Studies on Drug Delivery Efficacy of Silver Nanoparticles Synthesized using Human Serum Albumin as Tamoxifen Carriers in MCF-7 Cell Line

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Abstract: Tamoxifen (Tam) has a broad spectrum of anticancer activity, but is limited in clinical application. The aim of this study was to explore the smart pH-responsive drug delivery system based on Human Serum Albumin (HSA) reduced Silver nanoparticles (AgNps)for its potential in enabling more intelligent controlled release and enhancing chemotherapeutic efficiency of tamoxifen. Tamoxifen was loaded onto HSA reduced AgNps by forming complexes and Tamoxifen was released from the drug delivery system much more rapidly at pH 4.0 and 6.0 than at pH 7.4, which is a desirable characteristic for tumor targeted drug delivery. Tamoxifen loaded HSA reduced AgNps induced remarkable improvement in anticancer activity a demonstrated by MTT assay and AO/EtBr staining and the results indicated that the drug loaded nanoparticles could act as an efficient drug delivery system which import Tamoxifen into target cancer cells.

Keywords: Tamoxifen, Human Serum Albumin, Silver nanoparticles, Cancer, MCF-7

1. Introduction

Nanotechnology is now frequently used for various applications in fibre and textiles, agriculture, electronics, forensic science, space and medical therapeutics [1]. Biodegradable nanoparticles are frequently used to improve the therapeutic value of various water soluble/insoluble medicinal drugs and bioactive molecules by improving bioavailability, solubility and retention time [2]. Theses nanoparticle- drug formulation reduces the patient expenses, and risks of toxicity [3]. Nanoencapsulation of medicinal drugs (Nanomedicine) increases drug efficacy, specificity, tolerability and therapeutic index of corresponding drugs [4]. These nanomedicines have many advantages in the protection of premature degradation and interaction with the biological environment, enhancement of absorption in to a selected tissue, bioavailability, retention time and improvement of intracellular penetration [5]. Several disease related drugs/bioactive molecules are successfully encapsulated to improve bioavailability, bioactivity and control delivery [6].

Tamoxifen is a commonly used anti-cancer drug. It is most often used against breast cancer, carcinomas, osteosarcoma and soft tissue sarcoma. The effectiveness of tamoxifen in treating various types of cancers is greatly limited by the serious side effects caused by the drug. The initial side effects caused as a result of tamoxifen administration include less serious symptoms, such as nausea, vomiting, myelo suppression and arrhythmia, which are usually reversible [7]. However, tamoxifen associated cardiomyopathy and congestive heart failure have raised great concern among health practitioners [8]. A widely researched approach of increasing the efficacy, while lowering the deleterious side effects caused by anti cancer agents such as tamoxifen, is of developing nanoparticlesbased drug delivery systems [9]. The endogenous Human Serum Albumin (HSA) serves as a suitable material for nanoparticles formation as albumin is naturally found in the blood and is thus easily degraded, nontoxic and non immunogenic [10]. Albumin is an acidic protein and remains stable between pH range 4-9 and temperatures up to 60° C. Albumin-based nanoparticles delivery systems are easily accumulated in tumor tissue due to the enhanced permeability and retention (EPR) effect [11].

Metallic nanoparticles have fascinated scientists for over a century and are now heavily utilized in biomedical sciences and engineering. Silver nanoparticles (AgNps) have emerged as an arch product in the field of nanotechnology [12]. AgNps have become efficient vehicles to store and deliver medicines. They are used for the controlled release of drugs and albumin-drug complexed, where Nanoparticles acts as a carrier of drugs and liberate them on selective basis. They can be used a magic –bullets" that go directly to cells of a particular tissue [13].

In view of these developments, the aim of this paper is to explore a novel Tamoxifen delivery system based on AgNps synthesized using HSA that is chemically stable, environmentally friendly and non-cytotoxic. This type of Tamoxifen delivery system could provide simple and easy way to enable intelligently controlled release of drug and enhance its chemotherapeutic efficiency. To be the best of the authors' knowledge, no study has been carried out on AgNps synthesized using HSA for stimuli sensitive drug delivery system and is investigated here for the first time. In this study Tamoxifen was loaded on AgNps synthesized using HSA to construct drug loaded Nanoparticles as a drug delivery system. Loading efficiency and influence of pH value on drug release behavior were investigated. Additionally, tumor cellular uptake and cytotoxicity were evaluated in MCF-7 breast cancer cells.

