# HPLC Fingerprint Analysis with in - vitro Antimalarial Activity and Bioassay Guided Fractionation of Two Organic Extract from *Ageratum conyzoides* Leaves

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**Abstract:** We investigated the antimalarial activities of fractions from cyclohexane and dichloromethane extracts of Ageratum conyzoides Leaves. These extracts were fractionated using preparative - TLC and six fractions were obtained for each extract. The antimalarial activity of these fractions was evaluated against Plasmodium falciparum. Fractions 4 (F4) and 5 (F5) from dichloromethane extract showed the best antimalarial activities against the parasite with  $IC_{50}$  values of 0.25 and 0.30µg/mL, respectively. The HPLC profile of these most active fractions were developed and showed two and three major peaks respectively for fraction 4 (F4) and 5 (F5). Preliminary phytochemical screening revealed the presence of alkaloids, steroids and terpenoids in the fraction 4, whereas only steroids and terpenoids were found in the fraction 5 (F5). Thus, these fractions could be considered as the prospective fractions to be developed as antimalarial drugs.

Keywords: Ageratum conyzoides, antimalarial activity, fractionation, HPLC, Thin Layer Chromatography (TLC)

#### 1. Introduction

Malaria is a preventable and curable disease, yet it remains a devastating tropical disease, with a high infection and mortality statistics. It is the most prevalent parasitic disease and the most common cause of hospital visitation in Democratic Republic of the Congo [1]. Globally, there are approximately 214 million malaria cases in 97 countries with ongoing malaria transmission and 80% of these cases were reported in the Sub - Saharan Africa. The disease caused approximately 438, 000 deaths, 90% of which were in Sub - Saharan Africa, and 78% of these deaths occur in children under 5 years [2]. The main challenge to the effective management of diagnosed malaria cases is the resistance of the plasmodium parasite to commonly used antimalarial drugs, which results in the non - resolution of symptoms and in treatment failure [3, 4].

Cases of resistance are currently extending to the artemisinin combination therapy with report of resistance to artemisinin drugs coming from Thailand, Vietnam, and Cambodia [2, 5 -7]. Furthermore, no new class of antimalarial drug has beenintroduced in clinical practice in the last 10years [8, 9]. This strongly supports the need for further research into new products that could complement the existing antimalarial, bearing in mind that it takes minimum of 10 years to develop a drug from discovery stage to completion of clinical trials [10]. The development of safe and effective anti - malarial preparations by simple procedures from locally grown medicinal plants, may offer new and complimentary drugs for malaria control, especially in remote geographical locations and in rural areas where modern antimalaria drugs are not readily available and malaria mortality is higher [11].

Ageratum conyzoides (A. conyzoides) (Family: Asteraceae) is one of such plants with folkloric reputation as antimalarial agent. It is an annual herbaceous plant widely distributed in the tropics as an herbal weed. Extracts of the plant have been used in traditional medicines as remedies against wounds and burn, microbial infections, arthrosis, headache, inflammation dyspnea, pain, asthma, spasms, gynaecological diseases, leprosy and other skin diseases [12]. In Africa, *A. conyzoides* is also traditionally used totreat fever, rheumatism, headache, malaria, pneumonia, wounds, burns and colic [13]. An earlier pharmacology study of the plant showed significant positive antimalarial activity in rats [14]. The leaves extracts and essential oil from *Ageratum conyzoides* have been found to have antimicrobial activities *in vitro* [15, 16].

This research was designed to evaluate the the in - vitro antimalarial activity of fractions obtained from cyclohexane and dichloromethane extract of *A. conyzoides* leaves and their phytochemical screening.

#### 2. Materials and Methods

#### 2.1. Plant Collection and Authentication

The leaves of *Ageratum conyzoides* were collected from their natural habitats in Mont - ngafula Township, Kinshasa city, Democratic Republic of the Congo (DRC).

The collected plant leaves were authenticated by INERA (Institute National d' Etudes et RecherchesAgronomiques) Herbarium at Faculty of Science, University of Kinshasa. These leaves were steam - dried at 30°C for 24 hours then milled into powdered used to prepare extracts.

