

Figure 3: PL-studies of (a) Pure TiO₂ (b) Ag-TiO₂ (c) Mg-TiO₂ (d) Bi-TiO₂

3.4 Field Emission Scanning Electron Microscopy (FE-SEM) Analysis

Figure 4 shows the FESEM micrographs of the pure and doped TiO₂ nanoparticles prepared by acid modified sol-gel

method. It has been observed that the TiO₂ nanoparticles annealed at 500°C were almost reveals that the primary particles are quite uniform in size, quite clean and roughly spherical in shape, and that the agglomerates are fused together to form comparatively smaller irregular grains giving rise to highly porous materials which enhancing the photovoltaic performance. The FE-SEM micrographs of the pure TiO₂ nanoparticles depicted in figure (a) has non-uniform distribution of spherical particles.

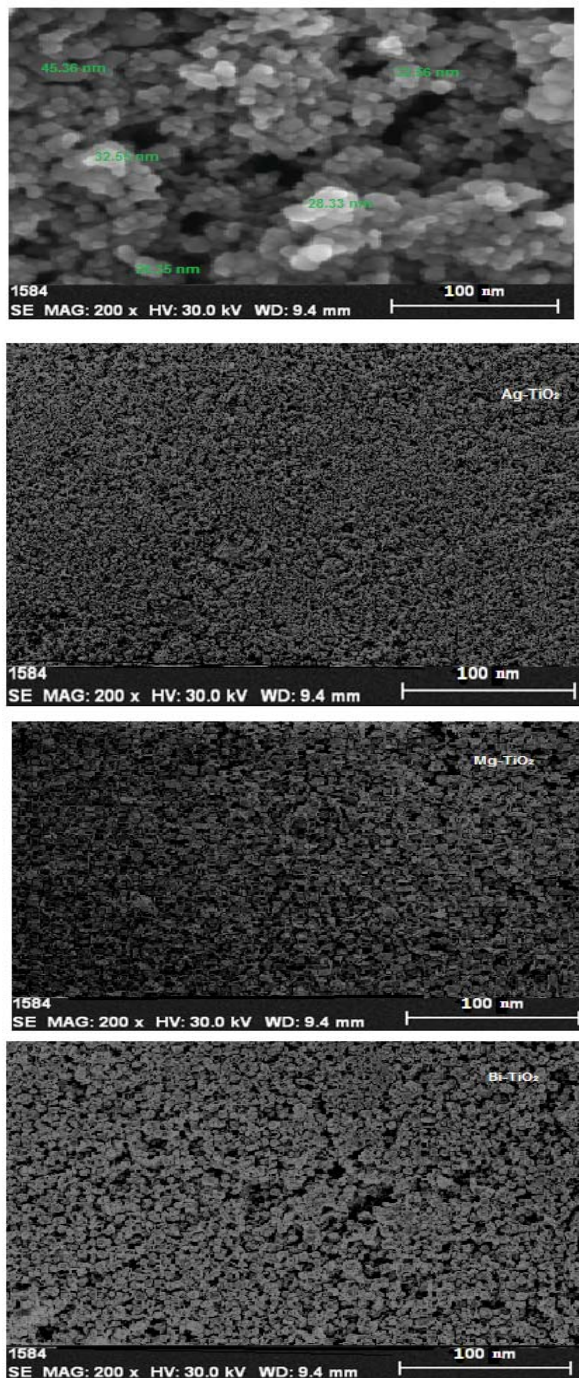


Figure 4: FESEM Micrographs (a) Pure TiO₂ (b) Ag-TiO₂ (c) Mg-TiO₂ (d) Bi-TiO₂

3.5 EDS Analysis

The semi quantification of elemental analyses to identify the weight percentage of major and minor elements present in the samples were done using Energy Dispersive X-ray Spectrometer (EDS), JEOL model, JSD-5610 LV with an

accelerating voltage of 20KV. The result of energy dispersive X-ray spectroscopic (EDS) analysis of pure TiO₂ and doped TiO₂ nanopowders are shown in fig 5(a) & 5 (b, c & d). Trace elements are estimated by determining the percentage abundance of elements such as Ti, O & Zn present in the sample. In pure TiO₂ the concentration of titanium is 89.20% and the concentration of oxygen is 10.80% this result gives to the titanium and oxygen presents in sample.