2. Materials and methods

AgNO₃ (99.98%) was used as a silver precursor, and was provided by Merk (Darmstadt, Germany). HNO₃ (70%), Human Serum Albumin (HSA) and HCL (37%) were obtained from Sigma-Aldrich (St Lois, MO). The human breast cancer cell line (MCF7) was procured from National centre for Cell Science (NCCS, Pune). All the reagents in this effort were analytical grade and were used as received without further purification. All solution were freshly prepared using double distilled water and kept in a dark to avoid any photochemical reactions. All glass wares used in experimental procedures were cleaned in a fresh solution of $HNO_3/HC1$ (3:1, v/v), washed thoroughly with double distilled water and dried before use

Synthesis of Silver Nanoparticles

Silver Nano particles were prepared by wet chemical synthesis. An albumin solution was prepared dissolving 0.1gms of HSA in 30ml of deionized water. The AgNO₃ solution was prepared dissolving 0.05gms of AgNO₃ in 10 ml of deionized water. In the HSA solution, AgNO₃ solution was added while stirring on a magnetic stirrer and after 5 minutes, ammonium hydroxide was added drop wise to maintain the pH. It was stirred further for 15 minutes, and to let the solution settle down for 24 hours. The silver nano particles thus formed were isolated by filtration and washed with a minimum of amount of ethanol.

Preparation of Tamoxifen-loaded HSA reduced AgNps

Tamoxifen loaded nanoparticles were prepared by solvent evaporation and emulsification cross linking method. For the preparation of tamoxifen-loaded HSA reduced AgNPs, 5 mg tamoxifen was dissolved in 2ml dehydrated alcohol and ultrasonicated for 15 minutes. This organic phase was then slowly poured into the aqueous solution of sodium dodecyl sulfate (SDS) and polyvinyl alcohol (PVA). The solution of 1.1 ml SDS and 0.4 ml PVA in 10ml de-ionized water was maintained at 35.5°C for 2 h under stirring rate of 500 rpm. The oil in water emulsion was formed after removal of alcohol under reduced pressure. The resulting aqueous suspension containing Tamoxifen was denoted as solution 1, which was then added to the Nanoparticles solution (Solution 2) by varying the concentrations of solution 1 and 2, the Tamoxifen -loading amount may be finely tuned. The resulting mixture was then sonicated at an energy output of 50W for 30 minutes to obtain a uniform solution, denoted as solution 3, which was then added drop wise to 30 ml liquid paraffin under stirring for 12 h. the water in oil emulsion was formed without extra emulsifier. Finally, 1 ml of 20 wt.% sodium citrate was added to crosslink the HSA reduced AgNps for 1 h, and the resulting nanoparticles were purified and stored after being dried.

Characterization of the nanoparticles

The prepared Ag-NPs were characterized by X-ray diffraction (XRD), Scanning electron microcopy (SEM), energy-dispersive X-ray spectrometry (EDAX) and Fourier-transform infrared spectroscopy (FTIR). The structures of silver Nanoparticles produced were examined by XRD (XRD-6000; Shimadzu). The XRD patterns were recorded at a scan speed of 4° / minute. SEM was performed using a Philips XL-30 instrument (Philips, Eindhoven, Netherlands)

to study the morphology of Ag-NPs. The EDAX was carried out on a DX-700HS spectrometer. Meanwhile, the FT-IR spectra were recorded over the range of 400-4000cm⁻¹using an FT-IR series 100, 1650 PerkinElmer spectrophotometer (Los Angeles, CA)

Determination of drug loading and encapsulation efficiency:

The drug loading efficacy was calculated by two ways, first based on indirect method by estimating the drug content of the supernatant and second based on direct estimation of the drug content present in the pellet obtained after centrifugation. The drug concentration in supernatant and redispersed pellets were determined by measurements of its UV absorbance at 254 nm using UV/visible spectroscopy and the percentage loading of drug on to nanoparticles was estimated by the following formula.

Loading efficiency =	<u>W₀</u> W	× 100%
Encapsulation efficiency =	$\frac{W_0}{W_1}$	× 100%

Where W_0 is the weight of drug enveloped in the silver nanoparticles, W is the weight of silver nanoparticles, and W_1 is the amount of drug added in the system.