#### 2.2. Plant Extraction and fractionation.

#### 2.2.1. Crude extract

The crude extraction was performed by maceration of 300g dried leaves powder in 1000mL of methanol - water (7: 3) at room temperature for 48 h. The solution was filtered and the filtrate named crude extract was stored and used for preparing organic extracts.

#### 2.2.2. Organic extracts

The crude extract filtrate was extracted successively by two solvents (cyclohexane, and dichloromethane) using a separating funnel. At the end of separation, all phases were evaporated under reduced pressure using a Rotary evaporator for obtaining Cyclohexane and Dichloromethane extracts which were stored in a refrigerator (4°C) until required for all assays.

## 2.3. Fractionation of cyclohexane and dichloromethane extract

Analytical - Thin Layer Chromatography (TLC), preparative - TLC and qualitative - HPLC were used according to the methods described in the literature [17, 18].

2.3.1. Analytical Thin Layer Chromatography (TLC)

The portions of cyclohexane or dichloromethane extracts were dissolved in dichloromethane solvent for obtaining the test solution.

A strip of the precoated silica gel was cut out. Using a capillary pipette, a spot of the test solution was applied on the plate about 1.5 cm from the edge. It was dried using hot air dryer. The strip was lowered into a small chromatographic jar containing the solvent system. The jar was covered with a glass lid. The solvent was allowed to ascend until the solvent front was about  $\frac{3}{4}$  of the length of the strip. The strip was removed and dried by a hot air dryer and viewed under UV lamp at 365 and 254 nm to identify the fluorescing spot. The fluorescent spot was marked and then sprayed with iodine and sulfuric vanillin spray in order to brown and purplish - pink marks respectively. The purplish - pink marks became black after heating the trip at  $120^{\circ}$ C.

The colour reaction was recorded and the relative Retention factor (Rf) value was calculated based on the formula described by Stahl [17]:

 $Rf = \frac{\text{Distance traveled by the streak from the starting point}}{\text{Distance traveled by the solvent from the starting point to the solvent front.}}$ 

#### 2.3.2. Preparative Thin Layer Chromatography (TLC)

Preparative TLC was performed on  $20 \times 20$  mm glass plates precoated with 1.5mm layers of silica gel GF254 (Kieselgel 60 G Merck).

A 7 mm margin on both sides of the plate was marked and the area from the edge of the plate to the mark was not streaked. In the application of fractions, dropping pipettes (10  $\Lambda$ ) were used to apply the various pooled fractions on the activated plates. Drops of the pooled fractions were applied in line to form a straight line streak or band. Each streak was dried before another one was superimposed on it.

The streaked plates were run in a chromatographic chamber containing 40 mL Cyclohexane - dichloromethane - ethyl acetate, 1: 4: 1 and ethyl acetate - cyclohexane, 1: 1 as the eluting solvent respectively for cyclohexane and dichloromethane extracts.

Each plate containing a single pooled fraction was viewed under ultra - violet lamp at 254 and 366 nm in a dark room. The marked separated zones were scrapped off the glass plates with a spatula onto a clean sheet of paper. The scrapings were transferred into centrifuge tubes containing 5 ml of absolute methanol. The content of the centrifuge tube was shaken manually for 10 mins. The eluent was separated from the adsorbent by centrifuging at 2500 rpm for 10 mins. This process was repeated until it was satisfactory that all the eluent was collected as much as possible. The collected eluent in methanol was evaporated to dryness using a hot air oven at the temperature of  $40^{\circ}$ C in order to obtain different fractions which were used for phytochemical screening, and antimalarial test. The most active fractions of dichloromethane extract (4 and 5) were finger - printed by HPLC (Agilent Technology). The concentration of extracts was 10 mg/mL. HPLC optimum conditions: Injection volume: 10  $\Box$ L, Detection wavelength: 280 nm, Mobile phase: Methanol: Trifluoroacetic acid (0.05%) (Gradient condition), Temperature: 25°C, Pump pressure: 400bar, Flow rate: 20 µL/min, and running time: 30min.

