

Antibacterial Effectiveness of NaOCl and Chlorhexidine in Infected Root Canal

Mimoza Canga MD,PhD¹, Brunilda Subashi MSc²

¹University of Tirana, 9400 Albania.

²University of Vlora, 9401 Albania

Abstract: ***Aim:** Root canal irrigants play an important role in the optimization of the root canal preparation, which is essentially a chemo-mechanical procedure. To evaluate the antimicrobial activities of the most widely used endodontic irrigants: sodium hypochlorite 2.5% and chlorhexidine gluconate 2%. **Materials and methods:** In this research were included 20 patients of both genders. Patients were divided in two groups: group I (n=10) patients where as a root canal irrigant was used NaOCl 2.5%, group II (n=10) patients as a root canal irrigant was used CHX 2%. A sample was taken with sterile paper point before and after the instrumentation, and again after three days after irrigants was used in infected root canal. The sample was set in sterile test tubes and was send to the microbiological laboratory. For statistical analysis was used the Anova test. **Results:** Our results have shown that NaOCl reduce aerobic bacteria after three days SD 637.5 ± 368.8, CHX SD 318.6 ± 431.4 Anaerobe bacteria was significantly reduced after three days that irrigants was used notably CHX SD 668.1 ± 415, SD NaOCl 222 ± 25.4 **Conclusion:** 2.5% NaOCl has a better antibacterial action than 2% CHX.*

Keywords: antibacterial effect, chlorhexidine, irrigants, sodium hypochlorite.

1. Introduction

Bacteria have long been recognized as the primary etiologic factors in the development of pulp and periapical lesions [2]. The endodontic treatment depends from the reduction and elimination of bacteria which are present in the root canal [2]. After the instrumentalization of the root canal, the curved root canals can still have some pulpar tissue, microorganisms and dentine dentrites and these is the reason that along with the biomechanical instrumentation should also be used various irrigants to destroy these microorganisms [3]. The irrigants have mechanical and biological action on the root canal. The mechanical action includes the detrite removal, canal lubrication, removing the organic and inorganic components and tooth whitening. The biological action is strongly related to its antimicrobial effect. During the endodontic treatment of root canal are used many different types of irrigants [2],[3]. Complete debridement and adequate elimination of microbial irritants, including microorganisms and their toxins, is a fundamental prerequisite for successful endodontic therapy [16], [18], [24]. Root canal irrigants play an important role in the optimization of the root canal preparation, which is essentially a chemo-mechanical procedure [24]. Potent antimicrobial activity, dissolving of remaining pulp tissues with no systemic hazards, reducing instrument friction during mechanical preparation and availability are among the main requirements for an ideal root canal irrigant [19]. Sodium hypochlorite (NaOCl) is the most common irrigant used in root canal treatment [5]-[6], [8]-[9]. Sodium hypochlorite NaOCl has widely been accepted as a root canal irrigant since its first reported use by Walker in 1936 [1], [4], [13], [25], [29]. It mainly acts as a potent antimicrobial agent and an effective organic solvent for vital, necrotic and fixed tissue. It is an effective antimicrobial agent, good lubricant and an excellent organic solvent [21]. However, it is highly irritating to the periapical tissues, especially at high concentrations. NaOCl is an efficient organic solvent that causes dentin degeneration

because of the dissolution of collagen by the breakdown of the bonds between carbon atoms and disorganization of the proteic primary structure [19]. Chlorhexidine (CHX) has been suggested as either an alternative or an adjunct root canal irrigant because of its antimicrobial qualities and substantivity [7], [10]-[12]. Chlorhexidine gluconate (CHX) is routinely used in dentistry as a mouth rinse in the prevention and treatment of periodontal disease and caries [14]-[15], [17]. Chlorhexidine gluconate (CHX) is a potent antimicrobial agent, holds substantivity and has a low grade toxicity. CHX is bacteriostatic at low concentrations (0.2%), bactericidal at high concentrations (2%), and adsorbs to dental tissue resulting in its prolonged gradual release at therapeutic levels. However, chlorhexidine is unable to dissolve pulp tissue and debris may remain on canal walls, obstructing the dentinal tubules [20], [22]-[23]. CHX of 2% can cause desquamation of the mucosae of oral cavity, tooth coloration and toxic effect in epithelial cells. For this reason during the endodontic treatment should be used antibiotics that have antiseptic characteristics and reduced side effects [26],[28].