In Vitro drug release study

Release behavior of Tamoxifen from nanoparticles was investigated. Tamoxifen-loaded HSA reduced AgNPs were dispersed in PBS (pH 7.4, 5 ml) and transferred into a dialysis bag. The dialysis bag was then immersed in 95 ml of PBS at pH 5.0, 6.0, or 7.4. The release medium was continuously agitated with a stirrer at 50 g and 37°C. At predetermined time intervals, 2 ml of the external medium was collected and replaced with the same fresh PBS. The amount of released Tamoxifen in the medium was then determined at 254 nm.

In Vitro Cytotoxicity Study

The MCF-7 cells were cultured in Dulbecco's modified Eagle's medium (DMEM) supplemented with 1.5 g/ml sodium bicarbonate, 4 mm glutamine and 10% fetal bovine serum (FBS). The cultures were maintained in a humidified atmosphere with 5% CO2at 37°C. Briefly, the cells were cultured in 96-well plates with different concentration (0– 60μ g/ml) of drug loaded nanoparticles with the same amount of drug for 48 h. All the experiments are performed in triplicates.

Fluorescence microscopic observation

The MCF-7 cells were grown $(1 \times 10^5 \text{ cells/cover slip})$ and incubated with nanoparticles at their different concentration and then they were fixed in methanol: acetic acid (3:1, v/v). The cover slips were gently mounted on glass slides for the morphometric analysis. Morphological changes of MCF-7 cells were analyzed under fluorescence microscope at 400× magnification. Approximately 5µl of dye mixture (100 mg/ml acridine orange (AO) and 100 mg/ml Ethidium bromide (EtBr) was mixed with 9 ml of cell suspension (1 × 10⁵cells/ml) on a clean microscopic cover slip. After incubation for 2–3 min, cells were visualized under fluorescence microscope at $400 \times$ magnification with excitation filter at 510–590 nm. The percentage of total apoptotic cells was determined by the following formula;

Cell uptake efficiency of nanoparticles

The MCF-7 cells were seeded in 96-well micro titer plate. After the cells reached 70–80% confluence, they were incubated with drug or drug loaded nanoparticles 0% and 50% nanoparticles at 5 μ g/ml drug concentration for 1, 2, 3 and 4 h, respectively at 37°C. At the designated interval, the sample wells were washed three times with 50 ml cold PBS and then added with 100 ml culture medium. After that, all the wells were lysed by 50 ml 0.5% Triton X-100 in 0.2 N NaOH. The amount of Tamoxifen in the cells was fluorometrically determined for the lysate with an excitation wavelength of 254 nm and an emission wavelength of 380 nm using a fluorescence spectrometer. The cellular uptake efficiency was expressed as the percentage of the fluorescence associated with the cells vs. that present in the positive control solution.

3. Results and Discussion

Human serum albumin is a most abundant plasma protein and it is widely used in Bio-nanotechnological applications. It possesses a zwitter ionic character at the isoelectric point (pH 7.4) wit exposed ionic groups (C and N Terminus) at the side chains of the globular protein, which are present in solution and are promising sites for binding cationic and anionic groups [14]. The general process for synthesis of Nanomaterials in the foam matrix involves the electrostatic complexation of metal ions with oppositely charged surfactant molecules, followed by foam generation and subsequent insitu chemical reaction. HSA imparts greater flexibility by allowing the complexation of metal ions with opposite charges simultaneously [15]. In this study we prepare AgNps, by mixing HSA and Silver Nitrate solution. The formation of Ag Nanoparticles was further confirmed by using Scanning electron microcopy (SEM), X-ray diffraction (XRD), Energy-dispersive X-ray spectrometry (EDAX) and Fourier-transform infrared spectroscopy (FTIR).

Scanning Electron Microscopy was employed to visualize the size and shape of Ag-NPs. Figure 1 shows that the particles are relatively face centered cubical with the diameter of 20-30 nm. The Nanoparticles were not in direct contact even within aggregates, indicating stabilization of the Nanoparticles by HSA as capping agent.