#### 2.4. Phytochemical Tests

The cyclohexane and dichloromethane fractions were subjected to phytochemical investigation for identification of alkaloids, flavonoids, anthocyanins, tannins, quinones, saponins, steroids, and terpenoids using standard procedures [19 - 22].

#### 2.5. Antimalarial Activity

#### 2.5.1. Parasites

Fresh clinical isolates of Plasmodium falciparum were obtained from Hospital center of Kindele (Mont - ngafula township), in Kinshasa, Democratic Republic of the Congo (DRC). Rapid diagnostic test was realized for Plasmodium species identification. The parasite density was determined by counting the number of infected erythrocytes. From donor, 4 mL of venous blood was collected in a tube coated with EDTA (Greiner Labortechnik). Samples with monoinfection due to Plasmodium falciparum and a parasite density between 1% and 2% were used for the in vitro antimalarial tests.

2.3.3. HPLC Finger - Printing of most active fractions

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The parasites were cultivated and maintained continuously in human erythrocytes according to previously described methods [23].

#### 2.5.2. In Vitro Antimalarial Assay Procedure.

The antimalarial activity of each fraction was evaluated against the strain of *P. falciparum* using the method described in the literature [24]. A stock solution of  $100\mu$ g/mL of fractions was prepared in methanol (MeOH). These were further diluted in complete medium to attain the final concentrations of  $50\mu$ g/mL.

The stock solutions were prepared on the assay day. Basic quinine was used as the standard reference drug (positive control).

Extracts was serially diluted twofold in complete medium (RPMI) up to  $0.195\mu$ g/mL using a flat bottomed, 96 - well microtitre plate.

Unparasitised erythrocyte was added to column 1 (blank) which had no drugs, while parasitized red blood cells were added to columns 2–12. The plate was incubated at 37°C for 48 hours. After incubation,  $50\mu$ L of each well were placed on the glass slide and air - dried for 24 hours. GIEMSA was added and microscope lecture was realized. The concentration of any substance that inhibited 50% of the parasite growth (IC<sub>50</sub>) was determined in triplicata.

#### 3. Results and Discussion

## 3.1. Fractionation of cyclohexane and dichloromethane extract

The results of analytical - TLC test of cyclohexane and dichloromethane extracts and characteristics of the spots are resumed in table 1.

TLC profiling of the extracts gives an impressive result that directing towards the presence of number of phytochemicals. Various phytochemicals gives different Rf values in different solvent system. This variation in Rf values of the phytochemicals provides a very important clue in understanding of their polarity and also helps in selection of appropriate solvent system for separation of compounds by preparative - TLC.

Compound showing high Rf value in less polar solvent system have low polarity and with less Rf value have high polarity. Mixture of solvents with variable polarity in different ratio can be used for separation of pure compound from plant extract. The selection of appropriate solvent system for a particular plant extracts can only be achieved by analyzing the Rf values of compounds in different solvent system.

In the present study, each extract gave 6 spots which were viewed under UV lamp (366 and 254 nm), iodine and sulfuric vanillin spray. These spots indicated the presence of six major groups of phytochemical constituents in each extracts.

Cyclohexane - dichloromethane - ethyl acetate, 1: 4: 1 and ethyl acetate - cyclohexane, 1: 1 were found to be best eluting solvents for the separation of constituents from cyclohexane and dichloromethane extracts, respectively.

The results of fractionation of cyclohexane and dichloromethane extracts, the yield and the characteristics of the fractions are resumed in table 2.

Fractionation of cyclohexane and dichloromethane extracts using Preparative - TLC produced six (6) fractions for each extract which were viewed under UV lamp at 366 nm.

Based on this table, the yield values of fractions obtained ranged from 8.09 to 33.09% and 10.02 to 21.09% for cyclohexane and dichloromethane extracts, respectively.

For fractions from cyclohexane extract, Fraction number 6 (F6) had the highest yield (33.09%) followed by fractions 1, 5, 2, 3 and 4. Whereas, for the fractions from dichloromethane; Fraction number 4 (F4) had the highest yield (21.09%) followed by fractions 5, 1, 2, 3 and 6.