2. Aim

To compare the antibacterial effects between both irrigants.

3. Objective

Is to establish an acceptance score for a best case series in determining the antibacterial effects of two root canal irrigants: 2.5% sodium hypochlorite compared to 2% chlorhexidine gluconate.

4. Materials and Method

In our study we included 20 patients of both genders age 21-59 years old. In the first group (n=10) the patients were treated with NaOCl 2.5% for the disinfection of the root canal. In the second group (n=10) the patients were treated

CHX 2% for the disinfection of the root canal. The treated patients are diagnosed with chronic parodontitis apicalis and pulp necroses.

5. Statistical Analysis

The spectrophotometric results were analyzed using one-way ANOVA. The mean difference is significant at the 0.05 level, and $P < 0.05$ was considered to be statistically different.

6. Results

In the first group, the disinfection of the root canal were treated with NaOCl 2.5%. After determination of the diagnoses we performed radiographic examination on the teeth. Then we performed the opening of the root cavity. Immediately after treatment and instrumentalization of the root canal we took a sample to test for aerobic and anaerobic bacteria by using a sterile Pointit paper. After receiving the first sample, we estimated the length of tooth root canal 1mm shorter, than the actual length of the tooth root canal. Then we instrumented the root canal with the K-File instrument. The irrigation is done with 2.5% NaOCl. In order to neutralize the tooth root canal by NaOCl dissolution, the root canal is neutralized by dilution with NaCl 0.9% and the root canal was dried with sterile pointin paper. After drying the tooth root canal a second sample is taken from the canal in the same conditions as the first sample. After instrumentation of root canal we put there NaCl 2.5%. The interpretation of results is done with the help of apparatus Vitek 2.

Table 1: NaOCL and CHX reduce aerobe bacteria after three days

SD	Aerobe
NaOCL	637.5±368.8
CHX	318.6±431.4

In the second group the root canal is disinfected with CHX 2%. In the second group we also used the same techniques and procedures and the instrumentation of the tooth root canal, the only difference is that the irrigation is made with CHX 2% and inorganic components are also removed with EDTA 17%. The final irrigation is made with dilution of CHX 2%. In this group a first smear was taken before the canal instrumentalization and the second swab was taken after the canal instrumentalization and third swab was taken three days after this process.

Table 2: NaOCL and CHX reduce anaerobe bacteria after three days

SD	Anaerobe
NaOCL	222 ± 25.4
CHX	668.1 ± 415

In the first group we included (n= 10) patients treated with 2.5% NaOCl. In the second group (n=10) the patients were treated with 2% CHX. In the first and second group the samples are taken three times, one before treatment, one after treatment and three days after treatment. Before the

treatment all the sample taken from the root canals of both groups were positive for aerobic and anaerobic bacteria. Immediately after treatment 30% of samples taken in the first group were positive. After three days of treatment in samples of the two groups were found aerobic and anaerobic bacteria. Before the treatment 60% of samples taken in the first group were positive and 20% of the samples in the second group were positive. Immediately after treatment 10% of samples taken in the first group were all positive, while in the second group they did not have anaerobic bacteria.