Figure 2 shows XRD patterns for Ag-NPs synthesized using HSA. The XRD pattern indicated that the structure of Ag-NPs is face centered cubic [16]. In addition, all the Ag-NPs had a similar diffraction profile, and XRD peaks at 2θ of 34^{0} , 49.1^{0} , 58.9^{0} and 71.4^{0} could be attributed to the 111, 200, 220 and 311 crystallographic planes of the face-centered cubic silver crystals respectively [17]. The XRD pattern thus clearly illustrated that the Ag-NPs formed in

this study were crystalline in nature. The main crystalline phase was silver, and there were no obvious other phases as impurities were found in the XRD patterns. The average particle size of Ag-NPs calculated using the Debye-Scherrer equation

$$n=\frac{K\lambda}{\beta\cos\theta}$$

Where K is the Scherrer constant with value from 0.9 to 1 (shape factor), where λ is the X-ray wavelength (1.5418 Å), $\beta_{1/2}$ is the width of the XRD peak at half-height and θ is the Bragg angle. From the Scherrer equation, the average crystallite size of Ag-NPs for the sample at 24 hours of stirring are found to be 20-30 nm in size.

Figure 3 shows the clear elemental composition profile of the synthesized Ag-NPs. The intense signal at 3keV strongly suggests that Ag was the major element of these NPs, as it has optical absorption in this range due to the surface Plasmon resonance [18]. The other signals (carbon and oxygen) indicate the presence of tuber extract, which corresponds to the biomolecules that were capping over the Ag-NPs.

Tamoxifen loaded Ag-Nps with high drug payload was successfully prepared and characterized by FTIR spectra. The FTIR spectra of Ag-Nps alone (fig 4), Tamoxifen alone (fig 5 (a)) and Tamoxifen loaded Ag-Nps (fig 5(b)) are shown. The three characterization peaks at 3417.86 cm⁻¹ (- OH), 1510.26 cm⁻¹ (C-O-C) and 1060.85 cm⁻¹(N-H) were taken into account. In comparison, the peak at 3417.86 cm⁻¹ of Tamoxifen loaded Ag-Nps was much wider than Tamoxifen, indicating the stretching of hydrogen bonds. The peak at 1658.78 cm⁻¹ noticed in Tamoxifen was found to be shifted in Tamoxifen loaded Ag-Nps with a peak at 1653.00 cm⁻¹. From the dissolution point of view, the salvation of Ag-Nps is related to the protonation of free amine groups and breakdown of strong intra- and inters- molecular hydrogen bonding. The H- bonding established through -NH₂ linkage makes the molecule behave as a resonating structure owing to unshared electron pairs that are less available for protonation. In addition, nitro groups withdraw electrons, making them less accessible at the secondary amide and primary amide groups during interaction. It is evident from the graph that there is a possible interaction between Tamoxifen and Ag-Nps. A shift in the peak of Tamoxifen loaded Ag-Nps indicates the interaction of Ag-Nps with Tamoxifen. The change indicates encapsulation of Tamoxifen to form Ag-Nps successfully.

The drug loading on Ag-Nps is done and the amount of drug loaded is shown in the table 1. The result clearly indicates that the amount of drug loading depends on the ratio of drug and carrier. The drug loading percentage increases when the amount of Ag-Nps is increased and it decreases when the amount of drug is increased for a given amount of Ag-Nps. The amount of drug loaded per mg of Ag-Nps has been calculated. The amount of drug loaded per mg of Ag-Nps decreases with increase in the Ag-Nps and it increases with increase in the Ag-Nps and it increases with increase in the amount of drug is loaded per mg of Ag-Nps. When the amount of drug is increased, there is no further increase in the amount of drug loaded per mg of Ag-Nps.

Ag-Nps. 0.7 mg of Tamoxifen is the maximum of the drug loaded per mg of Ag-Nps.

As shown in the Figure 6, release of drug molecules depended on pH of the medium and release time. Drug release at pH 7.4 was slow and sustained with release ratio at about $16\pm 0.58\%$ within 48 h. However, Tamoxifen release rate was much faster at lower pH, with approximately $32 \pm 0.58\%$ (pH 6.0) and $53 \pm 0.63\%$ (pH 4.0) of the drug released within 48 h. Protonation of the drug occurred at lower pH, which released chemisorbed drug molecules in to the medium. Furthermore, the surface charge of Ag-Nps turned positive at lower pH, which blunted the electrostatic interaction of drug and Ag-Nps and facilitated the drug release process. These results demonstrate pH-triggered drug release behavior from the novel Tamoxifen Delivery System.

It may be hypothesized that most of the drug will remain in the carrier for a considerable time period at normal physiological conditions (pH 7.4), indicating the potential for prolonged drug retention time in blood circulation and thereby greatly reducing the side effects of normal tissues. On the other hand, once drug-loaded Ag-Nps are taken up by the tumor cells *via* endocytotic process, a faster release may occur at lower pH i.e., surrounding the tumor site or inside the endosome and lysosme of tumor cells, leading to significant improvement in cancer treatment efficacy [19].