Table 1: Characteristics of the spots from cyclonexane and dichloromethane extracts							
Extracts		R <sub>f</sub>	Visible light	UV: 254nm	UV: 366nm	Iodine spray	Sulfuric vanillin spray
	Spot 1	0	Brown	Brown	-	Brown	-
	Spot 2	0.11	-	Brown	-	-	-
Cuelehavana	Spot 3	0.16	-	-	Light blue	-	-
Cyclohexane	Spot 4	0.27	-	Brown	-	-	Black
	Spot 5	0.52	-	Brown	-	-	Black
	Spot 6	0.63	-	Brown	-	Brown	Black
Dichlorométhane	Spot 1	0	Brown	Brown	-	Brown	Black
	Spot 2	0.16	-	-	Light blue	-	-
	Spot 3	0.28	-	Brown	-	Brown	-
	Spot 4	0.40	-	-	Light blue	-	-
	Spot 5	0.80	-	-	-	-	Black
	Spot 6	0.9	-	-	-	-	Black

Table 1: Characteristics of the spots from cyclohexane and dichloromethane extracts

#### **3.2.** Phytochemicals tests of different fractions

Results of phytochemical screening of different fractions from cyclohexane and dichloromethane extracts of *Ageratum conyzoïdes* leaves were summarized in Table 3.

This table revealed for cyclohexane fractions, the presence of terpenoids in all fractions with a large amount in F1 and F2.

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The absence of tannins and anthocyanins in all fractions except F6 and F1, respectively. The alkaloids were found only in the fractions 3 and 4.

However, for dichloromethane fractions the presence alkaloids, tannins, steroids, and terpenoids were revealed in the fractions. The presence of terpenoids in all fractions with a large amount in F6, F2 and F3.

#### 3.3. Antimalarial Activity of fractions

The results of *in - vitro* antimalarial activity of of different fractions from cyclohexane and dichloromethane extracts of *Ageratum conyzoïdes* leaves are summarized in Table 4.

This table showed that only the fractions F1, F2, and F5 from cyclohexane extract were classified as active antimalarial activity whereas fractions F3, F4 and F6 were classified as inactive. Fraction 5 (F5) was the most active fraction with  $IC_{50}$  value of 5.17 µg/mL. This fraction contains only steroids and terpenoids. Thus, this antimalarial action may be attributed to these compounds.

In additional, fractions F2, F3, F4 and F5 from dichloromethane extract were classified as very active antimalarial activity whereas fractions F1 and F6 were classified as less active and inactive, respectively. Fraction 4 (F4), which had only alkaloids, steroids and terpenoids, was found to be the most active antimalarial fraction (IC<sub>50</sub>= 0.25  $\mu$ g/mL) following by fraction 5 (IC<sub>50</sub>= 0.30  $\mu$ g/mL) which

had only steroids and terpenoids. Thus, this antimalarial activity may be attributed to the combination these three secondary metabolites.

Both fractions (F4 and F5) showed also best antimalarial activity than Quinine ( $IC_{50}$ = 1.35 µg/mL) which was use as positive control. Therefore, these fractions are the prospective fractions to be developed as antimalarial agents.

According to the literature, extracts that have very active antimalarial activity with  $IC_{50}$  values  $\leq 5\mu g/mL$ , were classified into active, less active and inactive when  $IC_{50}$  values were 5 - 15, 15 - 30 and  $> 30\mu g/mL$ , respectively [25].

The HPLC profile of the most active fractions 4 and 5 showed three and two major peaks, respectively (Figure 1 and 2).

These results are similar to phytochemical screening which revealed the presence of two compounds (alkaloids, terpenoids and steroids) in the fraction 4 (F4) whereas two compounds (terpenoids and steroids) in the fraction 5 (F5).

By comparing chromatograms of both fractions, the first group of peaks which appears from 9 to 15 minutes could represent terpenoids and steroids, whereas the single peak which appears from 26 to 29.7 minutes could represent alkaloids.