7. Discussion

Carlos E., et al, Estrela C et al. in their study have reported a similar action between 2.5% NaOCl and 2% CHX [4], [8]. It has been advocated use NaOCl in concentrations ranging from 0.5% to 5.25%, but there has been no agreement on the optimal concentration. Baumgartner and Canga et al., confirmed that 2.5% NaOCl is extremely effective in removing vital pulp tissue from dentinal walls [2], [5]-[6]. Siqueira et al., reported that using NaOCl in 2.5% concentrations may significantly reduce the endodontic infection, but might not consistently dissolve all pulpal remnants in a reasonable time [20],[21],[23]. Some authors emphasized that the antibacterial effectiveness of 2.5% concentrations of NaOCl might be improved by usage larger volumes of solution and continuous exchange of agent. Siqueira et al. Reported that instrumentation and irrigation by using 2.5% NaOCl provided a decrease of 99.9% in the count of viable bacteria in the root canal. Vianna et al., compared that 2% CHX and 2.5% NaOCl in vitro and demonstrated that 2% CHX was more efficient to inhibition of growth of *E. faecalis*, *Staphylococcus aureus* and *Candida albicans* [26]. Jeansonne and White compared the antibacterial effect of different irrigation solutions against anaerobic bacteria and noticed that 2% CHX was more effective than 5.25% NaOCl, but the differences were not statistically significant [12],[28]. White et al., claimed that, after instrumentation, CHX continues to be released while 48-72 h. The stability of antibacterial efficacy of endodontic irrigants, especially in prolonged periods of treatment, is very important [28]. It was concluded that NaOCl should remain in the root canal for a considerable time, so that it can act on the residual bacteria placed in the irregularities in the root canal.

8. Conclusions

From the results of our study we can conclude: 2.5% NaOCl and 2% CHX have a remarkable antibacterial action in the infected root canal. 2.5% NaOCl has a better antibacterial action than 2% CHX.

9. Acknowledgment

To perform and report this study no financial support from any companies or from a national or international funding source was received.