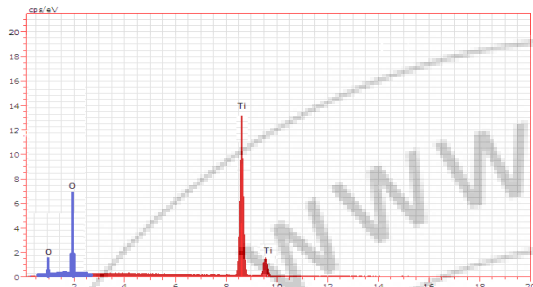


Figure 5 (a)

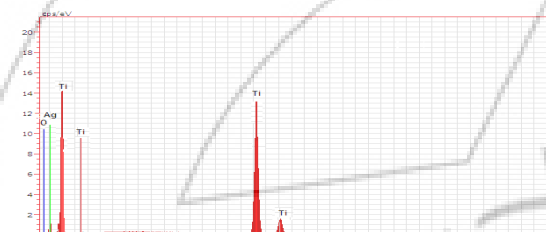


Figure 5 (b)

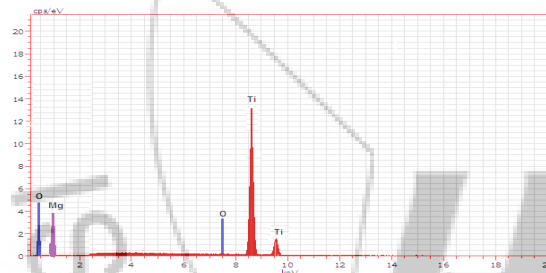


Figure 5 (c)

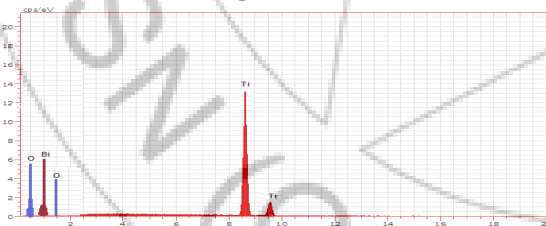


Figure 5 (d)

Figure 5: EDS Analysis (a) Pure TiO₂ (b) Ag-TiO₂ (c) Mg-TiO₂ (d) Bi-TiO₂

Table 2 (a)

Sample	Element	Weight (%)	Atomic (%)
Pure TiO ₂	O K	32.16	61.40
	Ti K	67.84	38.60
	Total	100.00	100.00

Table -2 (b)

Sample	Element	Weight (%)	Atomic (%)
Ag- TiO ₂	O K	35.25	63.60
	Ti K	61.45	34.25
	Ag K	3.30	2.15
	Total	100.00	100.00

Table -2 (c)

Sample	Element	Weight (%)	Atomic (%)
Mg- TiO ₂	O K	36.34	65.55
	Ti K	59.40	31.30
	Mg K	4.26	3.15
	Total	100.00	100.00

Table 2 (d)

Sample	Element	Weight (%)	Atomic (%)
Bi- TiO ₂	O K	33.55	62.33
	Ti K	60.23	33.42
	Bi K	6.22	4.25
	Total	100.00	100.00

Table 2: EDS Analysis (a) Pure TiO₂ (b) Ag-TiO₂ (c) Mg-TiO₂ (d) Bi-TiO₂

4. Conclusions

The pure and doped TiO₂ nanoparticles were successfully prepared by sol-gel method. The UV-Visible and PL result shows doped titanium nanoparticles extend the absorption edge to the visible light range and make the red shift more distinct and also analyses the band gap value. The Ti-O bond of anatase phase in all the synthesized samples was identified by FT-IR measurements. The FE-SEM analysis reveals that the morphology of doped TiO₂ was smooth and well defined spherical shape with grain size 25-50 nm with minimal agglomeration compare to pure TiO₂. Also analyses the morphology of doped TiO₂ was regular arrangement, spherical shape, uniform size, and good packing density. The doped and pure elements are identified from EDX spectrum, in which the spectrum confirms the elemental compositions of presented samples.

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