo [5,4-b]pyridines (Scheme 1). The readily available 5-amino-3-methylisoxazole **1** was reacted with symmetrical vinamidinium salts perchlorates **2a-e** [24] in methanol at 80°C for 24 h in the presence of sodium methoxide, to furnish the desired 5-substituted isoxazolo[5,4-b]pyridines **3a-d** in excellent yields (Table 1). Alternatively, when the reaction was carried out using the 2-(N-phthaloyl)-1,3-bis(dimethylimmonio)propane diperchlorate **2e**, **2-(3-methylisoxazolo[5,4-b]pyridin-5-yl)isoindoline-1,3-dione 3e** was obtained in good yield. The imidemoiety provides a synthetic handle for subsequent functionalization of bicyclic compound **3e** to a variety of different structures.

We assumed that the formation of **4** was directed via initial Michael addition of β -carbon of the enamino group of 5-amino-3-methylisoxazole **1** to the symmetrical vinamidinium salts **2** and its further cyclisation due the nucleophilic attack by the NH_2 -group giving the isoxazolo[5,4-b]pyridines **3** (scheme 2).



Scheme 2: A possible mechanism for the reaction shown in Scheme 1

3. Conclusion

In conclusion, we have developed an efficient and applicable protocol for the synthesis of 5-substituted isoxazolo[5,4-b]pyridines by the use of vinamidinium salts and 5-amino-3-methylisoxazole.

4. Experimental Section

4.1 General experimental procedure for the preparation of 3a-e:

To a flame dried one-necked round-bottomed flask equipped with magnetic stirring, reflux condenser and nitrogen atmosphere was added vinamidinium salt **2** (0.48 mmol), 5-amino-3-methylisoxazole **1** (0.48mmol), sodium methoxide (0.96mmol) and anhydrous methanol (6mL). The mixture was allowed to reflux overnight under a nitrogen atmosphere in an oil bath. The flask was cooled to room temperature and the obtained precipitate was filtered off to give product **3**.

Table 1: Synthesis of 5-substituted isoxazolo[5,4-b]pyridines 3a-e

Entry	vinamidinium salt	R	Product ^a	Yield (%)
1	2a	Ph	3a	96
2	2b	<i>p</i> -MeOC ₆ H ₄	3b	98
3	2c	OC ₆ H ₅	3c	93
4	2d	<i>p</i> -ClC ₆ H ₄	3d	95
5	2e	N-phthaloyl	3e	90

^aAll products were characterized from their ¹H NMR, ¹³C NMR and mass spectroscopic data

4.1 Spectral data of products 3a-e:

3-methyl-5-phenylisoxazolo[5,4-b]pyridine(3a):

Yield: 96 %; orange solid; mp = 257°C ; ¹H-NMR (300MHz, CDCl₃) δ (ppm): 2.67 (s, 3H), 7.41-7.61 (m, 5H), 8.17 (d, *J*=2.1 Hz, 1H), 8.83 (d, *J*=2.1 Hz, 1H); ¹³C-NMR (75 MHz, CDCl₃) δ : 169.2, 155.7, 150.0, 137.2, 133.6, 129.3, 129.2, 128.3, 127.5, 113.8, 10.9; Anal. Calcd for C₁₃H₁₀N₂O(210.24) : C, 74.27; H, 4.79; N, 13.32 %. Found: C, 74.24; H, 4.77; N, 13.35 %. Mass *m/z* (EI, 30 eV): M⁺ 210.

5-(4-methoxyphenyl)-3-methylisoxazolo[5,4-b]

pyridine(3b): Yield: 98 %; yellow solid; mp = 251°C ; ¹H-

NMR (300MHz, CDCl₃) δ (ppm): 2.65 (s, 3H), 3.90 (s, 3H), 7.05 (d, *J*= 8.7 Hz, 2H), 7.52 (d, *J*=8.7 Hz, 2H), 8.10 (d, *J*=2.6 Hz, 1H), 8.78 (d, *J*= 2.6 Hz, 1H); ¹³C-NMR (75 MHz, CDCl₃) δ : 168.9, 159.8, 155.7, 149.7, 133.3, 129.6, 128.7, 128.0, 114.7, 113.7, 55.4, 10.90 ; Anal. Calcd for C₁₄H₁₂N₂O₂ (240.26) : C, 69.99; H, 5.03; N, 11.66 %. Found: C, 70.04; H, 5.01; N, 11.68%. Mass *m/z* (EI, 30 eV): M⁺ 240.

3-methyl-5-(phenoxy)isoxazolo[5,4-b] pyridine(3c):

Yield: 93 %; white solid; mp = 163°C; ¹H-NMR (300MHz, CDCl₃) δ (ppm): 1.99 (s, 3H), 6.51-6.78 (m, 3H), 7.23 (d, *J* = 8.2 Hz, 2H), 8.33 (s, 1H), 8.51 (s, 1H) ; ¹³C-NMR (75 MHz, CDCl₃) δ : 170.2, 160.4, 160.2, 159.1, 159.0, 129.0, 128.9, 119.5, 114.5, 114.4, 11.3; Anal. Calcd for C₁₃H₁₀N₂O₂ (226.23) : C, 69.02; H, 4.46; N, 12.38 %. Found: C, 68.99; H, 4.51; N, 12.35%. Mass *m/z* (EI, 30 eV): M⁺ 226.