References

- [1] Abbou-Rass M, Piccinino MV:” The effectiveness of four clinical irrigation methods of the removal of debris”. Oral Surg Oral Med Oral Pathol, 54(3), 323-8.
- [2] Baumgartner JC, Falkler WA.” Bacteria in the apical 5 mm of infected root canals”. J Endod. 1991; 17:380-3.
- [3] Brugnera A, Zanin F, Barbin EL, Spano JC, Santana R, Pecora JD:” Effects of ER: YAG laser irradiation on radicular dentine permeability using different irrigating solutions”, Lasers Surg Med, 2003, 33(4):256-9.
- [4] Carlos E, Silva JA, Alencar AHG, Leles CR, Decurcio DA:”Efficacy of sodium hypochlorite and chlorhexidine against Enterococcus faecalis-a systematic review”, Journal of Applied Oral Science 2008, 16(6):1678-7757.
- [5] Canga M, Malagnino V A. “Testing of the antimicrobial efficiency of helbo laser and NaOCl against Candida Albicans”. IJSR Volume 4, issue 1, January 2015, 22-25.
- [6] Canga M, Malagnino V A.”Antimicrobial efficiency of Naocl and helbo laser against Enterococcus faecalis “IJSR Volume 4, issue 1, January 2015, 1035-1038.
- [7] Cervone F, Tronstad L, Hammond B. “Antimicrobial effect of chlorhexidine in a controlled release delivery system”. Endod Dent Traumatol. 1990; 1:33-6.
- [8] Estrela C, Estrela CR, Barbin EL, Spanó JC, Marchesan MA, PécoraJD.”Mechanism of action of sodium hypochlorite”. Braz Dent J. 2002; 13:113-117.
- [9] Kataoka H, Yoshioka T, Suda H, Iwasaki N, Takahashi H, et al.” Effect of NaOCl treatment on bonding to root canal dentin using a new evaluation method”. Dent Mater J. 2001; 20:24-33.
- [10] Leonardo MR, TanomaruFilho M, Silva LA, Nelson Filho P, Bonifácio KC, Ito IY.”*In vivo* antimicrobial activity of 2% chlorhexidine used as a root canal irrigating solution”. J Endod. 1999; 25:167-171.
- [11] Ferguson, D.B., Marley, J.T., Hartwell, G.R., (2003). “The effect of chlorhexidine gluconate as an endodontic irrigant on the apical seal: long-term results”. Journal of Endodontics. 29(2), 91-94.
- [12] Jeansonne M. J., White R. R.” A Comparison of 2.0% ChlorhexidineGluconate and 5.25% Sodium Hypochlorite as Antimicrobial Endodontic Irrigants”. Journal of Endodontics 1994, 20(6), 276-278.
- [13] Ferraz CC, Vianna ME, Berber VB, Teixeira FB, Souza-FilhoFJ.”*In vitro* antimicrobial activity of several concentrations of sodium hypochlorite and chlorhexidine gluconate in the elimination of Enterococcus faecalis”. IntEndod J. 2001; 34:424-8.
- [14] Jeansonne MJ, White RR.” A comparison of 2.0% chlorhexidine gluconate and 5.25% sodium hypochlorite as antimicrobial endodontic irrigants”. J Endod 1994; 20:276-8
- [15] Ercan E, Ozekinci T, Atakul F, G?l K: “Antibacterial activity of 2% chlorhexidine gluconate and 5.25% sodium hypochlorite in infected root canal: in vivo study”, J Endod 2004, 30(2): 84-7.
- [16] Ercan E, Dulgergil T, Yavuz I;” The effects of antibacterial solutions on microorganisms isolated from infected root canals in vivo”. Biotechnol. & Biotechnol. EQ 2006, 20: 1.
- [17] Gomes B P F A, Souza S F C, Ferraz C C R, Teixeira F B, Zaia A A, Valdriht L, Souza- Filho F J:” Effectiveness of 2% chlorhexidine gel and calcium hydroxide against Enterococcus faecalis in bovine root dentine in vitro”. IntEndod J 2003, 36, 267-75.
- [18] Haapasalo M, Shen Y, Qian W, Gao Y.” Irrigation in endodontics”. Dent Clin North Am. 2010; 54:291-312.
- [19] Mohammadi Z, Abbott PV.” Antimicrobial substantivity of root canal irrigants and medicaments: A review”. AustEndod J. 2009; 35:131-9.
- [20] Siqueira JF, Jr, Rocas IN, Favieri A, Lima KC.” Chemo-mechanical reduction of the bacterial population in the root canal after instrumentation and irrigation with 1%, 2.5%, and 5.25% sodium hypochlorite”. J Endod. 2000; 26:331-4.
- [21] Siqueira JF, Jr, Guimaraes-Pinto T, Rocas IN.” Effects of chemo-mechanical preparation with 2.5% sodium hypochlorite and intra-canal medication with calcium hydroxide on cultivable bacteria in infected root canals”. J Endod. 2007; 33:800-5.
- [22] Martinho FC, Chiesa WM, Marinho AC, Zaia AA, Ferraz CC, Almeida JF, et al.”Clinical investigation of the efficacy of chemo-mechanical preparation with rotary nickel-titanium files for removal of endotoxin from primarily infected root canals”. J Endod. 2010; 36:1766-9.
- [23] Siqueira JF, Jr, Alves FR, Almeida BM, de Oliveira JC, Rocas IN.” Ability of chemo-mechanical preparation with either rotary instruments or self-adjusting file to disinfect oval-shaped root canals”. J Endod. 2010; 36:1860-5
- [24] Zehnder M.” Root canal irrigants”. J Endod. 2006; 32:389-98.
- [25] Sassone LM, Fidel RA, Murad CF, Fidel SR, Hirata R., “ Antimicrobial activity of sodium hypochlorite and chlorhexidine by two different tests”. AustEndod J. 2008; 34:19-24.
- [26] Vianna ME, Gomes BP, Berber VB, Zaia AA, Ferraz CC, de Souza-Filho FJ. “*In vitro* evaluation of the antimicrobial activity of chlorhexidine and sodium hypochlorite”. Oral Surg Oral Med Oral Pathol Oral RadiolEndod. 2004; 97:79-84.
- [27] Valera MC, Rego JM, and Jorge AOC:” Effect of sodium hypochlorite and five intracanal medications on candida albicans”, J Endod, 2001, 27(6): 401-8.
- [28] White RR, Shays GL, Janer LR:”Rezidual antimicrobial activity after canal irrigation with chlorhexidine”: J Endod, 1997, 31, 23-229.
- [29] Zou L, Shen Y, and Li W, Hapasalo M:” Penetration of sodium hypochlorite into dentine”, J Endod, 2010, 36(5): 793-6.

Author Profile



Dr. Mimoza Canga is Laureate at the University Sapienza, Roma- Italy. PhD University of Tirana. Currently works in the University of Vlora, Department of Public Health, Vlora-Albania.



Brunilda Subashi born in 30 October 1981. Has graduated a General Nurse in University of Vlora. Has graduated MSc in University of Vlora. Works as a lecturer at Public Health Faculty at University of Vlora-Albania.