To further explore the anti cancer efficiency of the AgNps synthesised using HSA, MCF-7 cells were culture with drug at different concentrations (0-60µg/ml) and drug loaded nanoparticles with equivalent drug concentration for 48 h. Cytotoxic results were estimated by MTT assay (Figure 7). The MCF -7 cells treated by drug loaded HSA reduced AgNps showed decreases viability compared with drug alone (*p < 0.05). It is apparent that drug and even more effective drug loaded HSA reduced AgNps could arrest the growth of cancer cells and cause more significant morphological changes, indicating an increased probability in cell death.

The fluorescent microscopy study on MCF-7 breast cancer cells treated with different concentrations of HSA reduced AgNps loaded with drug was conducted using AO/EtBr staining in order to evaluate the influence of drug loaded nanoparticles on MCF-7 cells. Figure 8 shows the images of normal and Tamoxifen-loaded HSA reduced AgNps treated MCF-7 cells. In Ao/EtBr staining cells observed as orange color are apoptotic and necrotic cells were observed as red color fluorescence due to their loss of membrane integrity [20].

The cellular uptake efficiency of tamoxifen loaded HSA reduced AgNps was compared to the tamoxifen alone after 1, 2, 3 and 4 h in cell culture (Figure 9). Tamoxifen loaded HSA reduced AgNps exhibited enhanced cellular uptake (p < 0.05) for the MCF-7 cells after 1, 2, 3 and 4h culture, respectively, compared with tamoxifen alone. It can be seen that Tamoxifen loaded HSA reduced AgNps demonstrated much higher cellular uptake efficiency than the drug.

4. Conclusion

In the present study, Tamoxifen loaded HSA reduced AgNPs were formulated and pH-responsive drug release study revealed that the drug release was significantly accelerated by decreasing pH from 7.4 to 4.0. This mechanism has a particular interest in cancer therapy due to the acidic extracellular tumor environment in the cancer cells. The blank HSA reduced AgNPs had almost no cytotoxicity besides, the inhibiting rate of drug loaded HSA reduced AgNPs to MCF-7 breast cancer cells tended to increase with increase in the concentration of drug. All these characteristics demonstrate that the novel pH-responsive drug delivery system based on HSA reduced AgNPs for Tamoxifen is a promising strategy for cancer therapy.

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Figure1: Scanning Electron Microscopy image of HSA reduced Ag-NPs which are face centered cubical in shape



20 degree

Figure 2: X-ray diffraction pattern of Ag-NPs synthesized using HSA. The diffractions at 34⁰, 49.1⁰, 58.9⁰ and 71.4⁰ can be indexed to 111, 200, 220 and 311 planes of the face-centered cubic Ag-NPs, respectively



Figure 4: Fourier Transform Infra red spectra for Ag-NPs synthesized using HAS

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Figure 5: (a) Fourier Transform Infra red spectra for Tamoxifen. (b) Fourier Transform Infra red spectra of Tamoxifen loaded Ag-Nps

Table 1: Drug	loadin	ig in percenta	ge for	different	ratios o	f Tamoxife	en and Ag-Nps

S.No	Tamoxifen:	Drug	mg of Tamoxifen /
5.110	Ag-Nps	Loading (%)	mg of Ag-Nps
1.	1:1	51.6	0.51
2.	1:2	62.14	0.33
3.	1:4	77.35	0.31
4.	2:1	30.5	0.70
5.	4:1	10.5	0.70



Figure 6: In vitro drug release profiles from Ag-Nps at neutral condition (pH 7.4) and acidic condition (pH 6.0 and pH 4.0) at 37° C. Each point represents the mean \pm SD.

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Concentration µg/ml

Figure 7: MCF-7 cells were treated with AgNps, tamoxifen and Tamoxifen Loaded AgNps for 48 h and cell viability was determined b colorimetric MTT assay. Data represent mean ± SD.



Figure 8: Fluorescent microscopic image of MCF-7 cells treated with AgNps loading Tamoxifen as a drug delivery system by Ao/EtBr Staining



Figure 9: Time dependent cellular uptake efficiency of Tamoxifen loaded HSA reduced AgNps by MCF-7 cells after 1.0, 2.0, 3.0 and 4.0 h of incubation. Data represent mean \pm SD.

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