5-(4-chlorophenyl)-3-methylisoxazolo[5,4-b] pyridine

(3d): Yield: 95 %; white solid; mp= 212°C; ¹H-NMR (300MHz, CDCl₃) δ (ppm): 2.66 (s, 3H), 7.28 (m, *J* = 8.7 Hz, 2H), 8.05 (d, *J* = 8.7 Hz, 2H), 8.28 (d, *J*=1.8 Hz, 1H), 8.74 (d, *J*=1.8 Hz, 1H); ¹³C-NMR (75 MHz, CDCl₃) δ : 169.1, 155.7, 149.6, 134.2, 133.3, 129.5, 129.2, 127.0, 125.9, 113.7, 10.8 ; Anal. Calcd for C₁₃H₉ClN₂O(244.68) : C, 63.82; H, 3.71; N, 11.45; Cl, 14.49%. Found: C, 63.78; H, 3.76; N, 11.47%. Mass *m/z* (EI, 30 eV): M⁺ 244(100%), 246(32.5%).

2-(3-methylisoxazolo[5,4-b]pyridin-5-yl)isoindoline-1,3-

dione(3e): Yield: 90 %; Brown solid; mp= 192°C; ¹H-NMR(300MHz, CDCl₃)δ (ppm): 2.16 (s, 3H), 7.09 (s, 1H), 7.78 (d, *J* = 8.4 Hz, 2H), 7.92 (d, *J* = 8.4 Hz, 2H), 8.25 (s, 1H) ; ¹³C NMR (75 MHz, CDCl₃)δ ((ppm): 172.15, 168.96, 168.86, 161.25, 156.75, 153.26, 134.29, 134.21, 132.26, 132.24, 123.91, 123.85, 104.83, 101.27, 12.06 ; Anal. Calcd for C₁₅H₉N₃O₃(279.25) : C, 64.52; H, 3.25; N, 15.05 %. Found: C, 64.49; H, 3.27; N, 15.01 %. Mass *m/z* (EI, 30 eV): M⁺ 279.

5. Acknowledgment

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References

- [1] Chiarino, D.; Napoletano, M.; Sala, A. J. Heterocycl. Chem.1988, 25, 231.
- [2] Kidwai, M.; Nagi, N. Monatsh. Chem.1997, 85, 128.
- [3] Adhikari, W. A.; Badiger, W. V. Arch. Pharm.1987, 320,1124.
- [4] Kleinschroth, J.; Mannhard. K. Patent Ger. Offen.3447388, 1986;Chem. Abstr.1986,105, 172441x.
- [5] Koike, K.; Nishino, H. Yoshimoto, A. Patent Jpn.0240385, 1990;Chem. Abstr.1990,113, 40675g.
- [6] Perregaard, J. Patent Fr. 2494692, 1982; Chem. Abst.1982, 97, 182394q.
- [7] Perregaard, J. Patent Fr. 249691, 1980; Chem. Abst.1882, 97, 182395r,
- [8] Fuijkawa, Y., Suzuki, M. Patent. Eur. 369 323, 1990 ; Chem. Abst.1990, 113, 1359s.
- [9] Madsen, U.; Bang-Anderson, B.; Brehm, L.; Christiansen, J.; Ebert, B.; Kristofferson, J.; Lang, Y.; Krogsgaard-Larsen, P. J. Med. Chem.1996, 39, 1682.
- [10] Nordman, R.; Graff, P.; Maurer, R.; Gaehwiler, B. H. J.Med. Chem.1985, 28, 1109.
- [11]Gewald, K.; Bellmann, P.; Jänsh, H. J. Liebigs Ann. Chem.1980, 10,1623.
- [12]Rajanarendar, E.; Srinivas, M.; Ramu, K. Synth. Commun.2003, 33, 3077.
- [13]Rajanarendar, E.; Srinivas, M.; Ramu, K.Ind. J. Chem.2005, 44B, 1927.
- [14]Wafaa, S. H.; Mona, E. I.; Hanafi, H. Z.Synth. Commun.2013, 43, 2393.
- [15]Lloyd, D.; McNabb, H. Angew. Chem. Int. Ed. Engl.1976, 15, 459.
- [16]Hadj Ayed, M. A.; Khiari, J.E.; Ben Hassine, B. Mol. Divers.2008, 12, 61.
- [17]Mathew, T. W.; David, G. C.; Timothy, M. S.; Stanton, Q. S. Tetrahedron Lett.2010,51, 4150.
- [18]Ryan, T. C.; Stanton, Q. S. Tetrahedron Lett.2005,46, 1320.
- [19]Khiari, J.E.; Gmiza T.; Hadj Ayed, M. A.; Ben Hassine, B. Synth. Commun.2007, 37, 3939.
- [20]Gmiza, T.; Khiari, J.E.; Hadj Ayed, M. A.; Ben Hassine, B.Tetrahedron Lett.2011, 52, 764.
- [21]Hadj Ayed, M. A.; Khiari, J.E.; Ben Hassine, B. Synth.Commun.2012, 42, 1824.
- [22]Fayçel, J.; Samar, C.; Jameleddine, K. J. Heterocycl. Chem. 2013,50, 206.
- [23]Chtiba, S.; Jemmezi, F.; Khiari, J. E.Tetrahedron Lett.2011, 52, 3648.
- [24]Gupton, J. T.; Krolkowski, D. A.; Yu, R. H.; Riesinger, S. W.; Sikorski, J. A. J. Org. Chem.1990, 55, 4735.