Table 2: Yields and Characteristics of the fractions from cyclohexane and dichloromethane extracts	
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Extracts	Fractions	R <sub>f</sub>	Visible light	UV: 366 nm	Weight (g)	Yield (%)	Colour
Cyclohexane	F1	0.13	Dark green	Violet	0.32	23.53	Reddish
	F2	0.29	Darkyellow	Darkmallow	0.15	11.03	Dark green
	F3	0.43	Brown	Blue	0.13	9.56	Light Green
	F4	0.54	Dark green	Light mallow	0.11	8.09	Light Green
	F5	0.68	Dark green	Purple	0.20	14.71	Yellowish
	F6	0.93	Brown	Black	0.45	33.09	Dark green
					Tot: 1.36		
Dichloromethane	F1	0.29	Dark Brown	-	0.91	18.99	Reddish
	F2	0.38	Dark Brown	-	0.89	18.58	Yellowish
	F3	0.54	light Brown	-	0.55	11.48	Green
	F4	0.59	light Brown	-	1.01	21.09	Light Green
	F5	0.71	Green	-	0.95	19.83	Light Green
	F6	0.97	Yellow	-	0.48	10.02	light Brown
					Tot: 4.79		

 Table 3: Phytochemical screening of the different fractions

Extracts	Fractions	Alkaloids	Anthocyanins	Tannins	Steroids	Terpenoids
	F1	-	++	-	++	++
	F2	-	-	-	-	+
Cyclohexane	F3	++	-	-	-	+
	F4	+	-	-	+	+
	F5	-	-	-	+	+
	F6	-	-	++	++	++
Dichloromethane	F1	-	No tested	+	-	+
	F2	-	No tested	+	-	++
	F3	+	No tested	++	+	++
	F4	++	No tested	-	+	+
	F5	-	No tested	-	+	+
	F6	+	No tested	-	+	+++

+++ = abundantly present; ++ = moderately present; + = present; - = absent.

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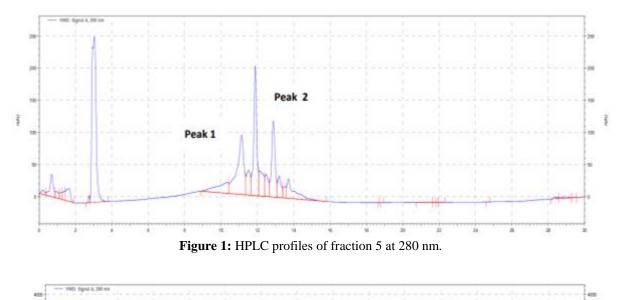
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Extracts	Fractions	IC <sub>50 (</sub> µg/mL)	Antimalarialactivity		
	F1	10.45	Active		
	F2	5.20	Active		
Cyclohexane	F3	>50	Inactive		
Cyclollexalle	F4	>50	Inactive		
	F5	5.17	Active		
	F6	>50	Inactive		
	F1	30	Less Active		
	F2	1.15	Very Active		
Dichloromethane	F3	0.75	Very active		
	F4	0.25	Very active		
	F5	0.30	Very active		
	F6	>50	Inactive		
	Quinine	1.38	Very Active		

**Table 4:** Antimalarial activity of the different fractions



Peak 2

Figure 2: HPLC profiles of fraction 4 at 280 nm.

Peak 1

4. Conclusions

1

In the present study, we evaluated the in - vitro antimalarial activity of fractions from cyclohexane and dichloromethane extracts of *Ageratum conyzoides* Leaves. Results revealed that the fractions 4 (F4) and 5 (F5) from dichloromethane extract showed the best antimalarial activities against the parasite with  $IC_{50}$  values of 0.25 and 0.30µg/mL, respectively. The HPLC profile of these most active fractions were developed and showed three and two major peaks respectively for fraction 4 (F4) and 5 (F5). Preliminary phytochemical screening revealed the presence

of alkaloids, steroids and terpenoids in the fraction 4, whereas only steroids and terpenoids were found in the fraction 5 (F5). According to these results, these fractions are the prospective fractions to be developed as antimalarial agents. Therefore, it is necessary to explore preparative - HPLC of these active in order to isolate, identify and characterize the active pure compounds, which can serve as useful leads for antimalarial drug discovery and development.

Peak 3

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#### 5. Conflicts of Interest

The authors declare that no competing interests exist regarding the publication of this paper.

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