system attributed to structural changes[55]. This research work proved that some of the novel molecules can stabilize the biochemical part of living beings[56-59].The measured and calculated thermodynamic parameters are useful to know the interactions like solute-solute or solute-solvent or solvent-solvent type.

6. Conclusion

The parameters are correlated with aqueous amino acid and glycol ether. The system containing aqueous amino acid and glycol ether has weak intermolecular H-bonding. The decrease in density, viscosity and ultrasonic velocity indicates the decrease in solute-solvent and solvent-solvent interactions which results structure-breaking of the solvent. The acoustical parameters proved that H-bonding interactions are weak at higher concentration. The increase in adiabatic compressibility with present work suggests that the medium become more and least compressible. The L_f is again a predominant factor in determining the existing interactions among the components of the mixture.

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